myosin as prepared by other workers is not extracted directly from the muscle but rather is made by ATP dissociation of previously prepared actomyosin2,10.

NATURE

Consideration was given to the possibility that the high levels of F-actin used were giving the same anomalous viscosity effects on artery actomyosin which have been observed for skeletal muscle actomyosin7. This, however, does not appear to be the case. The artery protein did not show the anomalous viscosity behaviour even when a large excess of skeletal muscle F-actin was added.

A different preparation of artery actomyosin, prepared according to the procedure of Laszt and Hamoir, did not have the actin-combining moiety associated with it. The data (Table 3) show that even large amounts of F-actin failed to induce an elevated Z_{η} or ATP sensitivity when added to actomyosin prepared by the method of Laszt and Hamoir1.

Table 3. Effect of $\emph{F}\text{-actin}$ on the Relative Viscosity of Artery Actomyosin*

	Pr	eparation 1	Preparation 2	
F-actin/AM	$Z\eta$	ATP sensitivity (%)	$Z\eta$	ATP sensitivity (%)
No actin	0.25	36	0.18	27
1/20	0.33	.35	0:11	32
1/10	0.29	4 5	0.13	40
1/5	0.37	39	0.26	72

* The two preparations of artery actomyosin were made according to the method of Laszt and Hamoir.

Viscometry conditions were as described in the text. The F-actin was from rabbit skeletal muscle. The F-actin/actomyosin was on a weight basis (mg protein). The concentrations of actomyosin in the viscometer were 1.5 mg/ml. and 1.8 mg/ml. for preparations 1 and 2, respectively.

The results of this study indicate that a firm association exists between cow carotid artery actomyosin and a myosin-like component. Since the association exists even after treatment to remove myosin it is suggested that artery myosin is different in solubility properties from skeletal muscle myosin.

This work was supported in part by research grants from the National Heart Institute (HE-07353 and HE-06315), U.S. Public Health Service, and by the Heart Association of South-western Ohio.

MORTON L. MALLIN

May Institute for Medical Research of the Jewish Hospital of Cincinnati, Ohio.

- ¹ Laszt, L., and Hamoir, G., Biochim. Biophys. Acta, 50, 430 (1961).
- Hamoir, G., and Laszt, L., Nature, 193, 682 (1962).
 Filo, R. S., Ruegg, J. C., and Bohr, D. F., Amer. J. Physiol., 205, 1247 (1963).
- Fiske, C. H., and Subba Row, Y., J. Biol. Chem., 86, 375 (1925).
 Lowry, O. H., Rosebrough, N. J., Farr, A. L., and Randall, R. J., J. Biol. Chem., 193, 265 (1951).
- ⁶ Mommaerts, W. F. H. M., in *Methods in Medical Research*, 7, 1-23, editor-in-chief, J. V. Warren.
- ² Szent-Györgyi, A. G., J. Biol. Chem., 192, 361 (1951).
- Needham, D. M., and Williams, J. M., Biochem. J., 89, 534 (1963).
- Fryar, A. J., and Gibbs, R. J., Experientia, 19, 493 (1963).
- 10 Needham, D. M., and Williams, J. M., Biochem. J., 89, 552 (1963).

Further Purification of Chick Interferon

CHICK allantoic fluid interferon has been purified to various extents by several groups of workers. Among those who prepared interferon of high specific activity, Lampson et al. achieved a purification factor of 1830 after treatment with perchloric acid, concentration with zinc acetate and two fractionations on 'CM-cellulose'. Such material still contained an inactive protein component that was only removed by electrophoresis on 'Pevikon'. The final material had been purified 4,500 times. Fantes et al.2 purified interferon several thousand times by adsorption on 'Doucil', elution with KSCN, precipitation of impurities in the presence of KSCN at pH 3.5, precipitation of interferon with methanol at pH 7.5 and chromatography on DEAE cellulose at pH 7.5. Merigan³ purified crude interferon with perchloric acid and zinc acetate by the method of Lampson et al.1, achieving a 15-fold purification. Such material, after one chromatographic fractionation on 'CM-Sephadex', using a pH gradient instead of Lampson's stepwise process for eluting

the active ingredient, was claimed to be purer than Lampson's material, though no evidence of electrophoretic homogeneity was presented.

We have slightly modified our original method². stead of neutralizing the acid KSCN supernatant fluid before precipitating with methanol, we lowered the pH further to 2.0 and added methanol (5 vol.) at this stage, precipitating some more inactive protein. The pH of the methanolic supernatant fluid was then adjusted to an apparent value of 7.5, causing precipitation of the active material. Chromatography on DEAE-cellulose was carried out as described before². The interferon, when chromatographed on DEAE cellulose, became re-diluted and was reconcentrated on 'CM-sephadex's. Adsorption was performed at pH 5.5 in 0.1 M phosphate buffer, with subsequent washing with 0.1 M phosphate buffer at pH 6 until the eluates contained no more than about 2 µg/ml. protein, as determined by the method of Lowry et al.4. Elution was carried out with a 0.1 M phosphate buffer gradient of rising pH.

Three such preparations were subjected to disc electrophoresis in acrylamide gel columns by Dr. J. Williams (personal communication) at the Laboratory of Molecular Biology in Cambridge. Two of the samples, after staining with amido schwarz, exhibited only one protein band. Dissection and assay of two control columns showed that the biological activity resided in those bands. sample contained some minor protein contaminants, estimated to amount to less than 5 per cent of the total.

The purity of the most potent DEAE-eluate fractions was sometimes of the same order as that of the post 'CM-Sephadex' fractions, but the 'CM-Sephadex' process enabled the less pure off-peak DEAE cellulose fractions to be purified to the same extent, thus increasing the overall yield of highly purified material.

The post-DEAE material—a mixture of highly and moderately purified fractions—applied to 'CM-Sephadex' had usually been purified about 500-fold. When, on the other hand, pre-DEAE material (purified about 80 times on average) was fractionated on 'CM-Sephadex' in a similar way, it was not found possible to free the active material from an inert protein with an elution peak at a pH only just above that of interferon.

The best fractions were purified well over 10,000 times and contained about one million interferon units per mg protein. As the specific activities of crude interferons and also assay methods vary from one laboratory to another, these figures do not necessarily mean that interferon prepared by the process described here was of a higher degree of purity than that obtained by other workers.

The iso-electric point of partly purified interferon was determined electrophoretically in a sucrose gradient by Dr. A. Polson (personal communication) at the University of Cape Town and found to be about pH 6.8-6.9.

Purified interferon was, as mentioned in our previous publication2, highly species specific. This has since also been reported by Merigan³ and Baron et al.⁵. Contrary to the claims of a patent for 'non-specific interferon's, we were unable to overcome the species specificity of crude or purified chick interferon by treatment with carboxy-

I thank Mr. C. F. O'Neill for his assistance and Mr. P. J. Mason for carrying out the interferon assays.

K. H. FANTES

Virus Unit, Glaxo Laboratories, Ltd., Sefton Park,

Stoke Poges, Buckinghamshire.

- ¹ Lampson, J. P., Tytell, A. A., Nemes, M. M., and Hilleman, M. R., Proc. Soc. Exp. Biol. and Med., 112, 408 (1963).
- Fantes, K. H., O'Neill, C. F., and Mason, P. J., Biochem. J., 91, 20P (1964).
 Merigan, T. C., Science, 145, 811 (1964) (and personal communication).
- 4 Lowry, O. H., Rosebrough, N. J., Farr, A. L., and Randall, R. J., Biol. Chem., 193, 265 (1951).
- ⁵ Baron, S., Barban, S., and Buckler, C. E., Science, 145, 814 (1964).
- Olin Mathieson Chemical Corp., U.S.P. 195,428; German patent 1,179,332.