

Synthesis of benzoheterocycles by palladium-catalyzed migratory cyclization through an unexpected reaction cascade

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Migratory functionalization of C–H bonds through metal migration from carbon to carbon under transition metal catalysis is a process of significant academic and industrial interest. Herein, a palladium-catalyzed migratory cyclization of α -bromoalkene derivatives ArXCBr=CH_2 , in which X denotes a phosphorus (P(O)R), silicon (SiR_2), sulfur (SO_2), carbon (C(O)), nitrogen (NTs), or oxygen-based moiety, affording various benzoheterocyclic compounds has been developed. Mechanistic investigations have demonstrated that the cyclization reaction proceeds through an unexpected cascade, with *trans*-1,2-palladium migration between sp^2 carbons being a key step of catalytic cycle. To the best of our knowledge, this type of metal migration has not been reported previously.

Migratory functionalization of C–H bonds through metal migration from carbon to carbon under transition metal catalysis is a process of significant academic and industrial interest^{1–12}. It provides a non-classical means of selectively installing a functional group at a remote C–H position using simple precursors, thus enabling the direct synthesis of challenging structures not accessible through traditional cross-couplings. Most notably, migratory functionalization of alkenes or alkyl halides through a 1,2-metal shift along a sp^3 chain and cross-coupling has been well developed for remote C–H bond functionalization (Fig. 1)^{8–12}. Migratory functionalization via *cis*-1,2-palladium migration between sp^2 carbons has been rarely reported^{13–19}. Another metal migration frequently exploited is a 1,4-metal shift in many transition metal-catalyzed tandem reactions^{1,4,5}.

Indenone and five-membered benzoheterocycles are ubiquitous structural motifs found in a wide array of natural products, pharmaceuticals, and agrochemicals (Fig. 2a)^{20,21}. In particular, benzophosphole is widely used in developing functional materials^{22–27}, biologically active compounds^{28–33}, and chiral ligands^{34–37}. The transition metal-catalyzed intramolecular coupling of aryl halides or aryl metals with alkenes has become a versatile tool in heterocycle synthesis (Fig. 2b,

top)^{38–43}. However, this traditional strategy requires the prior introduction of a halo or metal group at the *ortho* position of the aryl ring. Under this premise, we wondered whether an alternative synthetic route based on a cascade-type process involving 1,*n*-metal migration from an alkenyl carbon atom to an aryl carbon atom might be designed, thus setting the stage for carbometallation of a C=C double bond (Fig. 2b, bottom). Hayashi reported the Rh-catalyzed cyclization of arylpropargyl alcohols via a 1,4-Rh shift–aryl rhodation sequence, which validates our hypothesis^{44–46}. The very wide availability of mono-substituted olefins ArX-CH=CH_2 and their facile conversion into ArX-CBr=CH_2 by simple bromination and elimination should make this new strategy very synthetically useful. At the outset of our investigations, however, it was unclear whether such a strategy could be implemented as elimination from ArX-CBr=CH_2 to give alkynes could be envisaged as being highly feasible.

Herein, we report the successful realization of this goal through a palladium-catalyzed migratory cyclization of α -bromoalkene derivatives ArXCBr=CH_2 **1**, in which the X moiety is a phosphine oxide, silyl, sulfonyl, carbonyl, amide, or oxygen atom/group, to give benzoheterocycles **2** (Fig. 2c). In this process, two C–H bonds are simultaneously cleaved and

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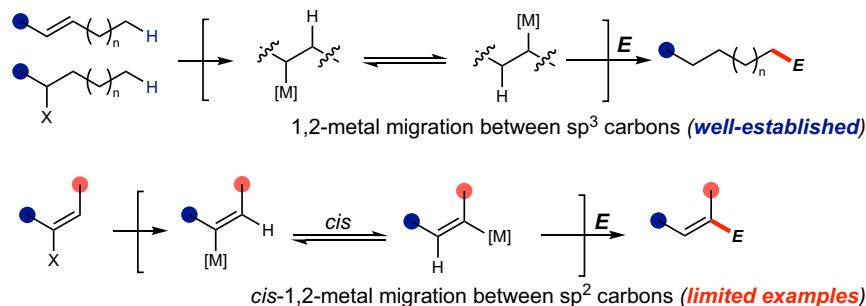


Fig. 1 | Migratory functionalization via 1,2-metal migration. 1,2-Metal migration between sp^3 or sp^2 carbons in catalytic cycles.

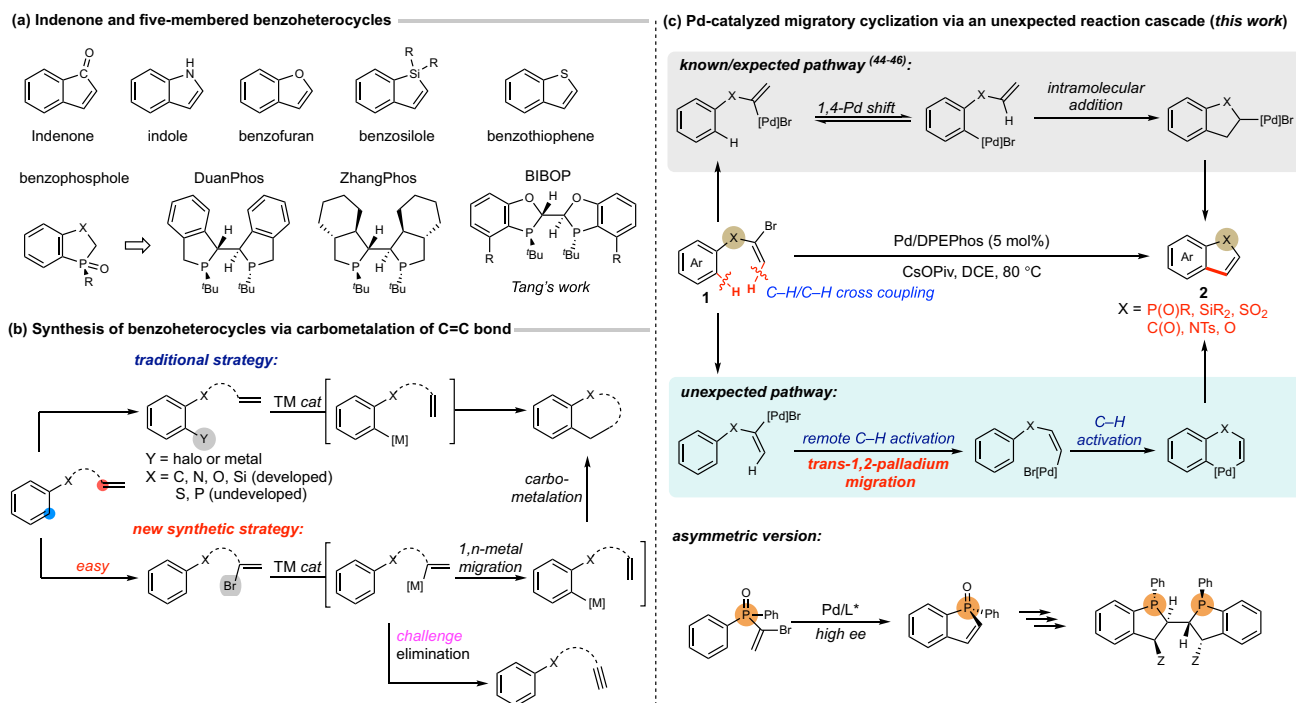


Fig. 2 | Transition metal-catalyzed migratory cyclization. **a** Indenone and five-membered benzoheterocycles. **b** Strategies to synthesize benzocycles. **c** This work: Pd-catalyzed migratory cyclization via an unexpected reaction cascade.

coupled to form a new carbon–carbon bond, whereas transition metal-catalyzed C–H/C–H coupling of arenes and alkenes always occurs via an aryl C–H metalation followed by a Heck-type alkenylation process^{47–54}. However, the reaction does not proceed through an oxidative addition–1,4-palladium migration^{55–61}–arylpalladation sequence as we expected. Detailed mechanistic studies have shown that *trans*-1,2-metal migration from the α -position to the *trans*- β -position of the C=C double bond is a key step of the catalytic cycle. This migratory cyclization is synthetically very useful and offers opportunities for the efficient synthesis of indoles and their phosphorus, silicon, sulfur, carbon, and oxygen congeners. Previously, 2,3-unsubstituted benzophosphole was prepared by ring-closing metathesis of phenylstyrylvinylphosphine oxide^{62,63}. As reported herein, the present method enabled the efficient synthesis of benzophosphole oxides, and the asymmetric version of the cascade reaction was also achieved in the presence of a chiral palladium catalyst to give enantio-enriched P-chiral products.

Results

Optimization of the reaction conditions

In a first set of experiments, the cyclization of (α -bromoethenyl) diphenylphosphine oxide (**1a**) was performed in the presence of 5 mol % of palladium catalysts bearing several types of bisphosphine ligands

and 2.0 equiv. of CsOPiv (OPiv = pivalate) in 1,2-dichloroethane (DCE) at 80 °C for 12 h (Table 1). The reaction with DPEPhos as a ligand gave a 94% yield of the cyclization product **2a** (entry 1). An alternative wide-bite-angle ligand, Xantphos, was not effective for the present reaction (entry 2). Some other bisphosphine ligands, such as dppf, dppe, dppp, dppb, and biphep, gave **2a** as the main product in somewhat lower yields (entries 3–7). The reaction performed in toluene afforded **2a** in moderate yield, whereas in CH_2Cl_2 it gave a mixture of **2a** and **2a'** in an 85:15 ratio (entries 8 and 9). The use of other bases, such as CsOAc and KO₂Piv, furnished **2a** in moderate yields with medium chemoselectivities (entries 10 and 11). The reaction performed at a lower temperature (60 °C) gave **2a** in a slightly lower yield (entry 12).

Pd-catalyzed migratory cyclization: substrate scope

Figure 3 summarizes the results obtained for the migratory cyclization of other $\text{ArXCBr}=\text{CH}_2$, in which X denotes phosphorus or another element, under the optimized conditions. Cyclizations of (α -bromoethenyl)diarylphosphine oxides **1b–1g**, in which the aromatic groups are phenyl moieties substituted with methyl, *t*-butyl, phenyl, methoxy, fluoro, and trifluoromethyl at the *para* position, proceeded well, giving the benzophospholes **2b–2g** in high yields, irrespective of the electronic properties of the substituents (Fig. 3). (α -

Table 1 | Optimization of the reaction conditions

Entry	Variations from standard conditions ^a (shown above)	2a:2a ^b	Yield (%) ^c 2a
1	None	>50:1	94
2	Xantphos	>50:1	41
3	Dppf	>50:1	82
4	Dppe	>50:1	46
5	Dppp	>50:1	13
6	Dppb	>50:1	23
7	Biphep	–	–
8	Toluene instead of DCE	>50:1	47
9	CH ₂ Cl ₂ instead of DCE	85:15	78
10	CsOAc instead of CsOPiv	74:26	63
11	KOPiv instead of CsOPiv	89:11	72
12	At 60 °C	>50:1	87

^a Reaction conditions: **1a** (0.20 mmol), Pd(OAc)₂ (5 mol% Pd), DPEPhos (10 mol%), CsOPiv (0.4 mmol), and DCE (1.0 mL) at 80 °C for 12 h.

^b Ratio was determined by ¹H NMR analysis of the crude reaction mixture.

^c Isolated yield.

Bromoethenyl)diarylphosphine oxides **1h–1k**, in which the aromatic groups are phenyl moieties substituted with methyl, methoxy, and trifluoromethyl at the *meta* position, all proved suitable for this reaction, affording exclusively the benzophospholes **2h–2k** in high yields, and the new C–C bonds were formed with high regioselectivity at the less hindered *ortho* C–H position. Moreover, *ortho*-substituted phenyl derivative **1l** gave 70% yield of the migratory cyclization product **2l**, the yield being lower due to some generation of the elimination by-product. The reaction of **1m**, bearing a benzo[*b*]thiophen-5-yl group, gave 67% yield of the corresponding product **2m** with high regioselectivity. The migratory cyclizations of alkyl-substituted (α -bromoethenyl)phenylphosphine oxides **1n–1r** also proceeded, furnishing the corresponding products **2n–2r** in moderate yields. The yields of **2n–2r** were moderate due to some generation of the elimination by-product.

The robustness of the present protocol was further demonstrated by the synthesis of other benzoheterocyclic compounds, in which the X moiety is an atom or group other than phosphine oxide (Fig. 3). It was found that the migratory cyclization of α -bromovinyl ketone **1s**, in which the X moiety of the substrate is a carbonyl group, was viable, giving the inden-1-one product **2s** in a high yield under the standard conditions. *N*-Tosylindoles **2t–2v**, in which X is a tosyl-protected amide group, were also efficiently obtained by using the present methodology. When the X moiety of the substrate was an oxygen atom, the migratory cyclization proceeded smoothly to afford benzofuran **2w** in 73% yield. Moreover, unsubstituted 1-silaindene **2x** was obtained in 63% yield under the present conditions. Benzo[*b*]thiophene 1-oxide and 1,1-dioxides are an emerging class of heterocyclic compounds with synthetic and medicinal chemistry applications^{64–68}. Fortunately, the reaction of α -bromovinylphenyl sulfoxide **1y** gave the benzo[*b*]

thiophene 1-oxide **2y** in 83% yield. The cyclizations of α -bromovinyl aryl sulfones **1z–1ab** also proceeded smoothly to afford benzo[*b*]thiophene 1,1-dioxides **2z–3ab** in high yields.

Asymmetric synthesis of P-chiral benzophospholes

To explore the potential of this methodology in asymmetric synthesis, a preliminary screening of chiral bisphosphine ligands was carried out. As shown in Fig. 4, reaction of **1a** in the presence of a chiral (*R*)-DM-segphos-palladium catalyst gave **2a** in 31% isolated yield with a promising 86% ee, whereby the relatively low yield was due to low conversion of **1a**. The asymmetric cyclization of (α -bromoethenyl)diarylphosphine oxide **1b**, in which the aromatic group is phenyl moiety substituted with methyl at the *para* position, gave the benzophosphole **2b** in 37% yield with 67% ee. The reaction of **1f**, bearing fluoro substitution at the *para* position of phenyl ring, gave the asymmetric cyclization product **2f** in 33% yield with 71% ee. The reaction of **1k**, bearing a 2-naphthyl group, gave 44% yield of the asymmetric cyclization product **2k** with 57% ee and high regioselectivity. The asymmetric cyclization of **1m**, bearing a benzo[*b*]thiophen-5-yl group, gave 29% yield of the corresponding product **2m** with 74% ee and high regioselectivity.

Mechanistic studies

Subsequently, we conducted preliminary experiments to provide insight into the reaction mechanism. We initially envisaged that the cyclization might proceed via an alkenyl-to-aryl 1,4-Pd migration as a key step. However, when the cyclization reaction of α -bromovinyl di(pentadeuteriophenyl)phosphine oxide (**1a-d₁₀**) was carried out under the standard conditions, the product was **2a-d₉**, bearing only nine deuterium atoms at the aryl carbon atoms (Fig. 5a, equation 1).

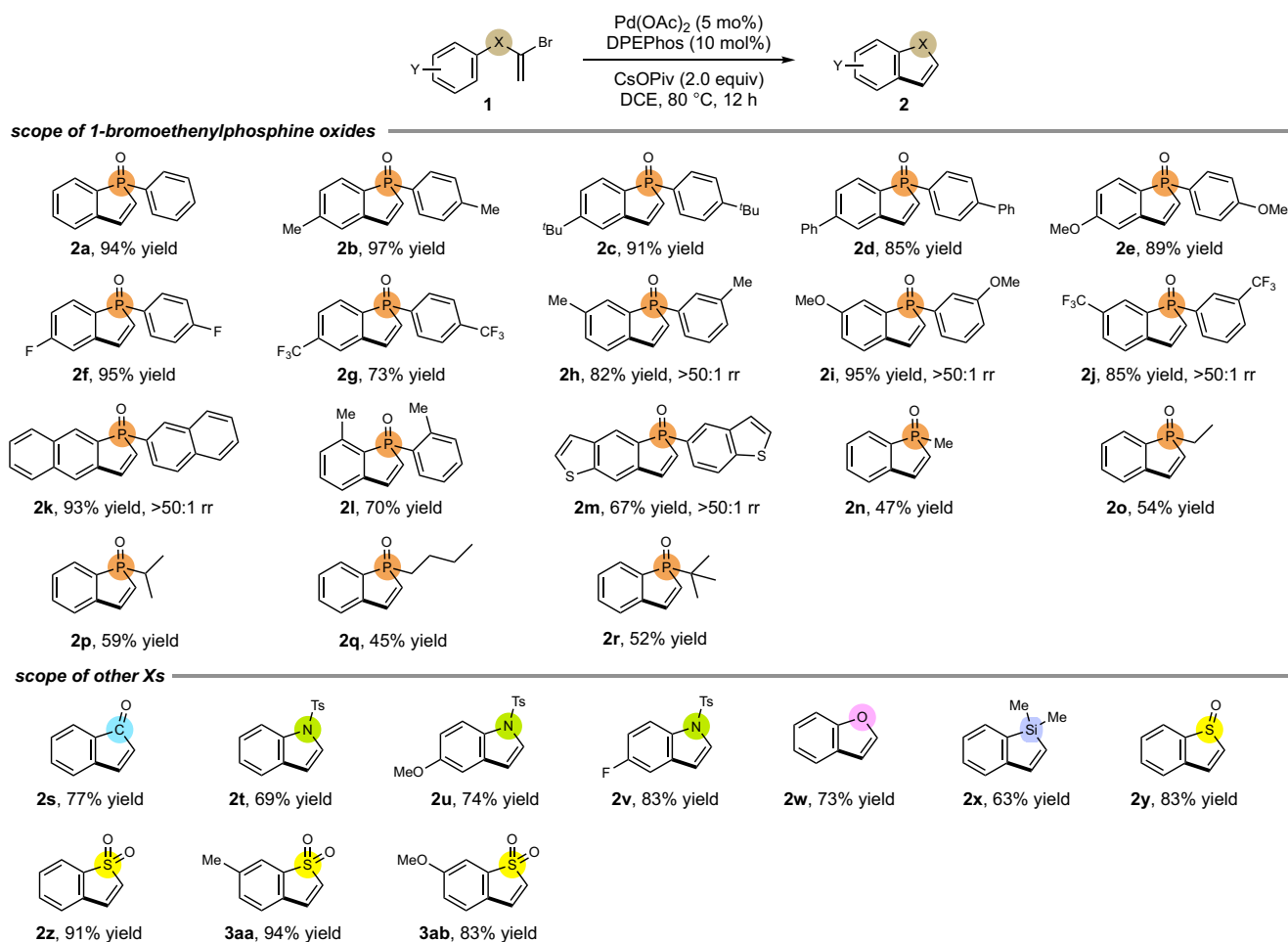


Fig. 3 | Pd-catalyzed migratory cyclization: substrate scope. Reaction conditions: **1** (0.20 mmol), Pd(OAc)₂ (5 mol% Pd), DPEPhos (10 mol%), CsOPiv (0.40 mmol), and DCE (1.0 mL) at 80 °C for 12 h. Isolated yield.

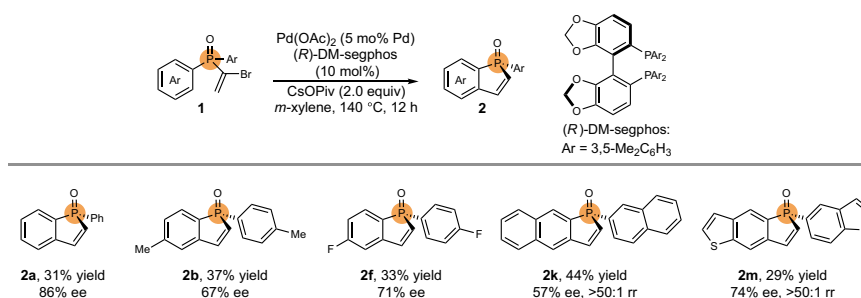


Fig. 4 | Asymmetric synthesis of P-chiral benzophospholes. Reaction conditions: **1** (0.20 mmol), Pd(OAc)₂ (5 mol% Pd), (R)-DM-segphos (10 mol%), CsOPiv (0.40 mmol), and *m*-xylene (1.0 mL) at 140 °C for 12 h. Isolated yield. The % ee was determined by HPLC on a chiral stationary phase column.

Furthermore, cyclization of α -bromovinyl di(2-deuteriophenyl)phosphine oxide (**1a-d₂**) under our standard conditions proceeded smoothly to give a mixture of **2a-d₂** and **2a-d₁** in a 2:1 ratio, with one or two deuterium atoms incorporated at the aryl carbon atoms (Fig. 5a, equation 2). No deuterium incorporation at the alkenyl carbon atoms of **2a-d₉**, **2a-d₂**, and **2a-d₁** clearly demonstrated that no 1,4-palladium migration was involved in the catalytic cycle. The fact that reaction of **1z-d₅** afforded benzothiophene sulfone **2z-d₄** with no deuterium incorporated at the alkenyl carbon atoms further confirmed this conclusion (Fig. 5a, equation 3). Reaction of ethynylphosphine oxide (**2a'**) in the presence of the DPEPhos-palladium catalyst, NEt₃·HBr, and CsOPiv gave the target product **2a** in 37% yield (Fig. 5b).

As shown in Fig. 5c, cyclization of *Z*-(β -bromoethenyl)diphenylphosphine oxide (**3**) gave a 29% yield of **2a**. This indicated that the migratory cyclization might involve intramolecular C–H bond activation to form the C–C bond. Under the standard conditions, compound (*E*)-**4** afforded the elimination product **5** in high yield without formation of the cyclization product, whereas no reaction of its (*Z*)-isomer took place (Fig. 5d).

DFT calculations and proposed mechanism

Control experiments, in concert with DFT calculations^{69–71} (Fig. 6), were used to inform the development of a mechanistic model. The density functional theory (DFT) calculations at the M06L/6-311+G(d,p)

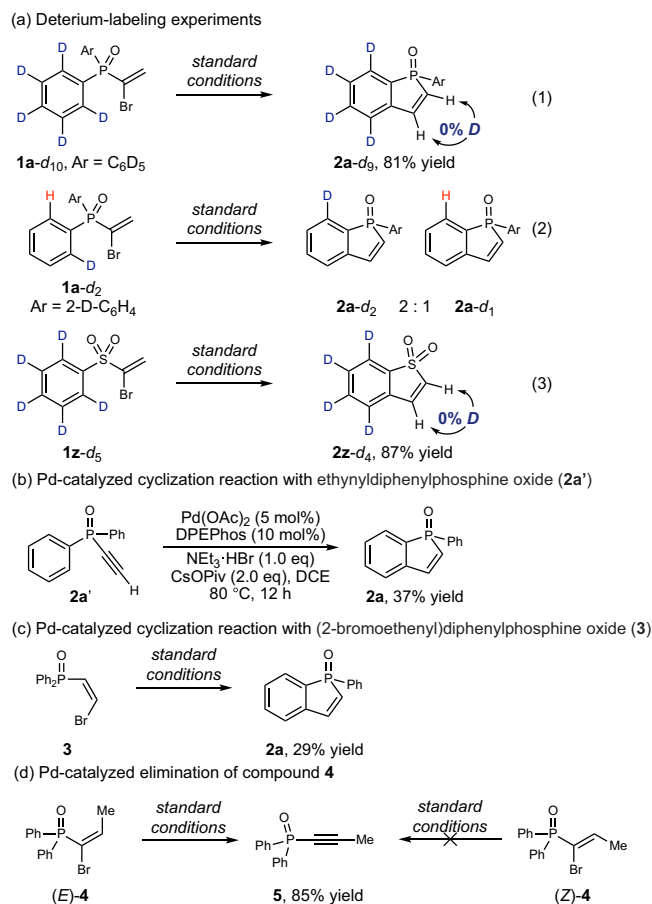


Fig. 5 | Mechanistic studies. **a** Deterium-labeling experiments. **b** Pd-catalyzed cyclization reaction with ethynylphosphine oxide (2a'). **c** Pd-catalyzed cyclization reaction with (2-bromoethenyl)diphenylphosphine oxide (3). **d** Pd-catalyzed elimination of compound 4.

(LANL2TZ for palladium atoms)/SMD(DCE)/B3LYP-D3/6-31G(d) (LANL2DZ for palladium atoms) level was performed to gain a theoretical understanding of the reaction mechanisms (see Supplementary Page S37). We chose the conversion of **1a** to **2a** as the model reaction, using Pd(OAc)₂ as the catalyst and DPEPhos as the ligand. Taking into account the coordination capability of olefins, the free energy profile is initiated by the Pd(0) species **INT-1**, wherein the ligand and the C=C bond are coordinated to the metal center. Subsequent C–Br bond oxidative addition to Pd center can occur via a three-membered ring-type transition state **2-ts** to form vinyl Pd(II) intermediate **INT-3** with an energy barrier of 22.9 kcal/mol, meanwhile releasing 7.5 kcal/mol of energy. In the presence of base, anion exchange of intermediate **INT-3** with OPiv leads to the formation of the thermodynamically stable Pd(II)-OPiv intermediate **INT-4**. In the generated Pd(II)-OPiv intermediate **INT-4**, the Pd-OPiv and Pd-ethylene bond lengths are 2.11 and 2.03 Å, respectively, indicating weak Pd-O bonding. An outer-sphere deprotonation with the assistance of base then takes place to give the alkyne-coordinated Pd(II) intermediate **INT-6** via transition state **5-ts**, which is endergonic 5.0 kcal/mol. The energy barrier for this step is 23.7 kcal/mol. The Pd-C1 bond length and Pd-C2 bond length of Pd-ethylene in the transition state **5-ts** are 2.07 and 2.40 Å, respectively, which indicate that Pd center can activate the vinyl moiety. Sequential protonation occurs to form *cis*-vinyl Pd(II) intermediate **INT-8** via transition state **7-ts** with the energy barrier of 19.9 kcal/mol. The formed *cis*-vinyl Pd(II) intermediate **8** can easily isomerize to a *trans*-one **INT-9**^{72,73}. The intramolecular phenylic C–H bond activation^{74–79} of

diphenylphosphine oxide happens through a concerted metalation-deprotonation (CMD) process via transition state **10-ts** to form a six-membered palladacycle **INT-11** with 1.4 kcal/mol exergonic, overcoming an energy barrier of 23.2 kcal/mol. Then a C–C bond reductive elimination generates the final product and regenerates Pd(0) species **INT-1** with an energy barrier of 12.7 kcal/mol. Meanwhile, we also explored the alternative blue and green pathways, which involves direct β -hydride elimination and phenylic C–H bond concerted metalation-deprotonation of intermediate **INT-4**. These pathways, however, have energy barrier of 27.5 and 44.1 kcal/mol, respectively, which are 3.8 kcal/mol and 20.4 kcal/mol higher than the pathway involving **5-ts**, indicating that these pathways are less favorable (see Supplementary Fig. 1).

According to the DFT calculations, we propose a plausible mechanism for the palladium-catalyzed migratory cyclization in Fig. 7. The mechanism involves C–Br oxidative addition of vinyl bromide, base-assisted outer-sphere deprotonation, sequential protonation, phenylic C–H bond concerted metalation-deprotonation and C–C bond reductive elimination. For comparative analysis, the direct β -hydride elimination pathway and the concerted metalation-deprotonation mechanism for the phenylic C–H bond are investigated using density functional theory; however, both pathways are determined to be energetically disfavored. The key *trans*-1,2-Pd migration core is driven by the cooperative C–H activation of the alkene coordinated to Pd. The electron-deficient Pd(II) center polarizes both the α - and β -carbons of the coordinated ethene, significantly weakening the terminal C–H bond. This electronic perturbation enables HOPIV to participate in an unconventional outer-sphere proton abstraction, bypassing the classical inner-sphere pathway. The resulting Pd–H intermediate undergoes stereochemical reversal migration and insertion, establishing a migration pathway distinct from the traditional β -hydride mechanism.

Synthesis of P-chiral bisphosphine ligands

Chiral bisphosphine ligands are of key importance in transition metal-catalyzed asymmetric synthesis of optically active products^{34,37,80–83}. Enantio-enriched **2a** obtained above was reduced to 1-phenylphosphindane (*R_P*)-**6** in the presence of Pd/C and H₂ in 94% yield with >99.5% ee after recrystallization from methanol (Fig. 8a). Moreover, P-chiral bisphosphine ligand **L1** was easily prepared by treatment of (*R_P*)-**6** with the strong base lithium diisopropylamide and CuCl₂ in THF, followed by reduction with HSiCl₃/NEt₃ to give the product in an overall yield of 46%. Asymmetric addition of 4-MeOC₆H₄B(OH)₂ to **2a** in the presence of a chiral diene*-Rh catalyst proceeded smoothly to afford 3-arylated phosphindane (*S_P*,*R_C*)-**7** in 42% yield with >99.5% ee after recrystallization from methanol (Fig. 8b). Following the above steps, P-chiral bisphosphine ligand **L2** was also synthesized in an overall yield of 32%. **L1** and **L2** were then successfully directly used as chiral ligands in asymmetric hydrogenation to give compound **8** with 92% ee and 97% ee, respectively (Fig. 8c). The Hayashi–Miyaura reaction^{84–91} of cyclohexanone and PhB(OH)₂ in the presence of the Rh/**L1** or Rh/**L2** catalysts also proceeded smoothly to give the 1,4-addition product **9** with >99% ee and 98% ee, respectively (Fig. 8d).

Discussion

In summary, we have reported a Pd-catalyzed migratory cyclization of ArXCBr=CH₂ to give In summary, we have reported a Pd-catalyzed migratory cyclization of ArXCBr=CH₂ to give benzoheterocycles, specifically indoles and their phosphorus, silicon, sulfur, carbon, and oxygen congeners. Detailed mechanistic studies have shown that the *trans*-1,2-palladium migration from the α -position to the *trans*- β -position of the C=C double bond is a key step of the catalytic cycle. The applicability of the present method has been showcased through the synthesis of new P-chiral bisphosphine ligands.

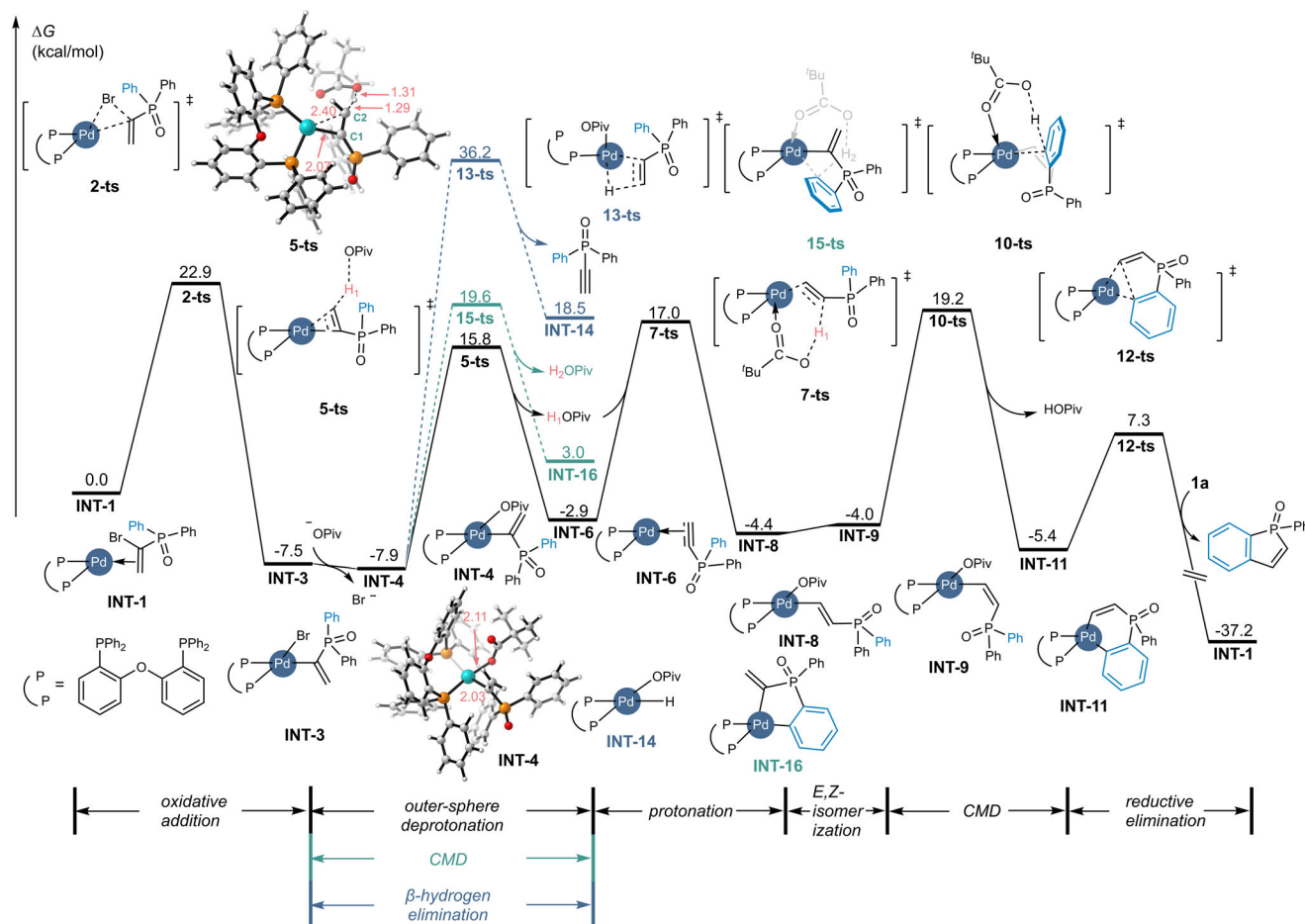


Fig. 6 | Free energy profile and selected transition state and intermediate structures for benzoheterocycles by palladium-catalyzed migratory cyclization. The energies are reported in kcal/mol and represent the relative free energies

calculated using M06L/6-311 + G(d,p)(with LANL2TZ for palladium)/SMD(DCE)//B3LYP-D3/6-31G(d) (with LANL2DZ for palladium).

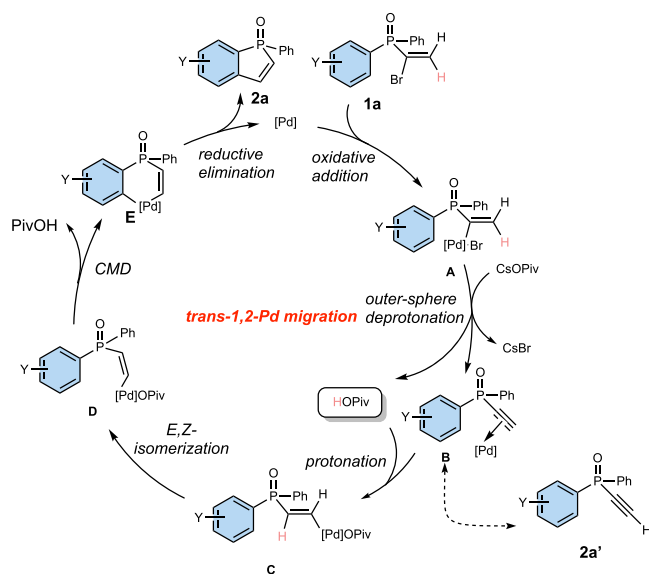


Fig. 7 | A proposed catalytic cycle for the palladium-catalyzed migratory cyclization of 1a. The catalytic cycle is proposed according to the control experiments and the DFT calculations.

Methods

A typical procedure for palladium-catalyzed migratory cyclization of $\text{ArXCBr}=\text{CH}_2$ (Table 1, entry 1)

An oven-dried sealed tube equipped with a PTFE-coated stir bar was charged with $\text{Pd}(\text{OAc})_2$ (2.24 mg, 10 μmol , 5.0 mol% of Pd), **1a** (61.2 mg, 0.20 mmol), DPEPhos (10.8 mg, 20 μmol), CsOPiv (93.6 mg, 0.40 mmol) under argon. DCE (1.0 mL) was added successively, and the mixture was stirred at 80 °C for 12 h. The reaction mixture was passed through a short column of silica gel with dichloromethane as eluent and the water stayed in silica gel. The solvent was removed on a rotary evaporator. After ^1H NMR analysis of the residue, the crude product was subjected to silica gel chromatography (eluent: dichloromethane/methanol (30/1)) as the eluent to give **2a** (42.5 mg, 94% yield, 0.19 mmol) as a green solid.

A typical procedure for palladium-catalyzed asymmetric migratory cyclization of $\text{ArXCBr}=\text{CH}_2$ (Fig. 4)

An oven-dried sealed tube equipped with a PTFE-coated stir bar was charged with $\text{Pd}(\text{OAc})_2$ (2.24 mg, 10 mmol, 5.0 mol% of Pd), **1a** (61.2 mg, 0.20 mmol), (*R*)-DM-segphos (14.5 mg, 20 mmol), CsOPiv (93.6 mg, 0.40 mmol) under argon. *m*-Xylene (1.0 mL) was added successively, and the mixture was stirred at 140 °C for 12 h. The reaction mixture was passed through a short column of silica gel with dichloromethane as eluent and the water stayed in silica gel. The solvent was removed on a rotary evaporator. After ^1H NMR analysis of the residue, the crude product was subjected to silica gel chromatography (eluent: dichloromethane/methanol (30/1)) as the eluent to give (*R*)-**2a** (14.0 mg, 31% yield, 0.062 mmol) as a green solid.

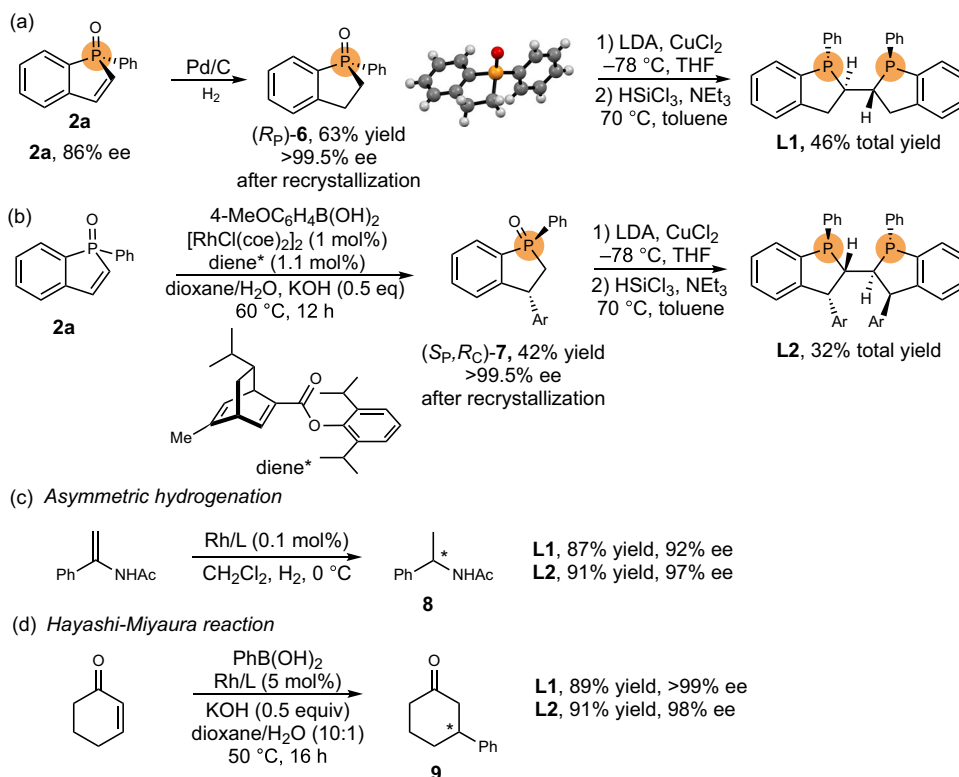


Fig. 8 | Synthesis of P-chiral bisphosphine ligands. a, b Asymmetric synthesis of P-chiral bisphosphine ligands. **c, d** Applications of these ligands in asymmetric synthesis.

Data availability

Detailed experimental procedures, characterization data, NMR spectra of new compounds, HPLC spectra for chiral compounds, detailed computational results, and calculated structures are available within Supplementary Information. Cartesian coordinates of the calculated structures are available from Source Data, which are provided with this paper. All the data supporting the findings of this work are available within the article and its Supplementary Information files or from the corresponding author upon request. Source data are provided with this paper.

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Author contributions

J.M. conceived the idea and guided the project. W.L., J.Z., S.B., and G.L. performed the experiments and analyzed the data. Y.L. directed the part of computational study. L.Z. carried out the computational study. J.M. and S.C. wrote the manuscript. All authors approved the submission of the manuscript.

Competing interests

The authors declare no competing interests.

Additional information

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