

Field-effect passivation for minimized voltage loss in highly efficient antimony selenosulfide solar cells

Received: 8 August 2025

Accepted: 27 November 2025

Published online: 06 January 2026

 Check for updates

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The inferior heterojunction quality and misaligned energy levels at the buffer/absorber interface cause severe interface recombination and large open-circuit voltage (V_{OC}) loss, limiting the power conversion efficiency (PCE) of antimony selenosulfide ($Sb_2(S,Se)_3$) solar cells. Here, we develop a field-effect passivation strategy by introducing a low-work-function Ta_2O_5 dielectric layer between the CdS and $Sb_2(S,Se)_3$ layers. This Ta_2O_5 layer serves as an optimal substrate for growing highly crystalline $Sb_2(S,Se)_3$ films while also enhancing interfacial charge transport. The positive fixed charges in Ta_2O_5 strengthen the built-in electric field and promotes electrons extraction while suppressing holes accumulation at the interface, thereby substantially suppressing non-radiative recombination probabilities. Implementing this passivation strategy yields a record PCE of 10.95% (10.65% certified) an V_{OC} of 695 mV, corresponding to a remarkably low voltage deficit. This work establishes a universal physical passivation paradigm for interface quality optimization and V_{OC} loss mitigation in high-performance $Sb_2(S,Se)_3$ photovoltaics.

In response to the depletion of traditional energy sources, there has been a growing demand for clean and renewable energy, driving the development of new thin-film solar cells. Antimony selenosulfide ($Sb_2(S,Se)_3$) has recently demonstrated significant potential in emerging thin-film photovoltaics due to its high stability, low toxicity, tunable band gap and excellent optoelectronic properties^{1–4}. Based on these advantages, $Sb_2(S,Se)_3$ solar cells have undergone a rapid development, achieving a reported power conversion efficiency (PCE) of 10.92%⁵. However, the current efficiency of these thin-film solar cells is limited by significant open-circuit voltage (V_{OC}) losses caused by severe non-radiative recombination at the absorber layer and interfaces^{6,7}. Currently, high-efficiency $Sb_2(S, Se)_3$ solar cells are often

fabricated through direct deposition on cadmium sulfide (CdS) substrates. An unfavorable energy level arrangement at the CdS/ $Sb_2(S,Se)_3$ interface, such as a cliff-like band structure, can cause severe carrier recombination at the interface. Therefore, developing an efficient passivation strategy to improve the quality of the CdS/ $Sb_2(S,Se)_3$ heterojunction and mitigate the V_{OC} losses of devices is crucial.

Researchers have conducted extensive studies on interface engineering to improve the quality of heterojunctions and suppress interface carrier recombination. To transform the band structure at the CdS/ $Sb_2(S,Se)_3$ interface from a cliff-like to spike-like structure, Liu et al. employed a heterojunction lithiation strategy and achieved a high PCE of 10.76% for the $Sb_2(S,Se)_3$ solar cells⁸. As for optimizing the CdS

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buffer layer, a hydrazine hydrate solution was used to remove surface oxides and improve the surface morphology of CdS films, resulting in a PCE of 10.30%. Besides, Chen group investigated the effects of potassium chloride (KCl) and ethylenediamine (EDA) post-treatments on CdS films, which yielded PCEs of 9.98% and 10.10%, respectively^{10,11}. They found that KCl treatment effectively reduced the surface roughness of CdS films and improved their conductivity, while EDA treatment promoted the grain growth and crystallinity of CdS films and suppressed the formation of interface defects. Although these passivation approaches reduced the V_{OC} losses of the device to some extent, such chemically-based passivation methods suffer from byproduct residues, poor long-term stability and potentially harmful chemical reactions at interfaces. This hinders the reproducibility of device performance and large-scale applications¹².

In addition to chemical passivation technology, a physical passivation strategy, such as field-effect passivation by introducing a dielectric layer, is a promising approach. In crystalline silicon (Si) solar cells, the zinc oxide (ZnO), titanium oxide (TiO₂), strontium oxide (SrO_x) thin films have been deposited on the Si substrate to provide moderate surface passivation and form the additional interfacial fields, which facilitate the charge-selective transport and reduce the probability of interface recombination^{13–15}. In perovskite solar cells, a molybdenum oxide (MoO_x) thin film with a high work function is inserted between the perovskite and hole transport layer. This field-effect passivation of MoO_x effectively enhances the built-in field and improves interface charge transport thanks to the formation of interfacial polarization¹⁶. Furthermore, fixed charge passivation is used to modify carrier concentration distribution near the heterojunctions and selectively extract charge carriers^{17,18}. Despite these advantages, the application of the physical passivation strategy to inorganic thin-film solar cells remains underexplored, and the impact of field-effect passivation on the charge transport and recombination dynamics in Sb₂(S,Se)₃ solar cells has rarely been reported.

Herein, we fabricated an ultrathin, transparent Ta₂O₅ dielectric layer between CdS and Sb₂(S,Se)₃ layers to investigate the effect of field-effect passivation on the V_{OC} loss in the device. Ta₂O₅ is a wide-bandgap transition metal oxide that exhibits exceptional optical transparency and chemical stability^{19,20}. Its material properties remain unaffected during subsequent high-temperature hydrothermal and annealing processes. There are positive fixed charges generated in Ta₂O₅ films, which can attract electrons while concurrently repelling holes away from the interface²¹. Thanks to the reduced V_{OC} losses resulting from this field-effect passivation, the champion device achieves a certified PCE of 10.65% with V_{OC} of 692 mV, corresponding to 54.96% of the theoretical limit ($V_{OC}/V_{OC,SC}$), which is comparable to the highest reported value (55.14%) for Sb₂(S,Se)₃ solar cells⁵. This type of inorganic thin-film solar cells also exhibits good operational stability.

Results and discussion

Characterization of CdS/Ta₂O₅ buffer layers

In this study, a highly transparent and ultrathin Ta₂O₅ dielectric layer was fabricated via the sol-gel spin-coating technique. Fig. 1a depicts the specific reaction pathway and conditions for Ta₂O₅:Ta⁵⁺ ions in the solution form ethanol-tantalum complexes preferentially, followed by the generation of the desired Ta₂O₅ precipitate under the influence of high temperature and hydroxyl groups. The inset shows the transparent, homogeneous Ta₂O₅ precursor solution. To demonstrate the formation of Ta₂O₅, X-ray diffraction (XRD) was conducted on the thin-film samples to determine their composition and crystal structure. As shown in the XRD pattern, all diffraction peaks of the sample closely match the orthorhombic phase of Ta₂O₅ (JCPDS #25-0922), with a pronounced (001) grain orientation. The Ta₂O₅ dielectric layer has a high dielectric constant of ~28²², a wide band gap of 4.25 eV (Supplementary Fig. 1), and an ultralow electrical conductivity of ~2.11 × 10⁻⁵ S/cm

based on Hall effect measurements. Scanning electron microscopy (SEM) analysis of CdS films without and with Ta₂O₅ modification revealed distinct morphological characteristics. As shown in Fig. 1b, e, the pure CdS films exhibited a rougher surface morphology with numerous pinholes. Such low-quality seed layers can degrade the quality of subsequent Sb₂(S,Se)₃ films and the resultant device performance. Remarkably, the deposition of Ta₂O₅ effectively filled surface pinholes and cracks in the CdS films, resulting in a smoother and more uniform surface topography. To verify the uniformity of the ultrathin Ta₂O₅ layer, we performed EDS mapping analysis on the CdS/Ta₂O₅ substrate. As shown in Supplementary Fig. 2, the Ta signal (representing Ta₂O₅) exhibits a continuous and homogeneous distribution on the surface of CdS films, with no observable island-like aggregation or large-scale uncovered areas. This indicates a uniform and complete coverage of the CdS film by the deposited Ta₂O₅ layer. Simultaneously, the signals of Cd and S also display a homogeneous distribution, further confirming that the Ta₂O₅ layer is sufficiently ultra-thin, thereby allowing the underlying CdS signals to remain clearly detectable. The CdS and CdS/Ta₂O₅ films exhibited similar transmittance across the entire visible spectrum (Supplementary Fig. 3). We also used atomic force microscopy (AFM) to analyse the surface morphology of CdS films (Fig. 1c, f). Following the deposition of Ta₂O₅, the roughness (R_q) of the CdS films was decreased from 27.8 nm to 23.1 nm, consistent with the SEM results. We also conducted the contact angle tests on CdS films (Fig. 1d). The contact angle was reduced from 61.9° to 44.0° after the Ta₂O₅ layer was introduced, positively impacting the heterogeneous nucleation and grain growth of the Sb₂(S,Se)₃ films. To confirm that the Ta₂O₅ layer facilitates carrier transport, we measured the conductivity of CdS films, and the results are shown in Fig. 1g (the inset shows the structure of device). After depositing the Ta₂O₅ dielectric layer, the conductivity of the CdS films was increased from 3.74 × 10⁻⁵ S/cm to 6.22 × 10⁻⁵ S/cm.

Morphological analysis of Sb₂(S,Se)₃ layer

According to previous studies, the polarity of CdS surfaces has a significant influence on the growth orientation of the absorber layer¹. As shown in Supplementary Fig. 4, the Sb₂(S,Se)₃ films tend to form a favorable (hk1) grain orientation on the non-polar (100) plane of CdS films. In contrast, exposure of the polar (101) and (10 $\bar{1}$) planes leads to complex anisotropy in the grain orientation of Sb₂(S,Se)₃, negatively affecting the carrier transport. To demonstrate that the Ta₂O₅ layer reduces the exposure of polar planes in CdS, we performed XRD measurements on CdS films. As shown in the Supplementary Fig. 5, compared to the pure CdS film, the CdS/Ta₂O₅ sample exhibits significantly reduced peak intensities of polar planes, such as (101), and enhanced peak intensities of nonpolar planes, such as (100)²³. To investigate the effect of Ta₂O₅ dielectric layer on the quality of Sb₂(S,Se)₃ films, we implemented XRD and SEM measurements. The diffraction patterns of Sb₂(S,Se)₃ films matched well with the orthorhombic Sb₂S₃ (JCPDS#42-1393) and Sb₂Se₃ (JCPDS#15-0861) phases (Supplementary Fig. 6a). We calculated the texture coefficients (TC) of the crystal planes for Sb₂(S,Se)₃ films deposited on CdS and CdS/Ta₂O₅ substrates based on the XRD data, with detailed results presented in Supplementary Fig. 6b²⁴. The Sb₂(S,Se)₃ films deposited on the CdS/Ta₂O₅ substrate exhibit significantly higher TC values for (hk1) planes, such as (211), (221), (301) and (421) compared to those of the control Sb₂(S,Se)₃ films. This provides evidence for the optimized grain orientation of Sb₂(S,Se)₃ films induced by the Ta₂O₅ layer. As shown in Supplementary Fig. 7, the local XRD analysis revealed that Sb₂(S,Se)₃ films deposited on Ta₂O₅ substrates exhibited stronger (hk1) diffraction peaks and narrower full width at half maximum than the control Sb₂(S,Se)₃ films. Regarding the slight shift in diffraction peaks, we attribute it to the changes in the chemical composition of the absorber layer (Supplementary Fig. 8 and Supplementary Table 1).

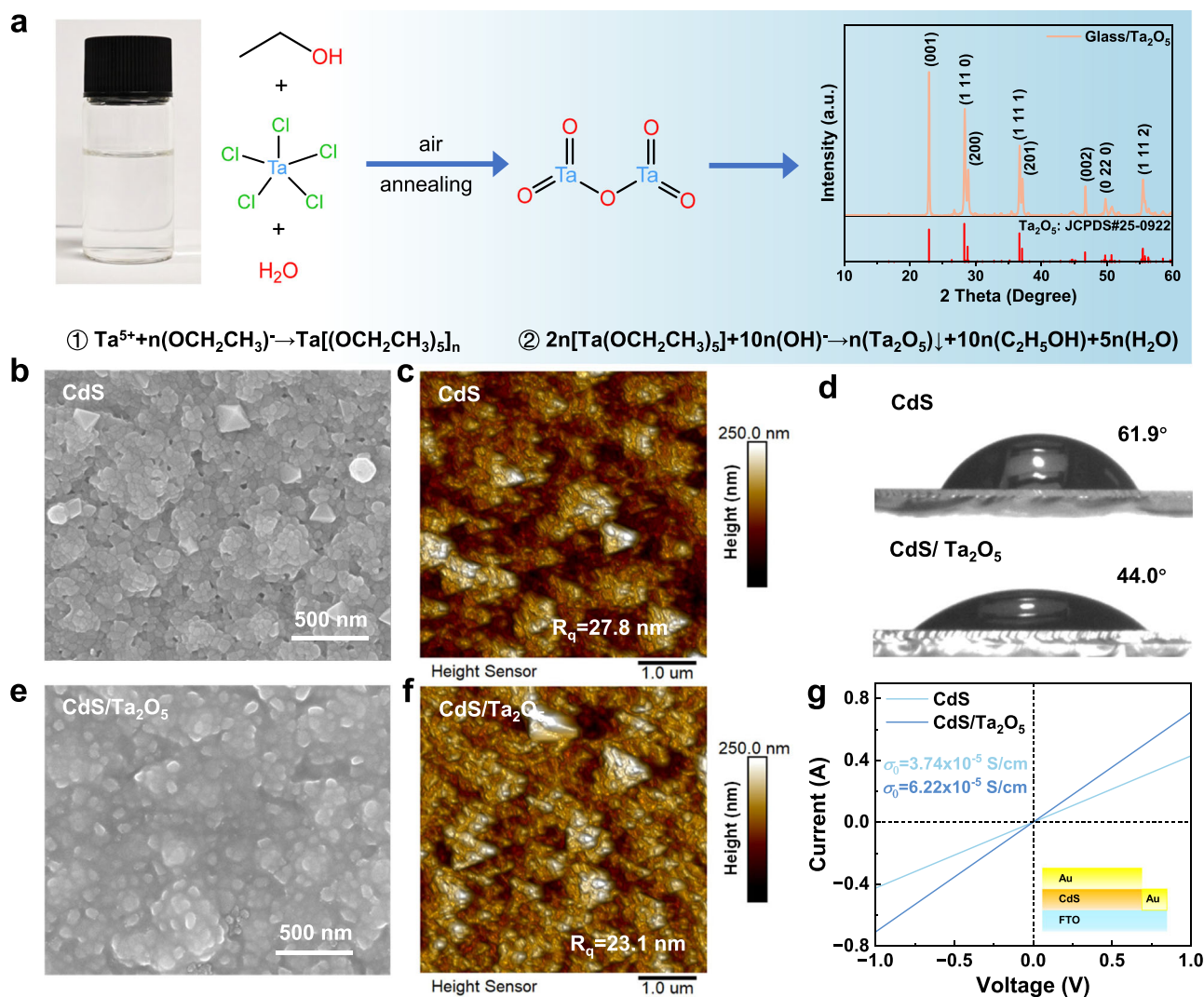


Fig. 1 | Characterization of CdS/Ta₂O₅ buffer layers. **a** Reaction schematic and XRD pattern of Ta₂O₅ (The inset shows the precursor solution of Ta₂O₅). **b, e** Scanning electron microscopy (SEM) images, **c, f** atomic force microscopy

(AFM) mappings, and **d** hydrophilicity measurements of CdS and CdS/Ta₂O₅ films. **g** *I*-*V* characteristics of FTO/CdS/Au and FTO/CdS/Ta₂O₅/Au devices for conductivity evaluation.

The introduction of Ta₂O₅ substrate modifies the crystallization kinetics, leading to an altered S/Se ratio in Sb₂(S,Se)₃²⁵. SEM images reflected that both the control and Ta₂O₅-modified Sb₂(S,Se)₃ films had a compact morphology. However, the control films displayed a higher density of large granular precipitates, and the phenomenon was effectively mitigated in the Ta₂O₅-modified Sb₂(S,Se)₃ films. Supplementary Fig. 9 showcased the AFM images of the Sb₂(S,Se)₃ films, indicating that the R_q was reduced for films deposited on Ta₂O₅ substrate (39.6 nm) compared to that of the control films (50.3 nm). We attributed this to the suppression of large granular precipitates in the absorber layer. The reduced surface roughness facilitates uniform interfacial contact and minimizes localized recombination, which is critical for enhancing charge transport and device performance. These results suggest that the insertion of Ta₂O₅ layer improves the crystallinity, grain orientation and morphology of Sb₂(S,Se)₃ films. Therefore, the Sb₂(S,Se)₃ films deposited on Ta₂O₅ substrate exhibit an enhanced light absorption due to the improved film quality (Supplementary Fig. 10). Since interfacial stability is fundamental to device reproducibility and long-term reliability, we performed HAADF-STEM imaging and corresponding EDS mapping analysis for a cross-sectional sample with structure of FTO/CdS/Ta₂O₅/Sb₂(S,Se)₃. As shown in Supplementary Fig. 11, the

HAADF-STEM image reveals sharp and distinct interfaces between all functional layers. Notably, a continuous thin layer with a thickness of ~3 nm, corresponding to Ta₂O₅, is clearly observed between the CdS and Sb₂(S,Se)₃ layers. EDS mapping further demonstrates a sharp and confined Ta signal precisely at the CdS/Sb₂(S,Se)₃ interface, while other elements such as Cd, S, Sb, and Se exhibit negligible interdiffusion. These results confirm that the Ta₂O₅ dielectric layer maintains robust chemical and structural stability during high-temperature hydrothermal and annealing processes, forming well-defined heterointerfaces with adjacent layers.

Fixed charge passivation

In crystalline silicon solar cells, Ta₂O₅ has been widely recognized as an efficient passivation layer, which is capable of introducing fixed charges at the silicon surface. Theoretically, when Ta₂O₅ contacts with Sb₂(S,Se)₃ layer, it may exhibit a similar fixed charge effect. To validate our analyses, we measured the fixed charges by preparing the metal oxide semiconductor (MOS) structured devices for *C*-*V* measurements (Fig. 2a). By differentiating the *C*-*V* data, the voltage corresponding to the extreme value represents the flat-band voltage. As shown in Fig. 2b, the flat-band voltages of the CdS and CdS/Ta₂O₅ samples are -0.35 V and -0.50 V, respectively. We therefore confirm that the Ta₂O₅ enables

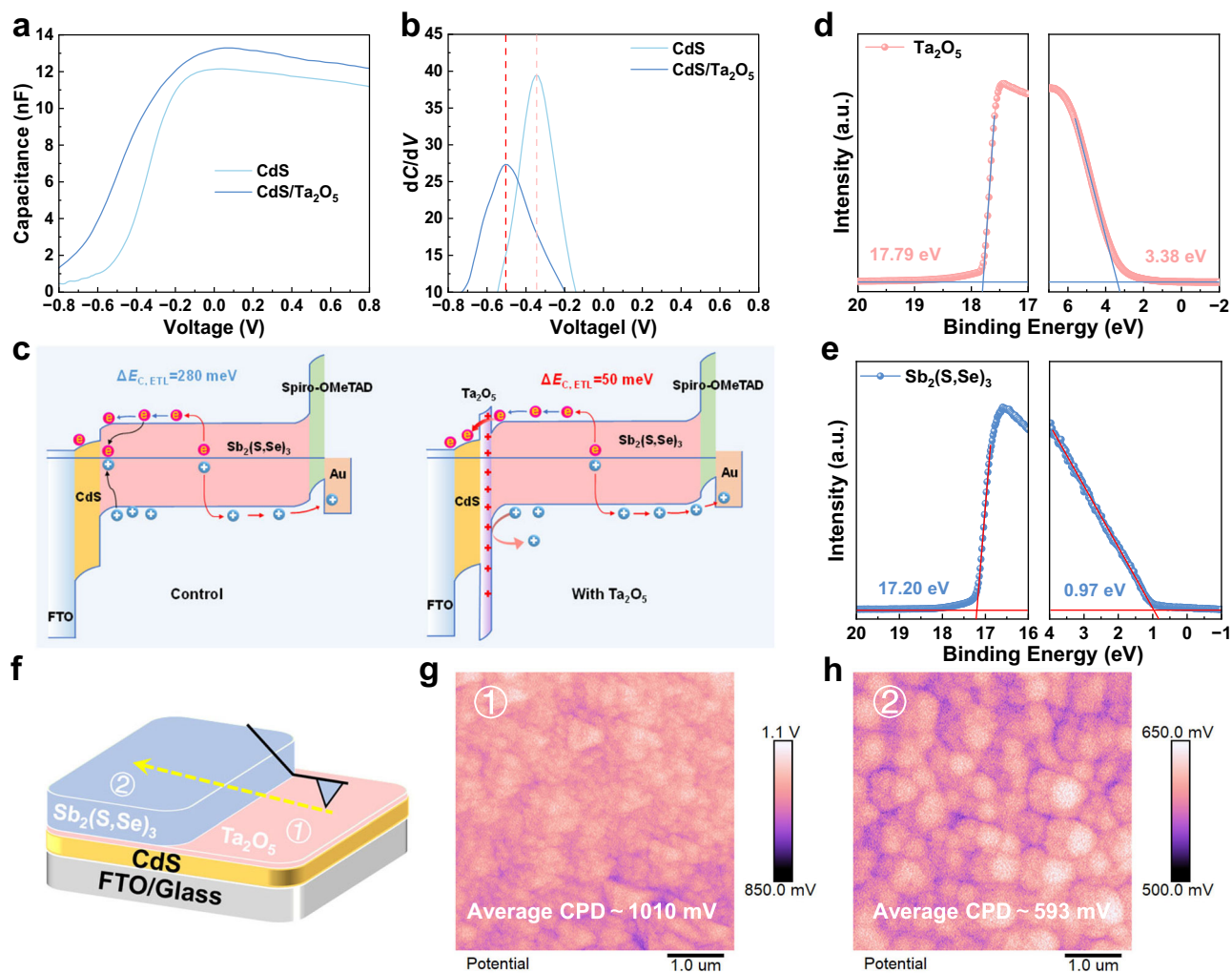


Fig. 2 | Fixed charge passivation. **a** C - V curves and **b** flat band voltage of CdS and CdS/Ta₂O₅ devices. The C - V measurements were performed on devices with the structures FTO/CdS/Sb₂(S,Se)₃/Au and FTO/CdS/Ta₂O₅/Sb₂(S,Se)₃/Au. **c** Schematic band diagrams of the control and Ta₂O₅ incorporated devices. UPS spectra of secondary electron cutoff and valence band of **(d)** Ta₂O₅ and **e** Sb₂(S,Se)₃ films. **f** Schematic diagram of the KPFM measurements. **g, h** The corresponding KPFM images of regions 1 and 2.

positive fixed charges at the interface, consistent with reported literatures^{26,27}. The fixed charge (Q_f) density was calculated using the following formula²⁸:

$$Q_f = \frac{C_{in}(V_{MS} - V_{fb})}{A \times q} \quad (1)$$

Here, C_{in} denotes the intrinsic capacitance of the MOS structure, V_{MS} represents the metal-semiconductor potential difference determined by the work function difference between FTO (4.8 eV) and Sb₂(S,Se)₃, V_{fb} is the flat-band voltage, A is the electrode area, and q is the elementary charge. The calculated fixed charge density Q_f of the Ta₂O₅ layer is $+2.24 \times 10^{11} \text{ cm}^{-2}$ (Supplementary Table 2). When positive fixed charges of Ta₂O₅ appear at the interface, more electrons of Sb₂(S,Se)₃ gather therein, resulting in a downward band bending of Sb₂(S,Se)₃ (Fig. 2c). These positive fixed charges enable field-effect passivation at the Ta₂O₅/Sb₂(S,Se)₃ interface, facilitating efficient electron extraction and suppressing hole accumulation^{17,18}. To theoretically verify the field-effect passivation mechanism of the Ta₂O₅ layer, we performed the device simulations by using the SCAPS-1D software^{29,30}. A device model is established that accurately corresponds to the actual structure of solar cells. All the essential materials

parameters are shown in Supplementary Table 3. The simulation results (Supplementary Fig. 12) demonstrate that the insertion of the Ta₂O₅ layer significantly reduces the conduction band offset ($\Delta E_{C,ETL}$) between CdS and Sb₂(S,Se)₃ from 280 to 50 meV. The Ta₂O₅ layer provides a more favorable path for efficient transport of photo-generated electrons from the absorber layer to the electron transport layer.

In addition to the fixed positive charges, Fermi level alignment between Ta₂O₅/Sb₂(S,Se)₃ also results in band bending with the same trend. We performed the ultraviolet photoelectron spectroscopy (UPS) for the Ta₂O₅ and Sb₂(S,Se)₃ films, respectively. Fig. 2d, e present that the secondary electron cut-off edges are 17.79 and 17.20 eV, and the distances between Fermi energy level and valence band maximum are 3.38 and 0.97 eV for the Ta₂O₅ and Sb₂(S,Se)₃, respectively. The corresponding work functions are 3.43 eV and 4.02 eV, similar to the values reported in the literature^{11,18}. As shown in Fig. 2c, the band bending of Sb₂(S,Se)₃ is downward while the band bending of Ta₂O₅ is upward when the Fermi-level alignment is achieved between Ta₂O₅/Sb₂(S,Se)₃. Kelvin probe force microscopy (KPFM) was conducted to probe the contact potential difference (CPD) across the Ta₂O₅/Sb₂(S,Se)₃ heterojunction (Fig. 2f). A significant reduction in CPD was observed as the probe traversed from the Ta₂O₅ region to the Sb₂(S,Se)₃ region (Fig. 2g, h). The Ta₂O₅ region exhibited an average

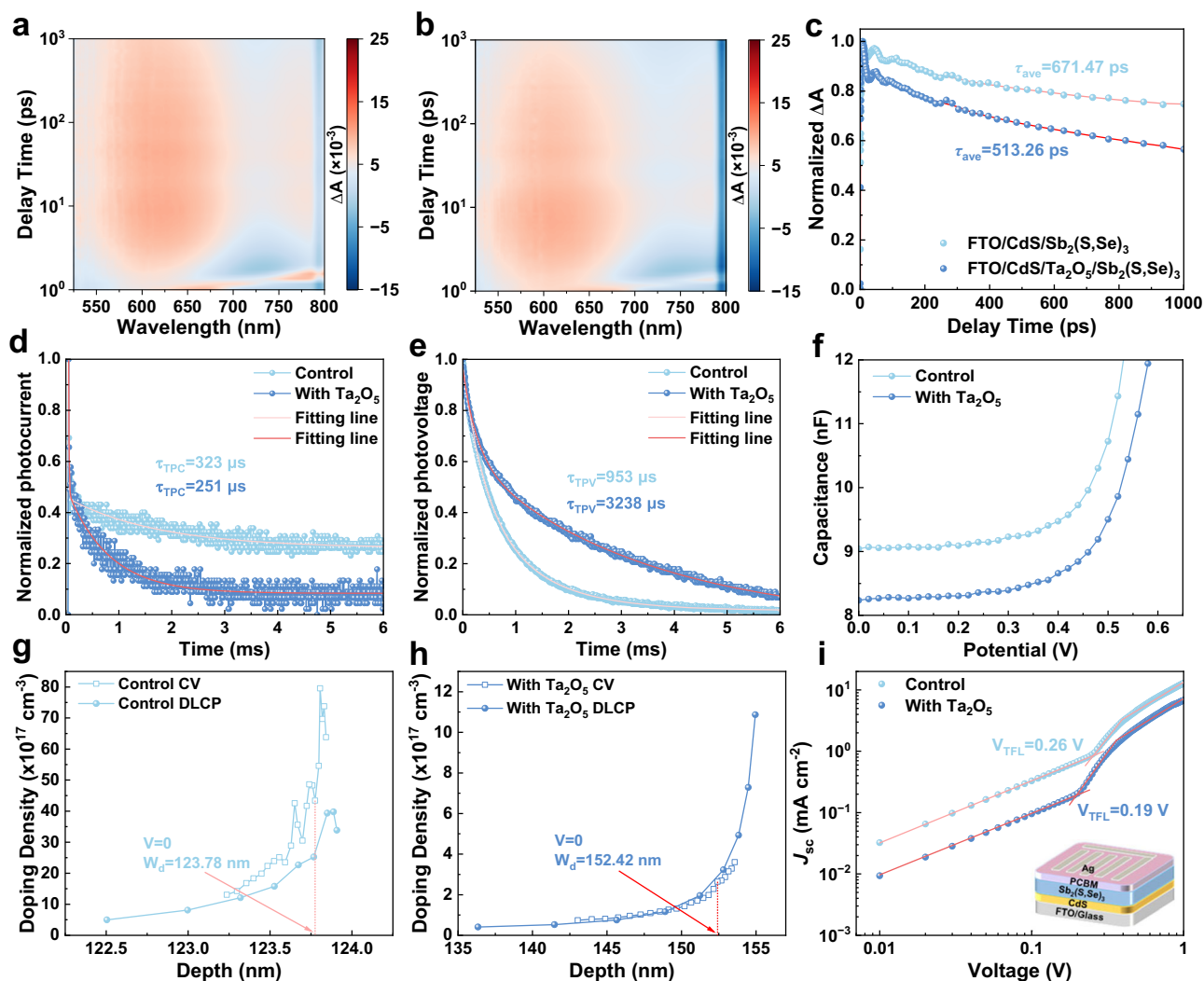


Fig. 3 | Carrier dynamics and interfacial defect analysis. Two-dimensional pseudo-color images of TAS spectra of the (a) FTO/CdS/Sb₂(S,Se)₃ and (b) FTO/CdS/Ta₂O₅/Sb₂(S,Se)₃ samples. **c** Transient decay kinetics of the Sb₂(S,Se)₃ films deposited on CdS and CdS/Ta₂O₅ substrate. **d, e** Transient photocurrent/photovoltage (TPV/TPC) curves, **f** capacitance–voltage (C–V) curves, **g, h** C–V and DLCP profiles, and **i** space charge limiting current (SCLC) curves for the control and Ta₂O₅ incorporated devices.

CPD of approximately 1010 mV, while the Sb₂(S,Se)₃ region exhibited a decreased CPD of approximately 593 mV. This difference in CPD is consistent with the UPS analysis. With the insertion of Ta₂O₅ layer, the presence of fixed positive charges and the significant work function difference induce more pronounced downward band bending of Sb₂(S,Se)₃.

Carrier dynamics and interfacial defect study

To further verify that the Ta₂O₅ dielectric layer facilitated carrier transport, we used transient absorption spectroscopy (TAS) to analyze the carrier transport dynamics. Fig. 3a, b showcased the two-dimensional pseudo-color maps of the TAS spectra for Sb₂(S,Se)₃ films. Broad photo-induced absorption (PIA, $\Delta A > 0$) in the range of 525–700 nm appeared in both samples due to the formation of S⁻ radical and the trapping of holes within Sb₂(S,Se)₃ films^{31,32}. Compared to the control samples, the Sb₂(S,Se)₃ films deposited on Ta₂O₅ dielectric layer exhibited lower PIA signals. Supplementary Fig. 13 displayed the time-resolved absorption spectra at different times, in which the Ta₂O₅ incorporated Sb₂(S,Se)₃ films showed a weaker peak intensity than the control films at the same delay time. In general, the TAS intensity reflects the number of carriers in the conduction and valence bands of the absorber, so the decreased intensity can be

attributed to the effective charge transport via the Ta₂O₅ dielectric layer³³. Fig. 3c showed the decay kinetics of the photobleaching peak intensity at 622 nm for the samples, and the carrier decay lifetimes were obtained by fitting the double-exponential equations in Supplementary Note 1. The corresponding results were listed in Supplementary Table 4. The average carrier decay lifetimes of the Sb₂(S,Se)₃ films on the Ta₂O₅ layer are much lower than those of the control films, suggesting that Ta₂O₅ layer enables the fast electron extraction and transportation to cathode. To explore the carrier recombination behavior, we performed the transient photovoltage/photocurrent (TPV/TPC) measurements on the devices without and with Ta₂O₅ dielectric layers. As shown in Fig. 3d and Supplementary Table 5, the TPC results indicated a reduced photocurrent decay lifetime of 252 μ s in the Ta₂O₅ incorporated devices compared to that of the control devices (328 μ s), consistent with the analysis of TAS measurements. Fig. 3e presented the TPV decay curves of the devices, in which the Ta₂O₅ incorporated devices exhibited a longer carrier recombination lifetime of 3238 μ s than that of the control devices (953 μ s) (Supplementary Table 5). We attributed this to the suppressive carrier recombination rate by the field-effect passivation of Ta₂O₅. Additionally, the capacitance–voltage (C–V) measurements showed a decreased capacitance in the Ta₂O₅ incorporated devices (Fig. 3f). This suggests

Table 1 | Summary of the certified photovoltaic parameters of the Sb₂(S,Se)₃ solar cells

Ref.	E_g (eV)	V_{OC} (V)	FF (%)	PCE (%)	$V_{OC}/V_{OC, SQ}$ (%)	$V_{OC} \times FF$ (%)
Nat. Photon. (2015) ³	1.1	0.4	55.7	5.6	46.61	22.28
Nat. Energy (2017) ⁴	1.1	0.391	57.8	5.93	45.56	22.6
Nat. Energy (2025) ⁵	1.3	0.576	68.96	10.7	55.14	39.72
Energy Environ. Sci. (2024) ⁸	1.53	0.672	65.9	10.5	53.38	44.28
Angew. Chem. Int. Ed. (2025) ²⁴	1.19	0.47	65.36	9.17	49.89	30.72
ACS Energy Lett. (2017) ²⁹	1.15	0.427	59.3	6.5	47.19	25.32
Nat. Energy (2020) ³¹	1.49	0.655	63.5	10	53.61	41.59
Adv. Energy Mater. (2025) ³²	1.73	0.692	64.45	8.08	47.88	44.6
Energy Environ. Sci. (2024) ³⁸	1.25	0.477	69.63	10.18	47.8	33.21
Nat. Commun. (2018) ³⁹	1.19	0.42	60.4	7.6	44.58	25.37
Chem. Eng. J. (2022) ⁴⁵	1.49	0.61	67.3	9.01	49.93	41.05
Nat. Commun. (2019) ⁴⁶	1.24	0.4	70.3	9.2	40.46	28.12
Sol. RRL (2017) ⁴⁷	1.65	0.56	52.4	5.71	40.85	29.34
Adv. Electron. Mater. (2019) ⁴⁸	1.65	0.744	55.19	6.06	54.27	41.06
Adv. Energy Mater. (2022) ⁴⁹	1.71	0.703	61.46	7.15	49.27	43.21
This work	1.53	0.692	68.70	10.65	54.96	47.54

incorporated devices measured under AM 1.5 G illumination (100 mW cm⁻²). The control devices achieved a PCE of 9.21% with a V_{OC} of 0.647 V, a short circuit current density (J_{SC}) of 22.53 mA cm⁻² and a fill factor (FF) of 63.19%. Specifically, the devices with optimal thickness of Ta₂O₅ reached a PCE of 10.95% with a V_{OC} of 0.695 V, a J_{SC} of 23.22 mA cm⁻² and a FF of 67.87%, which is the highest efficiency for the Sb-based thin-film solar cells so far. Besides, the J - V curves simulated by SCAPS-1D show significant improvements in V_{OC} and FF (Supplementary Fig. 16 and Supplementary Table 8), which are fully consistent with our experimental results. Moreover, the Ta₂O₅ incorporated devices exhibited a good reproducibility (Supplementary Fig. 17). Fig. 4b showed the external quantum efficiency (EQE) of the control and Ta₂O₅ incorporated devices, and the integrated current densities are in good agreement with J_{SC} obtained from the J - V curves. The response in the spectral range of 330–500 nm is enhanced due to the optimization of the CdS buffer layer and the heterojunction interface, while the response in the spectral range of 500–750 nm is increased because of the improved quality of Sb₂(S,Se)₃ absorber layer³⁶. We further calculated the Urbach energy of the devices based on the EQE spectra. As shown in Supplementary Fig. 18, the Ta₂O₅ incorporated devices displayed a reduced Urbach energy of 28.43 meV compared to 31.47 meV for the control devices. The lower Urbach energy indicates that the Sb₂(S,Se)₃ films deposited on Ta₂O₅ layers have a higher crystallinity and less bulk defects^{37,38}.

To investigate the impact of the Ta₂O₅ dielectric layer on the FF of devices, we decomposed the FF losses into non-radiative recombination losses and transport losses through Supplementary Note 4 (Fig. 4c and Supplementary Table 9). The theoretical maximum fill factor (FF_{max}) values for the control and Ta₂O₅ incorporated devices were 74.2% and 76.64%, respectively. Introducing the Ta₂O₅ dielectric layer reduced FF losses caused by non-radiative recombination from 19.90% to 13.16%. The optical band gaps calculated from EQE measurements show no variation between the control and Ta₂O₅ incorporated devices (Supplementary Fig. 19). This indicates that the improvement of V_{OC} is due to the field-effect passivation of Ta₂O₅ and the increased built-in electric field of devices. As shown in the capacitance–voltage (C^2 - V) curves, the extracted built-in voltage (V_{bi}) of the control devices is 649 mV, which is increased to 703 mV for the Ta₂O₅ incorporated devices (Fig. 4d). Furthermore, the optimized devices were submitted to an authoritative, independent laboratory (South China National Center of Metrology) for performance certification (certification report provided in Supplementary Fig. 20). Consequently, the Ta₂O₅ incorporated devices demonstrated a certified PCE of 10.65% (10.57%)

with a V_{OC} of 0.692 (0.691) V, a FF of 68.70% (68.17%) and a J_{SC} of 22.42 (22.45) mA cm⁻² measured in reverse (forward) scan. It is worth noting that the certified PCE of 10.65% is second only to the certified record efficiency of 10.70% (Fig. 4e and Table 1). As shown in Fig. 4f (calculation method detailed in Supplementary Note 5), we calculated the product of $V_{OC, SQ} \times FF_{SQ}$ according to the Shockley-Queisser limit values across different band gaps, and compared the $V_{OC} \times FF$ values of the reported certified devices. The product of $V_{OC} \times FF$ in this work reached 47.54%, which was the highest value among all the reported certified devices (Table 1). Concurrently, the $V_{OC}/V_{OC, SQ}$ ratio of the certified device attained 54.96%, indicating a substantially reduced V_{OC} deficit ($V_{OC, def}$) (Supplementary Fig. 21 and Table 1). These results fully demonstrated the effectiveness of the field-effect passivation strategy in enhancing device performance.

To further demonstrate that carrier recombination was suppressed, we implemented the dark J - V characterization, light-intensity-dependent J - V measurements, and electrical impedance spectroscopy (EIS) under both dark and different light intensity conditions. It can be seen in Supplementary Fig. 22a that the reverse leakage currents were significantly reduced upon the incorporation of Ta₂O₅ layer. By processing and fitting the dark J - V curves based on the equations in Supplementary Note 6, we can obtain relevant device parameters, including shunt conductance (G), series resistance (R_s), diode ideality factor (A) and reverse saturation current density (J_0). As shown in Supplementary Fig. 22b-d, the Ta₂O₅ incorporated devices exhibit a G value of 0.0005 mS cm⁻², an R_s of 2.55 Ω cm², a diode ideality factor of 1.78 and a J_0 of 6.28×10^{-6} mA cm⁻². All these parameters show improvement over those of the control device, which had a G value of 0.006 mS cm⁻², an R_s of 6.43 Ω cm², a diode ideality factor of 1.92 and a J_0 of 2.48×10^{-5} mA cm⁻². We then measured the V_{OC} versus light-intensity curves that followed the relationship of $V_{OC}/n(k_B T/q)$ in (I), where the k_B , T , and q represent Boltzmann's constant, the temperature in Kelvin, and the elementary charge, respectively. In the case of ideal trap-free recombination, the slope of the curves corresponds to $k_B T/q$. Any deviation of the measured slope from $k_B T/q$ indicates the extent of trap-assisted recombination under open-circuit conditions³⁹. As shown in Fig. 4g, the Ta₂O₅ incorporated devices exhibit a lower slope of 1.219 $k_B T/q$ than that of the control devices (1.323 $k_B T/q$), indicating that the Ta₂O₅ layer suppresses trap-assisted recombination at the CdS/Sb₂(S,Se)₃ interfaces through the effective field-effect passivation. Furthermore, the light-intensity dependence of J_{SC} (Supplementary Fig. 23) also demonstrates that the introduction of the Ta₂O₅ dielectric layer effectively suppresses carrier recombination in the

devices. We performed EIS on the devices without and with Ta₂O₅ dielectric layer under dark conditions (Supplementary Fig. 24). The starting point in the high-frequency region represents the series resistance R_s , and the semicircle arc in the low-frequency region denotes the recombination resistance R_{rec} . The extracted R_s of solar cells decreased from 19.79 to 3.35 Ω and the R_{rec} increased from 1358.85 to 4676.42 Ω with the introduction of Ta₂O₅ dielectric layer. To investigate the interfacial charge transfer properties of the devices, we performed EIS measurements under different light intensities. As shown in Supplementary Fig. 25a, b, all the Nyquist plots exhibit a single semicircular arc, indicating that the interfacial process is dominated by a single time constant⁴⁰. The equivalent circuit model was used to fit the impedance data (Supplementary Fig. 25c), where R_s represents the series resistance, R_{ct} corresponds to the charge transfer resistance, and the constant phase element denotes the geometric capacitive behavior of the device⁴¹. The R_{ct} value of both devices decreases linearly with the increasing light intensity (Supplementary Fig. 25d), which can be attributed to that the increased photo-generated carrier concentration effectively reduces the energy barrier of charge transfer. Notably, the R_{ct} values of the Ta₂O₅ incorporated device were significantly lower than those of the control device under different light intensities. This demonstrates that the fixed charges in the Ta₂O₅ dielectric layer effectively reduce the charge transfer barrier through the field-effect passivation effect, thereby promoting efficient extraction and transport of charge carriers^{42–44}. Clearly, the Ta₂O₅ incorporated devices exhibit enhanced electrical properties and suppressed carrier recombination in comparison to the control devices, accounting for the improvement of V_{OC} and FF⁴⁵. Finally, we performed the stability measurement of an unencapsulated target device that maintained ~96% of its initial PCE after continuous tracking at maximum power point for 200 h, demonstrating good operational stability (Fig. 4h).

Discussion

In summary, we have demonstrated a physical passivation strategy by introducing a low work function Ta₂O₅ dielectric layer between the CdS and Sb₂(S,Se)₃ layers, to promote charge transfer and suppress carrier recombination at interfaces. The introduction of Ta₂O₅ not only realizes a twofold optimization of the CdS buffer layer and the Sb₂(S,Se)₃ absorber layer, but also yields a field-effect passivation at the CdS/Sb₂(S,Se)₃ heterojunction. This strategy improves the built-in potential of the devices and provides additional driving force for extracting electron-carrier to the cathode and mitigating hole-carrier accumulation at the interface, which effectively reduces the non-radiative recombination losses and the V_{OC} deficit. By this strategy, the optimized Sb₂(S,Se)₃ solar cell achieved a PCE of 10.95% with a high V_{OC} of 695 mV and a FF of 67.87%, which also reached a certified efficiency of 10.65% with a V_{OC} of 692 mV and a FF of 68.70%. This study reveals that field-effect passivation induced by fixed charges effectively improves carrier transport dynamics and heterojunction quality, which can be extended to other electronic devices.

Methods

Materials

All chemicals in this work were used as received without further purification unless otherwise specified. Cadmium nitrate tetrahydrate (Cd(NO₃)₂·4H₂O, AR), ammonium hydroxide (NH₄OH, 25–28%), thiourea (CH₄N₂S, GR), Cadmium chloride hemipentahydrate (CdCl₂·2.5H₂O, AR), antimony potassium tartrate (C₄H₄KO₇Sb·0.5H₂O, CP), and sodium thiosulfate pentahydrate (Na₂S₂O₃·5H₂O, AR) were purchased from Sinopharm Chemical Reagent Co., Ltd. Tantalum(V) chloride (TaCl₅, 99.99%) was obtained from Macklin. Selenourea (CH₄N₂Se, 98%) and acetonitrile (C₂H₃N, 99.8%) were sourced from Sigma-Aldrich Trading Co., Ltd. Lithium bis(trifluoromethanesulfonyl) imide (Li-TFSI, 99%) and Spiro-OMeTAD (99.9%) were procured from

Yolight Solar Technology Co., Ltd. 4-tert-Butylpyridine (tBP, 96%) and ethanol (C₂H₆O, 99.8%) were acquired from Aladdin Biochemical Technology Co., Ltd.

Devices fabrications

Fabrication of Ta₂O₅ dielectric layer. The fabrication process initiated by dissolving 10 mg tantalum chloride (TaCl₅) in 1.8 mL ethanol with 30 min ultrasonication. Subsequently, 1.12 mL ethanol and 0.08 mL ultrapure water were added to the solution, followed by another 30 min ultrasonication. The homogenized mixture was filtered through a 200 nm membrane filter to remove large aggregates, after which 1 mL ethanol was introduced and magnetically stirred for 1 h to obtain the Ta₂O₅ precursor solution. This precursor (50 μ L) was then spin-coated onto CdS substrates at varying spin speeds (4000–6000 rpm) and thermally annealed at 400–600 °C for 5 min under ambient air conditions, ultimately yielding densely compacted Ta₂O₅ films. The thicknesses of the Ta₂O₅ films prepared by different spin speeds were determined by spectroscopic ellipsometry to be 1 \pm 0.5, 3 \pm 0.5 and 5 \pm 0.5 nm, respectively.

Fabrication of Sb₂(S,Se)₃ solar cells: The cleaned FTO substrates were first treated with UV-ozone for 20 min, followed by CdS film deposition via chemical bath deposition at 66 °C for 16.5 min. The CdS precursor solution was prepared by thoroughly mixing 20 mL cadmium nitrate tetrahydrate (Cd(NO₃)₂·4H₂O), 26 mL ammonium hydroxide (NH₄OH, 25–28%), 14 mL thiourea (CH₄N₂S), and 140 mL ultrapure water. A cadmium chloride methanol solution (20 mg/mL) was then spin-coated onto the CdS film at 3000 rpm for 30 s and annealed in air at 400 °C for 10 min to complete the CdS buffer layer. Subsequently, Ta₂O₅ dielectric layers with varying thicknesses were fabricated on the CdS surface. For the deposition of Sb₂(S,Se)₃ absorber layer, antimony potassium tartrate (C₄H₄KO₇Sb·0.5H₂O), sodium thiosulfate pentahydrate (Na₂S₂O₃·5H₂O), and selenourea (CH₄N₂Se) served as Sb, S, and Se sources. Specifically, 0.260 g antimony potassium tartrate was dissolved in 39 mL ultrapure water, followed by sequential addition of 0.774 g sodium thiosulfate and 19.2 mg selenourea to form the precursor solution. The solution was transferred into a Teflon-lined autoclave containing FTO/CdS or FTO/CdS/Ta₂O₅ substrates and hydrothermally deposited at 135 °C for 2 h. Post annealing was conducted at 400 °C for 10 min under a nitrogen atmosphere to enhance the crystallinity of Sb₂(S,Se)₃. Spiro-OMeTAD hole transport layers were fabricated by thoroughly mixing 36.6 mg Spiro-OMeTAD, 29 μ L 4-tert-butylpyridine (tBP), 9.5 μ L lithium bis(trifluoromethanesulfonyl)imide (Li-TFSI, 520 mg/mL in acetonitrile), and 1 mL chlorobenzene, followed by spin-coating at 2000 rpm for 30 s under ambient conditions and subsequent annealing at 100 °C for 10 min. Device fabrication was finalized through thermal evaporation of an approximately 80 nm Au back electrode.

Characterization

Scanning electron microscopy (SEM, TESCAN MIRA LMS) was used to analyze the surface morphology of the CdS and Sb₂(S,Se)₃ films. The surface roughness and CPD were measured using a Bruker Multi-mode 8 equipped with an AFM and a KPFM. The absorption and transmission spectra were measured by a spectrophotometer (UV-2600i). The crystal structure of the Sb₂(S,Se)₃ films was measured by an X-ray diffractometer (XRD, Bruker D8 Discover) with a radiation source of Cu α . The energy level structures of Ta₂O₅ and Sb₂(S,Se)₃ films were measured by UV photoelectron spectroscopy (UPS, Thermo Fisher Scientific ESCALAB Xi) at –5 V bias. Core energy level variations of different elements in Ta₂O₅ and Sb₂(S,Se)₃ films were analyzed using an X-ray photoelectron spectrometer (XPS, Thermo Scientific Nexsa). Femtosecond TAS of Sb₂(S,Se)₃ films was tested using a pumped probe system (Helios, Ultrafast System LLC). TPC, TPV, EIS, and C^{-2} - V curves were measured using a Zahner electrochemical workstation. $C-V$ and DLCP measurements were conducted

using a Keithley 4200 semiconductor characterization system on device structures of Glass/FTO/CdS/without or with Ta₂O₅/Sb₂(S,Se)₃/Spiro-OMeTAD/Au. The *C*-*V* measurements were performed in the dark at room temperature with an alternating current (AC) amplitude of 30 mV, a frequency of 100 kHz and a direct current (DC) bias voltage range of 1 to 0.3 V. DLCP measurements were performed with an AC amplitude range of 15 to 140 mV and a DC bias voltage range of -0.2 to 0.2 V. The *J*-*V* curves of the devices were measured by combining a standard xenon sunlight simulator (LSS-55) with a test source meter (Keithley 2450) at a simulated AM 1.5 G (100 mW·cm⁻²) light intensity. The *J*-*V* curves were measured from -0.02 V to 0.70 V with a step size of 5 mV and no wait time. The EQE of the devices was measured using a quantum efficiency test system (LST-QE).

Reporting summary

Further information on research design is available in the Nature Portfolio Reporting Summary linked to this article.

Data availability

All data generated or analysed during this study are included in the published article and its Supplementary Information. Additional data are available from the corresponding authors on request. Source data are provided with this paper.

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Acknowledgements

This work is financially supported by the National Natural Science Foundation of China (52402104, C.L.), the Natural Science Foundation of Guangxi Province (2024GXNSFBA010280, C.L.), the National Key R&D

Program of China (grant no. 2024YFB4205104, Y.M.) and the Guangxi Science and Technology Major Project (AA23073018, B.Z.).

Author contributions

A.G. and C.L. conceived the study and designed the experiments. A.G., J.Y., S.Y., and R.Y. fabricated the devices. B.L., Y.W., and K.S. were responsible for device performance and film morphology characterization. Q.X. carried out TA measurements. A.G., Z.L., and J.W. characterized and analyzed the various photoelectric properties. A.G. and C.L. co-wrote the paper. B.Z. and Y.M. revised the manuscript.

Competing interests

The authors declare no competing interests.

Additional information

Supplementary information The online version contains supplementary material available at <https://doi.org/10.1038/s41467-025-67334-y>.

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Peer review information *Nature Communications* thanks Peng Wang, Yi Zhang, and the other anonymous reviewer(s) for their contribution to the peer review of this work. A peer review file is available.

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