

High-performance Ag₂Se-based thermoelectrics for wearable electronics

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Flexible thermoelectric materials and devices hold enormous potential for wearable electronics but are hindered by inadequate material properties and inefficient assembly techniques, leading to suboptimal performance. Herein, we developed a flexible thermoelectric film, comprising Ag₂Se nanowires as the primary material, a nylon membrane as a flexible scaffold, and reduced graphene oxide as a conductive network, achieving a record-high room-temperature ZT of 1.28. Hot-pressed Ag₂Se nanowires exhibited strong (013) orientation, enhancing carrier mobility and electrical conductivity. Dispersed reduced graphene oxide further boosts electrical conductivity and induces an energy-filtering effect, decoupling electrical conductivity and the Seebeck coefficient to achieve an impressive power factor of $37 \mu\text{W cm}^{-1} \text{K}^{-2}$ at 300 K. The high-intensity between Ag₂Se and reduced graphene oxide interfaces enhance phonon scattering, effectively reducing thermal conductivity to below $0.9 \text{ W m}^{-1} \text{K}^{-1}$ and enabling the high ZT value. The nylon membrane endowed the film with exceptional flexibility. A large-scale out-of-plane device with 100 pairs of thermoelectric legs, assembled from these films, delivers an ultrahigh normalized power density of $>9.8 \mu\text{W cm}^{-2} \text{K}^{-2}$, outperforming all reported Ag₂Se-based flexible devices. When applied to the human body, the device generated sufficient power to operate a thermo-hygrometer and a wristwatch, demonstrating its practical potential for wearable electronics.

The growing adoption of self-powered wearable electronics has accelerated advancements in portable energy harvesting technologies¹. Among these, thermoelectric generators (TEGs), which enable direct conversion between heat and electricity, have gained significant attention for their battery-free operation and low maintenance costs². Flexible wearable TEGs with high output performance are particularly advantageous, as their adaptability to various curvatures ensures optimal thermal contact and efficient energy conversion³. Integrating flexible substrates with flexible

thermoelectric films provides distinct benefits over rigid thermoelectric legs, which often require reduced thickness to achieve flexibility. This trade-off diminishes temperature differences (ΔT) and lowers power density (ω)⁴. However, due to the limited thickness of thermoelectric films, traditional film-based TEGs are typically designed with an in-plane structure, where the film aligns parallel to the substrate⁵. This configuration reduces the contact area with heat sources and limits ΔT , restricting such devices to functioning as signal sources for sensors rather than as power supplies for wearable

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electronics^{6–8}. To overcome these limitations, rational device design is crucial. Optimized designs should maximize the collection of body heat across large areas while efficiently utilizing cross-plane ΔT , thereby enhancing the practical applicability of film-based TEGs in powering wearable electronic devices.

In addition to optimizing TEG structures⁹, the development of high-performance thermoelectric materials is a crucial strategy for advancing efficient flexible thermoelectric devices¹⁰. The thermoelectric performance of materials is quantified using the dimensionless figure of merit, $ZT = S^2\sigma T/\kappa$ ^{11,12}, where S , σ , $S^2\sigma$, T , and κ represent the Seebeck coefficient, electrical conductivity, power factor, absolute temperature, and thermal conductivity, respectively. Among thermoelectric thin-film materials, Bi₂Te₃-based films are renowned for their exceptional thermoelectric performance¹³. However, their high cost¹⁴ and inherently layered structure¹⁵ limits their flexibility and adaptability for wearable applications. Moreover, achieving high-performance flexible Bi₂Te₃ films often relies on expensive equipment and complex fabrication processes, such as magnetron sputtering to deposit materials onto flexible substrates¹⁶. These challenges significantly hinder their scalability and broader application of Bi₂Te₃ films. Therefore, exploring cost-effective and flexible alternatives is imperative for advancing the practicality of thermoelectric devices in wearable technologies.

Ag₂Se has recently attracted significant attention for thermoelectric properties, with ZT values comparable to those of Bi₂Te₃ near room temperature¹⁷, making it highly promising for converting low-grade thermal energy into electricity. To enhance the thermoelectric performance and flexibility of Ag₂Se thin films^{18–24}, various strategies and methods have been developed to enhance the thermoelectric performance and flexibility, including adjusting the synthesis temperature²⁵, tuning the ratio of Ag/Se^{26,27}, hybridizing with polymers^{28–31}, sandwich engineering³², and doping with other elements (e.g., Cu³³ and S³⁴). Notably, inducing a strong (013) orientation in polycrystalline Ag₂Se films has been shown to improve carrier mobility (μ) and σ ^{35–39}, as demonstrated in studies utilizing techniques like Te doping for orientation optimization⁴⁰. Beyond doping and conventional compositional optimization, reducing grain-boundary resistance in thin films has emerged as an effective strategy for improving σ . Grain-boundary resistance significantly hinders μ at lower temperatures (below 500 K), which negatively impacts σ ^{41,42}. Increasing grain size to reduce grain boundary intensity has shown promise in addressing this issue. However, much of the current research overly prioritizes achieving a high $S^2\sigma$ while overlooking κ and the overall ZT . Notably, high σ , often responsible for elevated $S^2\sigma$ values, can adversely affect ZT ⁴³, by increasing both electronic (κ_e) and lattice thermal conductivity (κ_l), as $\kappa = \kappa_l + \kappa_e$. Larger grain sizes, which reduce phonon scattering at grain boundaries, may further increase κ_l , complicating efforts to optimize overall thermoelectric performance. Achieving a high ZT while maintaining an elevated $S^2\sigma$ remains a critical and ongoing challenge for Ag₂Se-based thermoelectric thin films.

Results and discussion

To address these challenges, we propose an innovative structural design for a high-performance, highly flexible thermoelectric film. This film incorporates Ag₂Se nanowires as the primary components, reduced graphene oxide (rGO) as a conductive network, and a nylon membrane as a flexible supporting scaffold, resulting in a film with exceptional thermoelectric performance and remarkable flexibility. Through the hot-pressing process, Ag₂Se nanowires with protrusions form a film with a strong (013) orientation, which enhances the intrinsic μ and σ . The incorporation of rGO forms a conductive network, which further improves the electron mobility in the film, boosting the film's σ (Fig. 1a and Supplementary Fig. 1). The uniform dispersion of rGO forms high-intensity Ag₂Se-rGO interfaces, which enhance the scattering of phonons, while activating the energy

filtering effect⁴⁴, effectively decoupling the σ and the S . As a result, this synergy achieves an impressive $S^2\sigma$ of 37 $\mu\text{W cm}^{-1} \text{K}^{-2}$ at 300 K, with a high S of $-158 \mu\text{V K}^{-1}$ and σ of 1481 S cm^{-1} , significantly outperforming previous reports (Fig. 1b)^{22,27,30,34,40,45–53}. Moreover, high-intensity Ag₂Se-rGO interfaces effectively suppress κ_l ($-0.09 \text{ W m}^{-1} \text{K}^{-1}$), reducing κ below 0.9 $\text{W m}^{-1} \text{K}^{-1}$ and enabling a high ZT value of 1.28. The nylon membrane, serving as a flexible scaffold, endows the film with outstanding mechanical flexibility. In contrast, the solvothermal method offers a more economical alternative to vacuum coating technologies, such as magnetron sputtering and pulsed laser deposition, which typically require costly equipment and high-purity targets. Further, compared to Bi₂Te₃, Ag₂Se demonstrates superior thermoelectric performance, greater flexibility, and lower material costs (Fig. 1c). Inspired by arched bridge structures, we fabricated a large-area ($14 \times 14 \text{ cm}^2$) out-of-plane TEG, as shown in Fig. 1d. This TEG, composed of 100 pairs of thermoelectric legs based on the Ag₂Se-rGO film, achieved a highly competitive ω of 6.7 mW cm^{-2} obtained at ΔT of 26 K (Fig. 1e), as well as an outstanding normalized power density (ω_n) exceeding 9.8 $\mu\text{W cm}^{-2} \text{K}^{-2}$. This ω_n value is higher than that of most published TEGs based on thin films⁵ and even TEGs based on bulk materials⁵⁴. When applied to the human body, the wearable TEG generated sufficient electrical energy to power a thermo-hygrometer and a wristwatch, demonstrating its immense potential for practical applications in wearable electronics, particularly in human health monitoring.

Microstructure analysis

To investigate the influence of varying rGO content on the thermoelectric performance of Ag₂Se films, we defined the samples as Ag₂Se + x wt% rGO films, where $x = 0, 0.01, 0.02, 0.03$, and 0.04. Unlike previous reports, this work employed a high-temperature-assisted ultrasonication method, instead of conventional aging techniques to promote the growth of Ag₂Se nanowires. The novelty of this approach lies in a three-step synthesis process: first, well-dispersed Se nanowires are prepared using high-temperature-assisted ultrasonication; second, during the synthesis of highly reactive Ag₂Se nanowires at elevated temperatures, rGO is uniformly incorporated; and finally, dense composite films with highly crystallized grains fabricated by employing an optimized hot-pressing process. The synthesis mechanism is detailed in Supplementary Fig. 2. In the initial stage, amorphous α -Se particles are formed (Supplementary Fig. 3a), which subsequently develop into crystalline t -Se seeds that gradually grow into Se nanowires. High-temperature-assisted ultrasonication plays a crucial role in coordinating the uniform formation of nuclei and promoting the growth of Se particles. Comparative studies with nanowires fabricated using low-temperature ultrasonication or non-ultrasonicated aging methods confirm this mechanism. In these cases, many unconverted Se particles and small Se nanowires were observed (Supplementary Fig. 3b, c). This is because the combination of high temperature and ultrasonication enhances the surface activity of α -Se, accelerating t -Se nucleation and growth rates, thereby forming nanowires with superior crystallinity^{55,56}. This method not only eliminates the time-consuming aging process, significantly improving preparation efficiency, but also produces Se nanowires with better crystallinity, higher uniformity, and excellent dispersion (100–120 nm in diameter), as demonstrated in Supplementary Fig. 4^{28,34,45}. In contrast, extending the ultrasonication time (e.g., up to 3 min) results in shorter and irregularly shaped nanowires, likely due to the rapid formation of excessive t -Se seeds during the brief ultrasonication process (Supplementary Fig. 3d).

Using these Se nanowires as templates, we synthesized Ag₂Se nanowires at higher temperatures than those reported in previous studies^{25,26,28,29,45}. The elevated temperatures accelerated the bonding between Ag and Se, further promoting the formation of Ag₂Se nanowires with superior crystallinity (Fig. 2a and Supplementary Fig. 5). Interestingly, certain protrusions formed on the Ag₂Se nanowires,

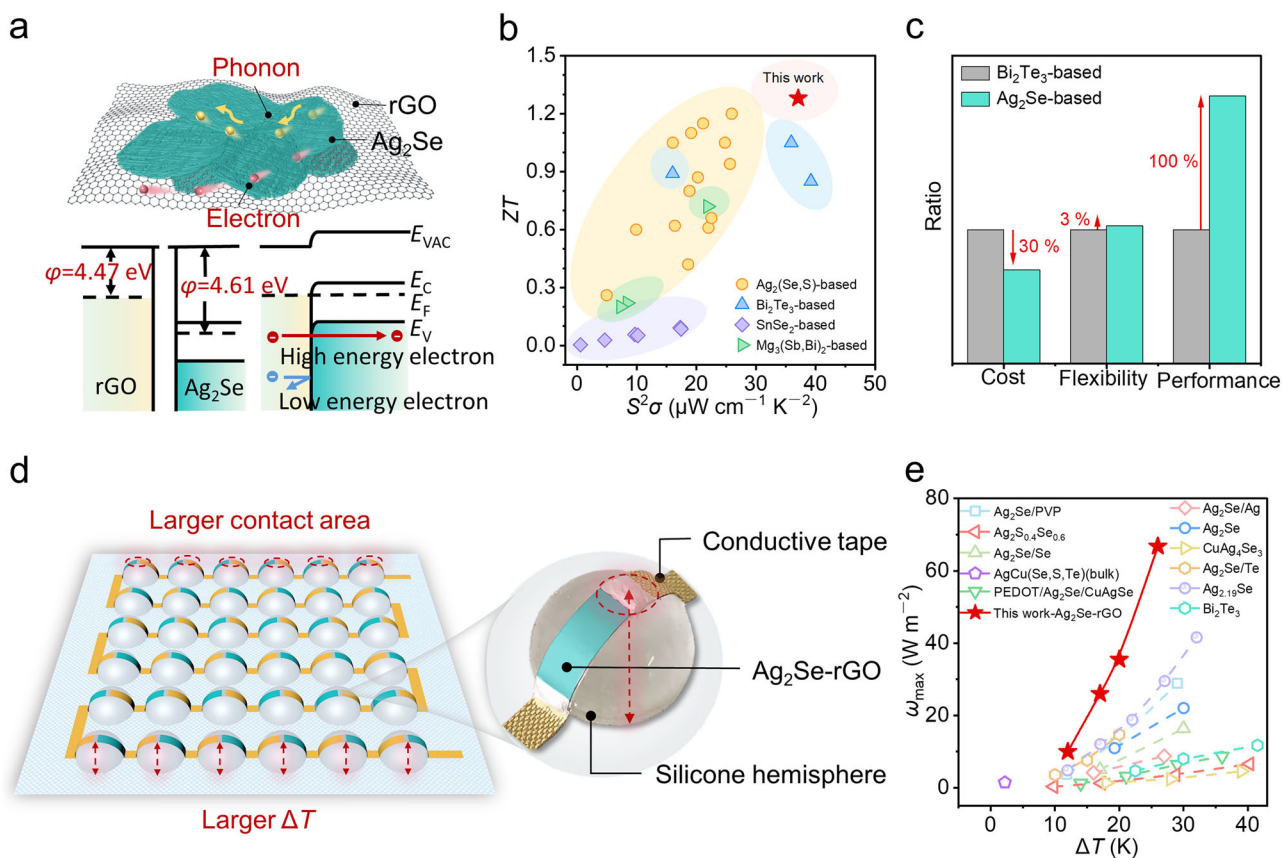


Fig. 1 | Introduction of high-performance Ag_2Se films and out-plane thermoelectric generator (TEG) for wearable electronics. a Transport mechanism of phonons and electrons within Ag_2Se -rGO films. Here rGO is abbreviated from reduced graphene oxide. **b** Room-temperature ZT between this work and some

representative reports^{22,27,30,34,40,45–53,75–78}. **c** Comparison of materials cost, flexibility, and performance between Ag_2Se and Bi_2Te_3 fabricated by solvothermal methods³. **d** Image of the TEG in this work. **e** Comparison of power density (ω) values between this work and reported Ag_2Se - and Bi_2Te_3 -based TEGs^{4,19,25–28,30,33,34,40,79}.

which can be attributed to the higher surface energy and activity at the edges and tips of the nanowires^{57–60}. These protrusions offer significant benefits during the hot-pressing process by increasing the contact area between nanowires, improving adhesion, and enhancing the mechanical strength of the resulting films (Supplementary Fig. 6). Additionally, the irregular protrusions exhibit higher reactivity, allowing them to interact more readily with neighboring nanowires. This interaction promotes stable assemblies and facilitates grain crystallization and growth, leading to films with superior structural properties. As a result, the prepared films exhibited better crystallization, reduced porosity, and higher density compared to previous reports^{25,26,29,33,45,52} (Fig. 2b, c and Supplementary Figs. 7 and 8). Furthermore, the (013) pole figure analysis revealed high-intensity red and yellow regions, indicating a strong preferential (013) orientation in the Ag_2Se films. This alignment suggests that the nanowires' growth direction optimizes carrier transport by favoring a high σ , which is crucial for enhancing thermoelectric performance. After incorporating rGO into the material matrix, the Ag_2Se grain sizes became even larger (Fig. 2e, f and Supplementary Figs. 9 and 10). This is because in addition to the high intrinsic mobility, rGO can also act as active nucleation sites, promoting the nucleation and growth of Ag_2Se grains during the hot-pressing process^{61,62}. Additionally, rGO increases the specific surface area of the material, improving grain contact and interaction, which further encourages the formation of larger grain⁶³. Remarkably, despite the larger grain sizes, the films retained their highly (013)-orientated feature, maintaining their favorable carrier transport properties. At the macroscopic level, the films exhibit a silver-white metallic luster, indicative of their high quality and dense structure. This dense structure enhances light reflection while reducing transmission

and absorption, as shown in Supplementary Fig. 11, further validating the excellent densification and overall quality of the films⁴⁶.

To explore the micro/nanostructure and composition of Ag_2Se -rGO film, high magnification SEM (Supplementary Fig. 12) and transmission electron microscopy (TEM) (Supplementary Fig. 13) were used for detailed characterization. There are no obvious pores in the Ag_2Se -rGO films. Furthermore, a typical cross-sectional TEM image reveals that amorphous material is surrounded by crystalline regions of Ag_2Se (Fig. 2g). Energy-dispersive spectroscopy (EDS) elemental mapping confirmed the crystalline regions as Ag_2Se and the amorphous regions as rGO. Importantly, the Ag_2Se grains showed excellent integration with the rGO, ensuring a well-connected structure conducive to enhanced thermoelectric performance. High-resolution TEM (HRTEM) images focused on Ag_2Se grains adjacent to rGO (Fig. 2h and Supplementary Fig. 13), particularly in three regions highlighted by yellow, red, and blue squares in Fig. 2h and Supplementary Fig. 13b. These regions displayed clear lattice fringes, indicating excellent crystallinity. Interestingly, all three regions had nearly identical lattice spacings of ~ 0.38 nm, consistent with the (002) plane of Ag_2Se . This uniformity suggests that the regions either belong to a single large grain or represent coherent grain boundaries, which are highly advantageous for charge transport by minimizing electron scattering. The corresponding selected area electron diffraction (SAED) patterns and fast Fourier transform images for the three regions (Fig. 2h and Supplementary Figs. 13e and 13f) revealed identical patterns, confirming their alignment along the Ag_2Se [210] zone axis. Such structural coherence further underscores the high crystallinity and order within the composite material. Additionally, the rGO layer within the composite was found to have a thickness of approximately ~ 29 nm (Fig. 2i). The

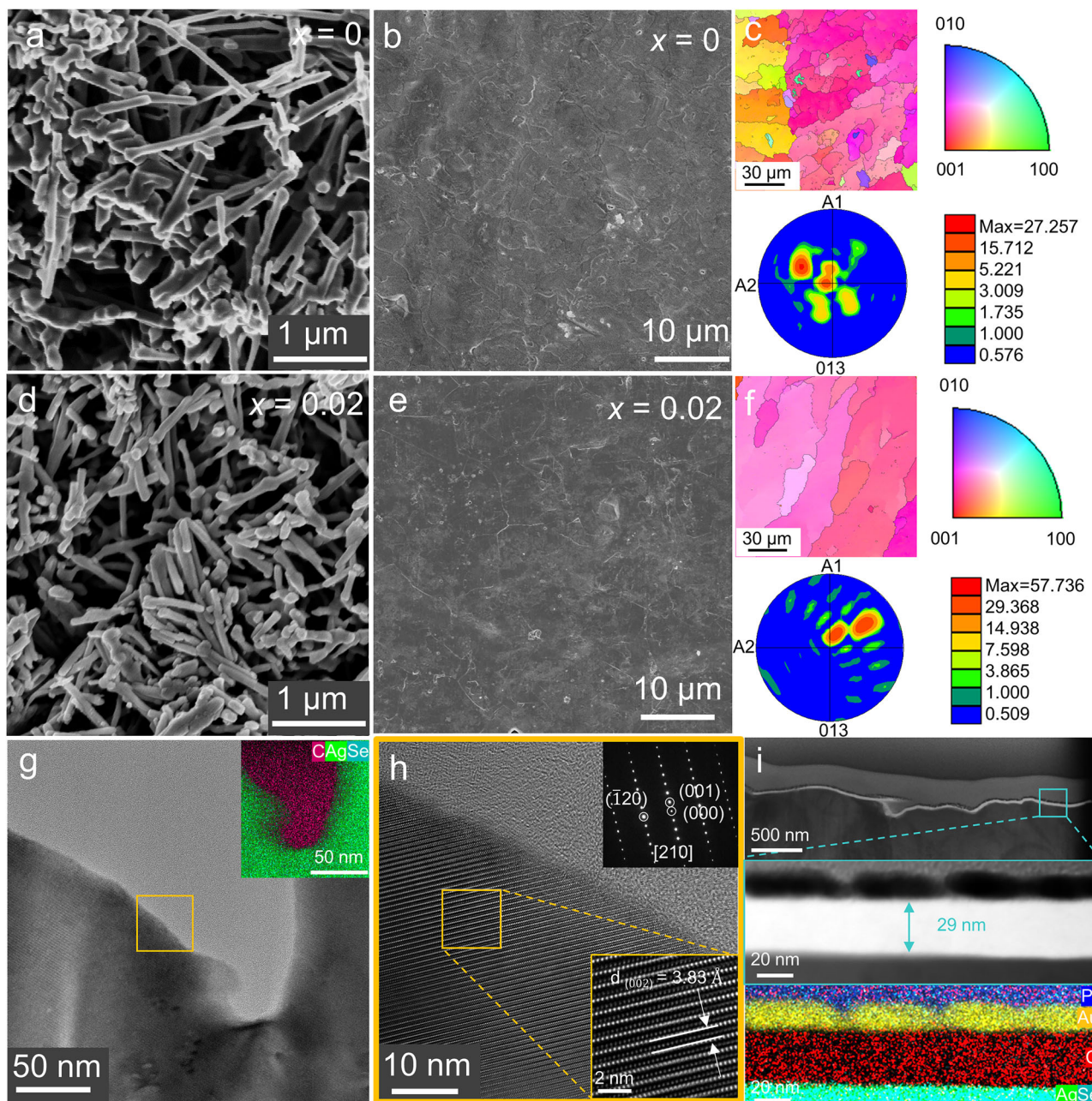


Fig. 2 | Morphological characterization of the $\text{Ag}_2\text{Se} + x \text{ wt\% rGO}$ films.

a, b Scanning electron microscopy (SEM) images of Ag_2Se nanowires and Ag_2Se film, **(c)** Electron backscattered diffraction (EBSD) image and polar diagram along the (013) direction of Ag_2Se film, **d, e** SEM images of $\text{Ag}_2\text{Se} + 0.02 \text{ wt\% rGO}$ nanowires and Ag_2Se film, **f** EBSD images and pole files along the (013) direction of Ag_2Se

film. **g** Transmission electron microscopy (TEM) image (the inset is the energy-dispersive spectroscopy (EDS) mapping of different elements). **h** High-resolution TEM (HRTEM) images of the marked yellow square in **(g)** (the inset is the corresponding enlarged image and SAED pattern). **i** rGO distributed in Ag_2Se film and the corresponding EDS mapping.

thinness of the rGO is attributed to its low concentration in the matrix and the improved dispersion facilitated by hydrogen bonding between ethylene glycol and the oxygen-containing functional groups on rGO. This optimized dispersion ensures that the rGO is uniformly distributed, further supporting its effectiveness in enhancing the thermoelectric properties of the composite⁶⁴.

To evaluate the crystallinity of Se nanowires and Ag_2Se films synthesized at different temperatures, X-ray diffraction (XRD) patterns were analyzed (Supplementary Figs. 14 and 15). The Se nanowires exhibited excellent crystallinity, with all diffraction peaks corresponding to Se (JCPDS #06-0362). For the Ag_2Se films, all peaks corresponded to Ag_2Se (JCPDS #24-1041), confirming the successful

synthesis of the material. Interestingly, as the synthesis temperature increased from 60 °C to 100 °C, the diffraction peaks shifted from (112) and (121) to (013), indicating that optimizing the synthesis temperature is essential for achieving desired orientation of Ag_2Se crystals. After incorporating rGO into the films, a strong (013) orientation was observed, which is beneficial for achieving high μ (Supplementary Fig. 16a)²⁴, aligning with the results from electron backscattered diffraction (EBSD) (Fig. 2f)²⁵. Furthermore, density-functional theory (DFT) calculations indicate that the (013) orientation sets the ground for high mobility and good electronic transport. Since characteristic peaks of rGO are difficult to observe in XRD, Raman spectroscopy was employed to analyze the Ag_2Se -rGO composite films (Supplementary

Fig. 16b). Two prominent peaks were observed at approximately 1344 and 1583 cm^{-1} , corresponding to the D-band and G-band of rGO⁶⁵, confirming the successful incorporation of rGO into the Ag_2Se matrix.

Further composition and bonding analysis were conducted using X-ray photoelectron spectroscopy (XPS) (Supplementary Figs. 16c–f and Supplementary Figs. 17, 18, and 19). For GO, the C 1s peak was deconvoluted into three components located at 284.78, 286.78, and 288.48 eV, corresponding to C–C, C–O, and O–C=O bonds, respectively (Supplementary Fig. 16d)⁶⁶. In the Ag_2Se -rGO composite films, the C 1s spectrum was reduced to two peaks: one at 284.78 eV (C–C) and the other at 286.78 eV (C–O), with the intensity of the C–O peak significantly diminished. This confirms that graphene oxide (GO) was successfully reduced to rGO.

The XPS spectra of Ag_2Se + 0.02 wt% rGO were similar to those of pristine Ag_2Se , with peaks corresponding to Ag 3d_{3/2}, Ag 3d_{5/2}, Se 3d_{3/2}, and Se 3d_{5/2}, respectively (Supplementary Fig. 16c). However, the binding energies of Ag 3d and Se 3d in Ag_2Se -rGO composites showed upward shift compared to pure Ag_2Se (Supplementary Fig. 16e, f), indicating stronger bonding interactions within the composite⁶⁷. These binding energy shifts were also observed in samples with different rGO concentrations (x = 0.01, 0.03, and 0.04 wt%) (Supplementary Figs. 17, 18, and 19). The stronger interfacial interactions between Ag_2Se nanowires and rGO, driven by the high surface energy of rGO, enhance the bonding between the nanowires. This facilitates the formation of more stable large grains during the hot-pressing process, ultimately contributing to the enhanced performance of the composite films^{63,68}.

Thermoelectric properties of Ag_2Se -based films

The thermoelectric performance of Ag_2Se films prepared at different synthesis temperatures is presented in Supplementary Fig. 20. All films exhibit negative S , confirming their n-type conductivity. Notably, while the S shows minimal variation across the samples, the σ increases initially and then decreases as the synthesis temperature rises. The Ag_2Se film synthesized at 90 °C demonstrates the highest σ , reaching approximately 1342 S cm^{-1} at ~300 K. As a result, this film achieves a significantly higher $S^2\sigma$ compared to previously reported Ag_2Se -based thermoelectric materials (Supplementary Table 2).

To enhance the thermoelectric performance of Ag_2Se films, varying concentrations of rGO were incorporated into the film matrix. The addition of rGO increased the absolute S ($|S|$) compared to pure Ag_2Se films (Fig. 3a), primarily by reducing carrier concentration (n , Fig. 3b). Due to the energy filtering effect, low-energy carriers are selectively scattered at the interfaces, leading to a reduction in the number of effective carriers within the composites. However, the μ and energy of the remaining charge carriers are enhanced, ultimately resulting in the synergistic optimization of both σ and S . The σ initially increased before declining with higher rGO concentration. Notably, the Ag_2Se + 0.02 wt% rGO film exhibited the highest σ among all samples in the 300–400 K range, reaching 1481 S cm^{-1} at 300 K (Fig. 3c), comparable to that of Ag_2Se single crystals⁵³. This optimized combination of S and σ , enables the Ag_2Se + 0.02 wt% rGO film achieving a room-temperature $S^2\sigma$ of $-37 \mu\text{W cm}^{-1} \text{K}^{-2}$ (Fig. 3d), representing the highest value reported for Ag_2Se -based thermoelectric materials to date (Fig. 3e and Supplementary Table 2). Moreover, an average $S^2\sigma$ ($S^2\sigma_{\text{avg}}$) between 300 and 390 K reached an impressive $41 \mu\text{W cm}^{-1} \text{K}^{-2}$, signifying a major advancement in Ag_2Se -based thermoelectric research (Supplementary Fig. 21). Additional testing on three samples produced consistent results (Supplementary Fig. 22).

To understand the origins of the high $S^2\sigma$, we analyzed the μ (Fig. 3f and Supplementary Fig. 23). As the doping concentration of rGO increases, the μ of the films first increases and then decreases. Notably, the Ag_2Se + 0.02 wt% rGO film exhibits an exceptional μ of approximately $1500 \text{ cm}^2 \text{V}^{-1} \text{s}^{-1}$ at room temperature, ranking among the highest values reported to date. Even the pristine Ag_2Se film

achieves a high μ of around $1100 \text{ cm}^2 \text{V}^{-1} \text{s}^{-1}$, consistent with previously reported values. The high μ observed in these films can be attributed to two key factors: (1) the pronounced (013) orientation in the Ag_2Se films, which significantly enhances carrier transport; and (2) the incorporation of rGO, which possesses high intrinsic μ and forms a continuous conductive network with Ag_2Se nanowires⁶⁹. This network not only improves inter-nanowire bonding but also facilitates efficient carrier transfer. As a result, all Ag_2Se + x wt% rGO films exhibit higher μ compared to the undoped Ag_2Se counterpart. However, excessive rGO addition (e.g., 0.04 wt%) leads to aggregation and increased carrier scattering, thereby reducing both μ and σ . We further calculated the density-of-states effective mass (m^*), using the single parabolic band (SPB) model (Fig. 3g). Despite the reduction in n with increasing rGO content, m^* remained high. This indicates that the intensive interfaces between rGO and Ag_2Se induce an energy filtering effect, decoupling the S and the σ to enhance both parameters simultaneously. Additionally, the deformation potential energy (E_{def}) of the films was calculated. The results showed that the E_{def} decreased with increasing rGO content (Fig. 3g). A lower E_{def} reflects greater lattice deformability, suggesting improved plasticity. This enhanced plasticity not only maintains structural stability but also improves flexibility during bending, making the composite films highly suitable for flexible thermoelectric applications. Furthermore, the electrical properties of the films were enhanced after the incorporation of an appropriate amount of single-walled carbon nanotubes (SWCNTs) into the Ag_2Se composite (Supplementary Fig. 24). We believe that these experimental findings may provide valuable insights for enhancing the thermoelectric performance of Ag_2Se -based composites in future studies.

To access the thermoelectric performance, we measured the in-plane κ of the films, which was approximately $0.87 \text{ W m}^{-1} \text{K}^{-1}$ at 300 K (Fig. 3h and Supplementary Table 3). Although the film exhibited a relatively high κ_e , the κ_l was successfully suppressed to $0.09 \text{ W m}^{-1} \text{K}^{-1}$ due to the high-intensity Ag_2Se -rGO interfaces. As a result, the ZT value reached -1.28 at 300 K, marking the highest performance reported for Ag_2Se -based films to date (Fig. 1b). This value is comparable to that of state-of-the-art Bi_2Te_3 films and bulk Ag_2Se materials^{22,25–30,34,40,45–53,70,71}, highlighting the exceptional thermoelectric performance of the fabricated films. It should be noted that due to inherent limitations in thermal conductivity measurement techniques of film, the parallel model was used to evaluate the thermal conductivity (the related details were provided in Supplementary Information), where the interfacial thermal resistance between Ag_2Se -rGO film and substrate was ignored, which may limit the accuracy of thermal conductivity measurement. Therefore, the ZT value is for reference only.

To evaluate the practicality of the films, we assessed their stability and flexibility (Fig. 3i and Supplementary Fig. 25). The electrical properties were evaluated over 10 consecutive thermal cycles, with each comprising heating from 300 to 380 K and subsequent cooling back to 300 K. The results revealed negligible changes in σ , S , or $S^2\sigma$, confirming excellent thermal stability. Mechanical flexibility was analyzed by measuring the normalized conductivity (Fig. 3i) (σ/σ_0 , where σ_0 and σ represent the initial and measured conductivity values) as a function of bending cycles. Flexibility improved with increasing rGO content, aligning with the E_{def} trends (Fig. 3g). Notably, the Ag_2Se + 0.02 wt% rGO film retained 96.9% of its initial σ after 1500 bending cycles at a bending radius (r) of 5 mm. Under a bending radius of 4 mm, the σ loss of the Ag_2Se + 0.02 wt% rGO film after 500 bending cycles was less than 4% (Supplementary Fig. 26), demonstrating exceptional flexibility, durability, and suitability for flexible thermoelectric applications (Supplementary Table 2). The main reasons for the enhancement of mechanical flexibility properties by rGO are as follows: First, the unique two-dimensional structure of rGO imparts exceptionally high Young's modulus and tensile strength, which significantly reinforces the mechanical performance of the Ag_2Se -based composite⁷². Second, a small amount of rGO can be uniformly

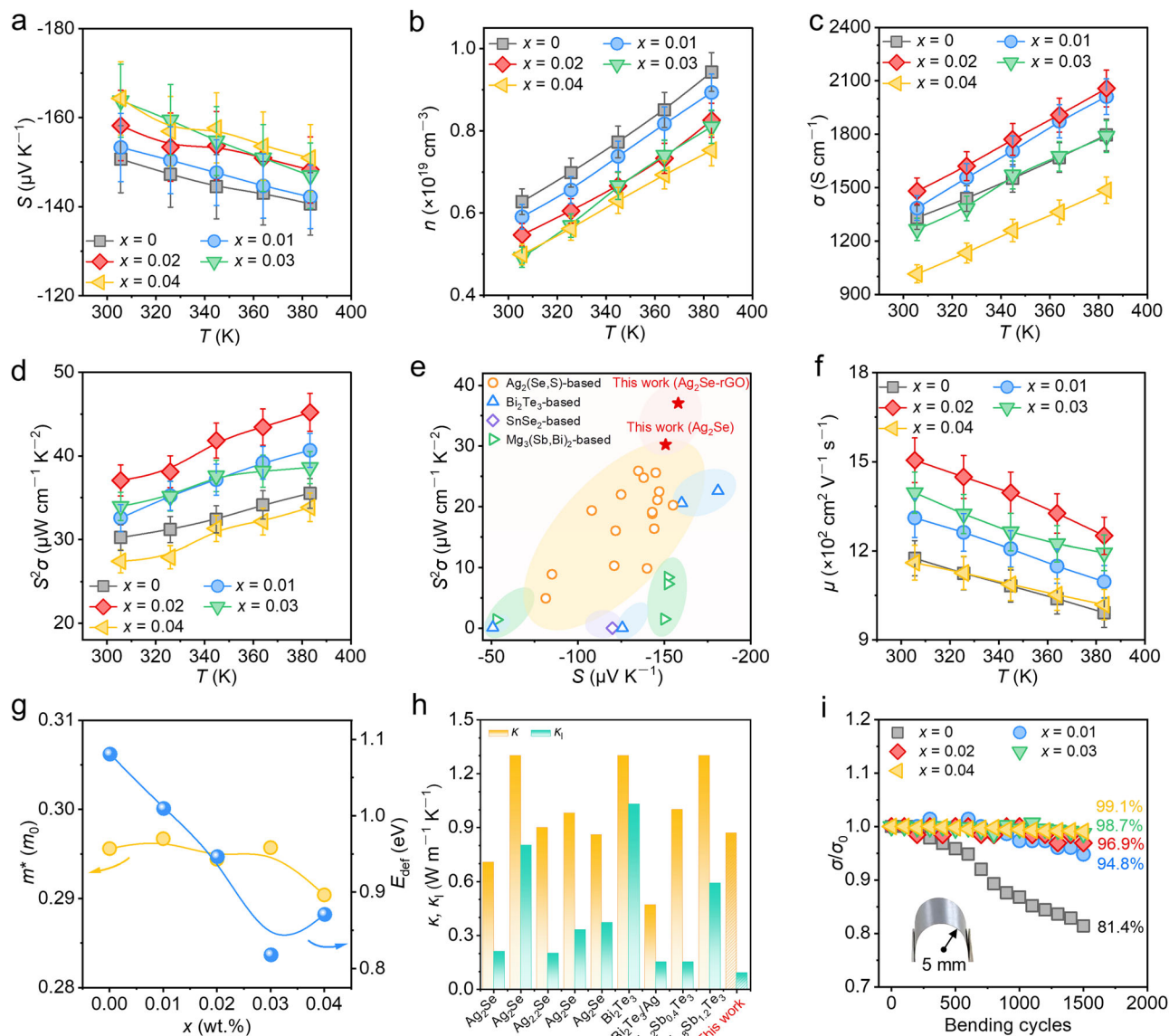


Fig. 3 | Thermoelectric properties, stability, and flexibility of $\text{Ag}_2\text{Se} + x \text{ wt\% rGO}$ films. Temperature-dependent (a) Seebeck coefficient (S), (b) Hall carrier concentration (n), (c) electrical conductivity (σ), and (d) $S^2\sigma$. e Comparison of $S^2\sigma$ for reported films and this work^{6,22,25,26,28–30,34,40,45–49,51–53,75,79–84}. f Temperature-dependent mobility (μ). g x -dependent effective mass (m^*) and deformation

potential (E_{def}). h Room-temperature thermal conductivities (κ) and lattice conductivities (κ_l) between this work and some representative reports^{19,25,53,78,79,85}. i Change in σ/σ_0 dependent on bending cycles for $\text{Ag}_2\text{Se} + x \text{ wt\% rGO}$ films (bending radius $r = 5 \text{ mm}$).

dispersed within and intimately contact the Ag_2Se matrix, forming a continuous conductive network that facilitates optimized stress distribution and improves the overall load-bearing capacity⁷³. Finally, the strong interfacial bonding between rGO and the Ag_2Se matrix enhances stress transfer efficiency, thereby further improving the material's strength and flexibility⁷⁴.

High-performance thermoelectric generator

A soft, comfortable, and breathable fabric substrate was selected for the wearable TEG. A silicone hemisphere was incorporated to create leveraging its specific hardness and compressibility for multiple functions: supporting the thermoelectric legs, enhancing the contact area with the heat source, and establishing a significant temperature gradient. The smooth curvature of the hemisphere ensured strong adhesion to the thermoelectric legs, effectively preventing breakage or damage caused by limited flexibility, thereby improving device stability. Flexible conductive tape was used to connect the thermoelectric

legs, allowing the TEG to bend during wear, while indium solder at the joints ensured reliable electrical contact (Supplementary Fig. 27).

A TEG comprising 100 pairs of thermoelectric legs was fabricated, with dimensions of $14 \times 14 \text{ cm}^2$. The TEG design is highly adaptable and can be customized into various shapes and sizes as required (Supplementary Fig. 28). The open-circuit voltage (V_{oc}) of the TEG increased linearly with the ΔT (Fig. 4a), reaching a maximum of $\sim 408 \text{ mV}$ at $\Delta T = 26 \text{ K}$. Figure 4b, c illustrates the relationship between the device voltage (V), output power (P), load current (I) at different ΔT s. The V decreased proportionally with the I . At $\Delta T = 12 \text{ K}$, the maximum power (P_{max}) was $\sim 15 \mu\text{W}$, corresponding to an internal resistance (R_{in}) of the TEG ($\sim 453 \Omega$). At $\Delta T = 26 \text{ K}$, the P_{max} reached approximately $100 \mu\text{W}$ with the R_{in} reducing slightly to $\sim 449 \Omega$. Under these conditions, the TEG achieved a highly competitive ω of 6.7 mW cm^{-2} and an excellent ω_n of over $9.8 \mu\text{W cm}^{-2} \text{ K}^{-2}$, demonstrating tremendous potential for practical applications (Supplementary Table 4).

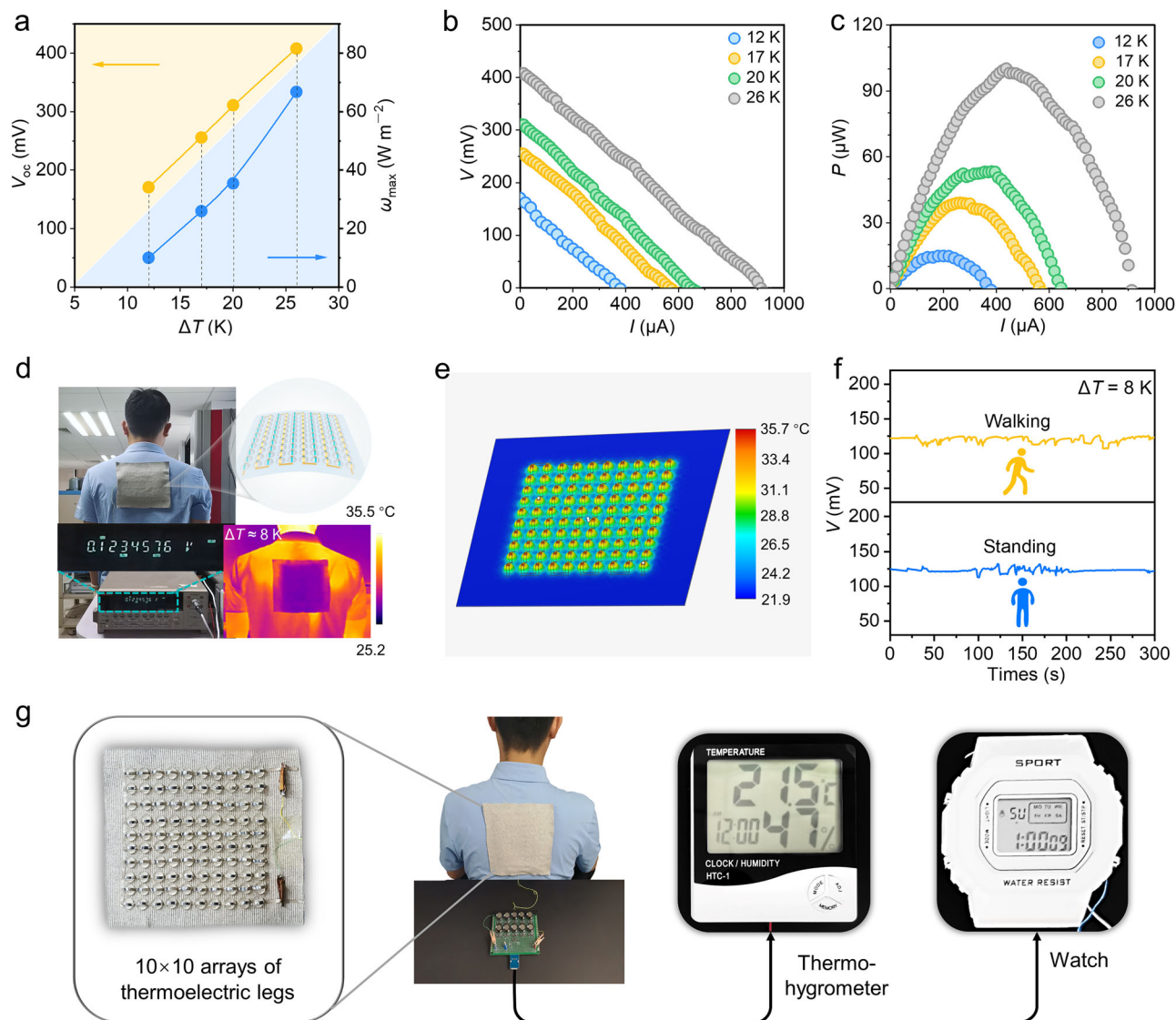


Fig. 4 | Device performance. **a** Open-circuit voltages (V_{oc}) and ω of out-plane TEG with 100-Ag₂Se + 0.02 wt% rGO legs evaluated at various temperature differences (ΔT s). **b** Voltage (V) and **c** output power (P) as a function of current (I) at various ΔT s. **d** Optical images of generating voltage based on the ΔT between the human body and environment (the inset is the corresponding front view of TEG and

infrared thermal image). **e** Temperature distribution of the simulated TEG. **f** Continuous monitoring of the V of the human wearing the TEG at different states. **g** Diagram of charging supercapacitors by using the human body and powering a thermo-hygrometer and watch.

To evaluate the output performance of the TEG in practical applications, the device was evaluated while worn on the back of a human subject (Fig. 4d). At an ambient temperature of 25 °C, the TEG achieved an out-of-plane ΔT of ~8 K, as confirmed by infrared imaging. This ΔT corresponded to a V_{oc} of ~123 mV, closely aligning with the simulated results (Fig. 4e). To monitor the continuous output performance of the TEG, the subject alternated between standing still and walking for 300 s while wearing the device (Fig. 4f). The results demonstrated that stable operation throughout the test, maintaining a V of ~120 mV, underscoring the TEG's reliability and promising potential for practical wearable applications.

To explore the potential of utilizing the energy generated from the ΔT between the human body and the environment, the TEG was used to charge ten parallel capacitors to approximately 70 mV. These capacitors were then reconfigured in series, producing a V of ~0.7 V. This output was subsequently amplified using a boost converter, as depicted in the circuit diagram (Fig. 4g and Supplementary Fig. 29). The generated energy successfully powered a thermo-hygrometer

(1.5 V) and a wristwatch (3 V), demonstrating that the out-of-plane TEG, with a carefully optimized design, can continuously and stably supply sufficient energy for low-power electronic devices. Looking forward, the development of larger-area TEGs to exploit greater temperature gradients could significantly increase energy output, enabling the power supply for higher-power devices such as smartphones. This underscores the promising applications of TEG technology in wearable electronics and sustainable energy solutions.

In summary, we fabricated Ag₂Se/rGO/nylon films with exceptional thermoelectric performance. The films, consisting of Ag₂Se nanowires, exhibit a macroscopic (013) orientation after hot pressing, which enhances the μ and σ . The well-dispersed rGO further boosts the σ and the S through the energy filtering effect, while the intensive interfaces effectively reduce the κ . As a result, the films achieve a record-high $S^2\sigma$ of 37 $\mu\text{W cm}^{-1} \text{K}^{-2}$ and a ZT value of 1.28 at 300 K, establishing a new benchmark for Ag₂Se-based materials. Additionally, an advanced out-of-plane TEG with 100 pairs of thermoelectric legs, incorporating silicone hemispheres, was constructed using these films,

leveraging a giant temperature difference between the human body and the environment to directly power a thermo-hygrometer and a wristwatch. These findings indicate the tremendous potential of Ag₂Se-based materials for practical applications in wearable electronics and health monitoring, paving the way for future energy-harvesting technologies.

Methods

Materials

L-Ascorbic acid, selenium dioxide, and AgNO₃ were purchased from Sinopharm Chemical Reagent Co., Ltd. (China). GO was supplied by The Sixth Element (Changzhou) Materials Technology Co., Ltd. β -Cyclodextrin was obtained from Shanghai Aladdin Biochemical Technology Co., Ltd. Ethylene glycol (EG, >99%) was sourced from Alfa Aesar.

Preparation of Se nanowires

1 g SeO₂ and 1 g β -cyclodextrin were added to 100 mL distilled water and stirred until fully dissolved, forming solution A. Separately, 1 g L-ascorbic acid was added to 100 mL distilled water and stirred until completely dissolved, forming solution B. Solution A was then slowly added dropwise into solution B under continuous stirring, resulting in the mixed product. After stirring for 4 h, the product was extracted, filtered, and centrifuged with absolute ethanol at least three times. Notably, before each centrifugation, the mixture was ultrasonically shaken in a warm water bath (−45 °C) for ~90 s. Finally, the brown Se nanowires were collected by filtration.

Preparation of Ag₂Se-rGO nanosheets

To prevent the agglomeration of graphene in solution, GO was selected as the precursor to synthesize rGO by a reduction reaction in an EG solution. First, a specified amount of AgNO₃ was dissolved in an appropriate volume of EG to form solution C. Next, a predetermined quantity of the as-prepared Se nanowires ($n_{\text{Ag}}:n_{\text{Se}} = 2:1$) and GO were added to solution C, and the mixture was stirred at 90 °C for 2 hours to ensure a sufficient reaction. After cooling to room temperature, ethanol was added to precipitate the Ag₂Se + x wt% rGO nanosheets. This process was repeated three times to remove any residual reactants.

Preparation of Ag₂Se-rGO composite films

Ag₂Se + x wt% rGO composite films were fabricated through vacuum filtration using a nylon membrane. The resulting films were dried at 80 °C for 5 min and then hot-pressed at 200 °C for 30 min under a pressure of 1 MPa. Based on the weight percentages of rGO in the composite films (0.01, 0.02, 0.03, and 0.04 wt%), the samples were designated as $x = 0$, $x = 0.01$, $x = 0.02$, $x = 0.03$, and $x = 0.04$, respectively.

Assembly of the flexible thermoelectric generator

Silicone hemispheres with a diameter of 5 mm, arranged in a 10 × 10 array, were fixed onto a flexible and breathable fabric at regular intervals. The film of Ag₂Se + 0.02 wt% rGO was selected as thermoelectric legs due to its excellent performance. Thermoelectric legs with dimensions of 8 × 2.5 mm² were attached to the silicone hemispheres, with both ends of the legs coated with silver paste. Once the silver colloid solidified, the other half of the hemispheres was covered with flexible conductive tape to connect the thermoelectric legs. To further minimize contact resistance and improve the stability of the device, indium solder was used to weld the Ag₂Se-rGO legs to the conductive tapes, ensuring the reliable electrical connection. The temperature applied at the top of the hemisphere was taken as the high-temperature end, while the temperature of the fabric was taken as the low-temperature end, taking the difference between the two as the ΔT .

Characterization

Morphology analysis was performed using scanning electron microscopy (SEM, ZEISS SIGMA, FEI-Siron), EBSD (EDAX OIM 6.0), and TEM (Thermo Scientific Talos F200S G2 200 kV). The TEM samples were prepared using a focused ion beam (FIB, Thermo Scientific Scios 2 DualBeam). Structural analysis was conducted using XRD (Bruker D8 four-circle diffractometer), Raman spectroscopy (Renishaw inVia, with an excitation wavelength of 532 nm), XPS (Thermo Scientific K-Alpha), and ultraviolet photoelectron spectroscopy (UPS, Thermo Fisher Scientific ESCALAB XI+). The temperature-dependent S and σ were measured simultaneously using commercial equipment (ZEM-3, ULVAC Riko, Japan) in a helium atmosphere (within $\pm 5\%$ uncertainty). The S and σ of all specimens were measured using bilayer films deposited on a nylon substrate, with a specimen size of 5 × 15 mm². As the nylon substrate is non-conductive, the measured thickness corresponds solely to that of the thermoelectric films (Ag₂Se-rGO). n and μ were determined using a Van der Pauw Hall measurement system (NYMS), with a measurement uncertainty of $\sim 5\%$. Flexibility was measured by fixing one end of the film, while the other end was bent according to a pre-determined bending radius. After completing a specified number of bending cycles, the resistance at both ends of the film is measured to obtain the relevant data.

Data availability

The data generated in this study is provided in the Source Data file. Source data are provided with this paper.

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Author contributions

F. D., Z.-G.C. and H. S. supervised the overall experiments. L.Z. designed the related experiments. L.Z., H.D., D.H., and X.W. prepared materials and measured the thermoelectric properties, designed device structures, fabricated devices, and measured the performance. H.S., X.-L.S., H.G., F.D., and Z.-G.C. analyzed the data. W.C. performed the device numerical simulation and M.L. conducted the DFT calculations. H.S., H.G., and X.-L.S. discussed the results. L.Z. wrote the manuscript with the help of all the authors.

Competing interests

The authors declare no competing interests.

Additional information

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