

# Atomically confined excitons in 2D perovskites for bright and sub-nanosecond scintillation

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Jiaqi Liu<sup>1</sup>, Mingquan Liao<sup>1,2</sup>, Riccardo Latella<sup>3,4</sup>, Junhui Yuan<sup>5</sup>, Zheng Liu<sup>6</sup>, Jiaxin Wen<sup>1</sup>, Qinghao Ling<sup>2</sup>, Yinsheng Xu<sup>7</sup>, Georgios Konstantinou<sup>3,8</sup>, Paul Lecoq<sup>3,4</sup>, Mengling Xia<sup>1</sup>✉ & Guangda Niu<sup>1</sup>✉

Ultrafast radiation detection is crucial for advancing medical imaging, high-energy physics, astronomy, and industrial applications, offering high spatio-temporal resolution and reduced radiation exposure. However, the bottleneck lies in the formidable challenge of achieving scintillators with both ultrafast response and high efficiency. Contrasting with previous approaches that focused on molecular-level confinement, here we propose the strategy of pushing exciton confinement to the limit of atomic scale. Using 2D perovskites as a model, we design organic A-site cations to selectively enhance in-plane distortion to localize excitons, while suppressing out-of-plane and intra-octahedral distortions to minimize the formation of inefficient, long-lived self-trapped excitons. Specifically, (1,4-CMA)PbBr<sub>4</sub> exhibits a rare combination of fast response (0.62 ns) and high light yield (19,700 photons MeV<sup>-1</sup>), surpassing leading commercial and research scintillators. These properties enable breakthroughs in advanced imaging, including fast positron emission tomography with timing precision of 43.3 ps and high-resolution X-ray imaging (32 lp mm<sup>-1</sup>).

Scintillators, as a fundamental technology, are widely applied in medicine, high-energy physics, astronomy, and industry, where fast response and high efficiency are universally pursued goals<sup>1–3</sup>. Ideal scintillators with these properties enable high spatiotemporal resolution while minimizing radiation doses for medical X-ray computed tomography (CT) and positron emission tomography (PET) scans, as well as aid in registering intense events for vertex localization in high-energy physics, astronomy, and nuclear reaction<sup>4,5</sup>. Nevertheless, achieving scintillators with both ultrafast response and high efficiency remains a formidable challenge<sup>6,7</sup>. According to Fermi's Golden Rule, response speed is proportional to the transition moment between

initial and final states, whereas efficiency corresponds to the density of emission states<sup>8,9</sup>. Conventional inorganic scintillators (e.g., LaBr<sub>3</sub>; Ce, LYSO: Ce) exhibit long response times on the order of tens of nanoseconds due to the limited dipole moment of rare earth d-f transitions<sup>7,10,11</sup>. While core-valence and intra-band luminescence (e.g., BaF<sub>2</sub>) can enhance transition matrix and achieve sub-nanosecond response, charge migration within bands seriously reduces the density of emission states, leading to poor efficiencies (<3000 photons MeV<sup>-1</sup>)<sup>12</sup>.

Exciton-based scintillators have emerged as promising alternatives<sup>13</sup>. Stronger exciton confinement enhances both the

<sup>1</sup>School of Materials Science and Engineering, Wuhan University of Technology, Wuhan, China. <sup>2</sup>Wuhan National Laboratory for Optoelectronics (WNLO), Huazhong University of Science and Technology (HUST), Wuhan, China. <sup>3</sup>Metacrystral S.A, Geneva, Switzerland. <sup>4</sup>Instituto de Instrumentación para Imagen Molecular (I3M), Centro Mixto CSIC—Universitat Politècnica de València, Valencia, Spain. <sup>5</sup>School of Physics and Mechanics, Wuhan University of Technology, Wuhan, China. <sup>6</sup>Paul C. Lauterbur Research Center for Biomedical Imaging, Shenzhen Institutes of Advanced Technology, Chinese Academy of Sciences, Shenzhen, China. <sup>7</sup>State Key Laboratory of Silicate Materials for Architectures, Wuhan University of Technology, Wuhan, China. <sup>8</sup>École Polytechnique Fédérale de Lausanne (EPFL), Neuchâtel, Switzerland. ✉e-mail: [xiamengling@whut.edu.cn](mailto:xiamengling@whut.edu.cn); [guangda\\_niu@hust.edu.cn](mailto:guangda_niu@hust.edu.cn)

transition dipole moment and the emission state density, offering the potential for both ultrafast response and high efficiency. As a result, materials like nanocrystals<sup>14–16</sup>, organic materials<sup>17–21</sup>, and 2D perovskites<sup>22,23</sup> have achieved response speeds below 10 ns. However, the sub-nanosecond barrier has rarely been broken. The fundamental limitation lies in the insufficient degree of exciton confinement in current schemes, which only allow for molecular-level confinement, typically involving hundreds to thousands of atoms (Fig. 1a). For example, nanocrystals comprise >1000 atoms, while organic molecules contain >100 atoms. In 2D perovskites, although excitons are confined in the out-of-plane direction within a single lattice (three atoms), they still extend across tens or more atoms in the in-plane direction, resulting in either unsatisfactory light yield or slow response speed.

To overcome this dilemma, we propose the strategy of pushing exciton confinement to the limit of atomic scale. Using 2D perovskites as a model system, we apply in-plane lattice distortion along with out-of-plane dielectric screening to achieve atomic-level exciton confinement (Fig. 1b). The key lies in the synergistic regulation of steric hindrance and polarity of organic A-site cations to selectively enhance in-plane distortion while suppressing out-of-plane and intra-octahedral distortions. The latter two distortions, if not properly controlled, can lead to inefficient, long-lived self-trapped exciton (STE) states (Supplementary Fig. 1)<sup>24–31</sup>.

Following these rules, we design a series of 2D perovskites with tailored A-site cations, including cyclohexanemethylamine (CMA), 4-aminomethyl-1-cyclohexanecarboxylate (CMA-COOH), and cyclohexane-1,4-diyldimethanamine (1,4-CMA) (Supplementary Fig. 2). All three compounds fall into the rare category of high light yield and fast response (Fig. 1c, Supplementary Table 1). Among them, (1,4-CMA) PbBr<sub>4</sub> achieves a remarkable level of exciton confinement by combining strong in-plane tilting with intense out-of-plane dielectric

confinement. This effective confinement in all dimensions endows (1,4-CMA)PbBr<sub>4</sub> with a rare combination of sub-nanosecond radioluminescence lifetime (0.62 ns) and high light yield (19,700 photons MeV<sup>-1</sup>). These properties enable breakthroughs in advanced imaging applications. We demonstrate PET detection with a coincidence timing resolution (CTR) of 43.3 ps, allowing for reconstruction-free PET imaging with significantly reduced radiation dose and acquisition time. Additionally, its fast response ensures minimal afterglow, enabling dynamic X-ray imaging with high resolution (32 lp mm<sup>-1</sup>).

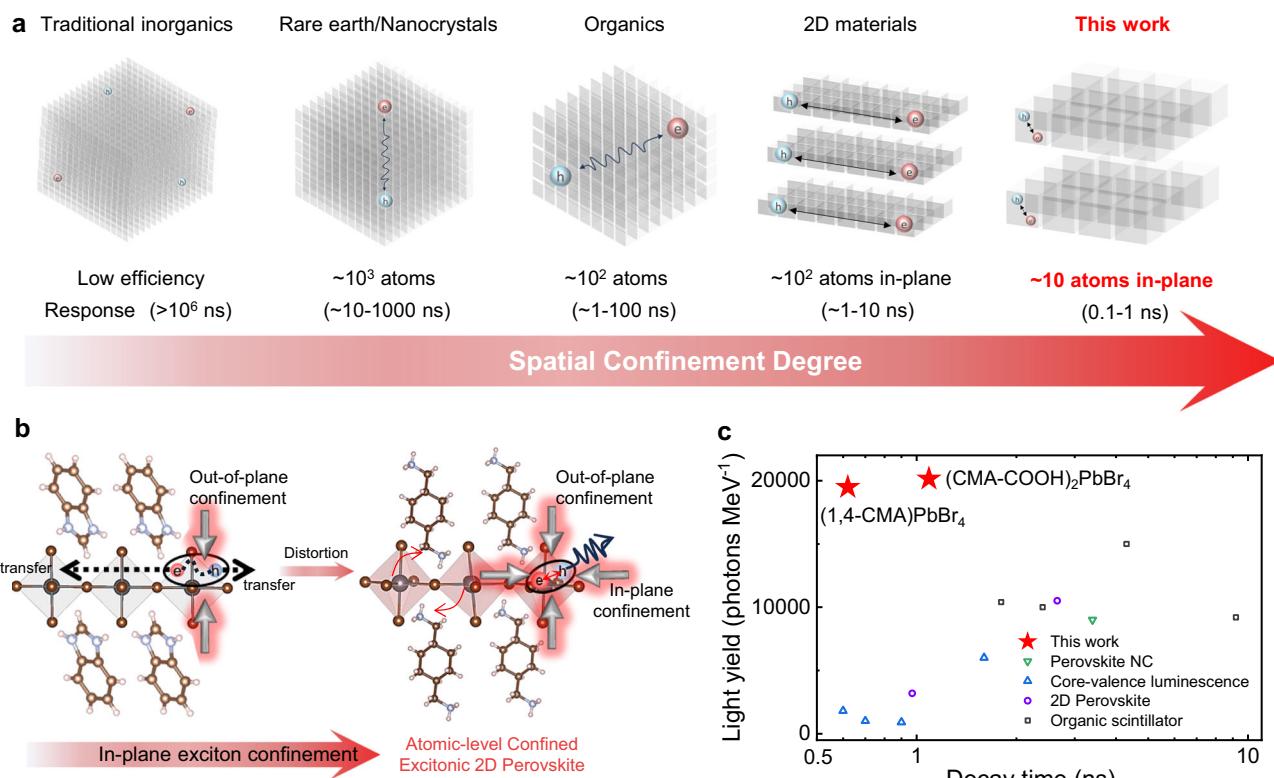
## Results

### Structural distortions influenced by A-site cations

First, we investigate the exciton confinement behavior in 2D perovskites. While their layered structure inherently isolates adjacent layers and the low dielectric constant of A-site cations ensures well-established out-of-plane quantum and dielectric confinement, the in-plane exciton behavior and the influence of lattice distortions remains less understood.

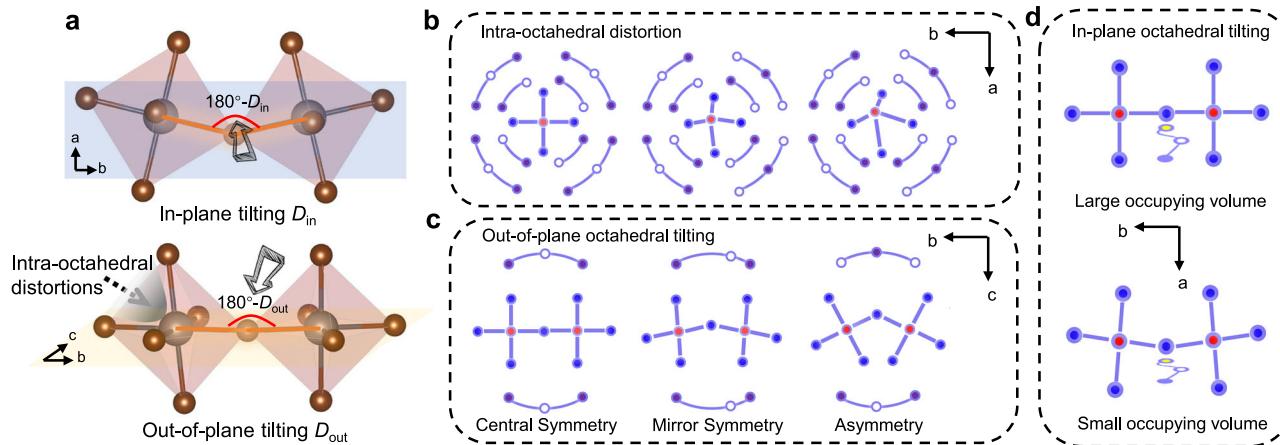
Structural distortions in 2D perovskites primarily fall into three categories: intra-octahedral distortions, where Pb<sup>2+</sup> deviates from the octahedral center, altering the Br-Pb-Br bond angles; out-of-plane octahedral tilting, which changes the Pb-Br-Pb bond angles along the stacking direction; and in-plane octahedral tilting, which affects Pb-Br-Pb bond angles within the inorganic plane (Fig. 2a). By analyzing our synthesized 2D perovskite materials alongside previously reported results, we identify a common trend—both intra-octahedral distortions and out-of-plane tilting tend to induce STE states, leading to slow and inefficient emission (Supplementary Fig. 3). Consequently, in-plane tilting emerges as a rare but effective strategy to achieve exciton confinement without triggering STE formation.

We further examine the structural origins of these three types of distortions and their correlation with A-site molecular properties



**Fig. 1 | Exciton confinement in scintillators.** **a** Schematic of exciton confinement in various scintillators and the corresponding response time range. **b** Design concept of atomic-level exciton confinement in 2D perovskites by regulated in-plane

tilting. **c** The summary of light yield and radioluminescence lifetime of commercial, state-of-the-art scintillators and our atomic-level confined excitonic 2D perovskite.



**Fig. 2 | Structural distortions influenced by A-site cations in 2D perovskites.**

**a** Schematic of in-plane tilting ( $D_{in}$ ), out-of-plane tilting ( $D_{out}$ ) and intra-octahedral distortion ( $\theta^2$ ).

**b** The influence of A-site cations on intra-octahedral distortion.

**c** The influence of A-site cations on out-of-plane tilting.

**d** The influence of A-site cations on in-plane octahedral tilting. Red balls represent Pb atoms, blue balls represent Br atoms, hollow ball represents the electropositive region of organic cations, and purple ball represents the electronegative region.

(Fig. 2b–d). The intra-octahedral distortion is influenced by the symmetric environment created by the nearest eight A-site cations, which exert electrostatic attraction and repulsion on the central  $\text{Pb}^{2+}$ <sup>32</sup>. In a highly symmetric arrangement,  $\text{Pb}^{2+}$  remains at the octahedral center, whereas asymmetric polarization induces its displacement (Fig. 2b). Thus, intra-octahedral distortion value ( $\theta^2$ ) is primarily affected by the C–N symmetry of the nearest A-site molecules, as well as the spatial arrangement of their outer carbon skeletons. For example, in  $[\text{SN}]_2\text{PbBr}_4$  (S–N represents S–(–)–1-(1-naphthyl)ethylammonium), the highly asymmetric C–N configuration induces significant polarity, resulting in pronounced intra-octahedral distortion (Supplementary Fig. 4)<sup>33</sup>. Meanwhile, out-of-plane tilting ( $D_{out}$ ) is more significantly affected by the centrosymmetry of the four organic molecules in the adjacent upper and lower layers (Fig. 2c). In  $(1,3\text{-PDA})\text{PbBr}_4$  (1,3-PDA represents 1,3-phenylenediammonium), the asymmetric arrangement of both the C–N ends and phenylamine across the upper and lower octahedral layers leads to substantial out-of-plane distortion (Supplementary Fig. 5)<sup>34</sup>. On the other hand, the steric hindrance of the functional ammonium group, which fits in the gap between Pb–Br octahedra, determines the in-plane tilting (Fig. 2d). Smaller functional groups induce more contraction and in-plane tilting (Supplementary Fig. 6)<sup>35,36</sup>.

Based on these findings, we identify key design principles for ideal A-site cations: 1) Strong symmetry in both the C–N configuration and the outer molecular skeleton to minimize intra-octahedral distortions. 2) Centrosymmetric arrangement across layers, eliminating polarity-driven out-of-plane tilting. 3) Small functional ammonium groups to facilitate in-plane tilting.

### Structure–property relationship

To implement these principles, we design non-conjugated molecular additives (CMA) and their derivatives as A-site cations to create a highly symmetric environment (Supplementary Fig. 2 and Supplementary Fig. 7). Unlike previously reported materials, three compounds— $(\text{CMA})_2\text{PbBr}_4$ ,  $(\text{CMA-COOH})_2\text{PbBr}_4$  and  $(1,4\text{-CMA})\text{PbBr}_4$ —exhibit pronounced in-plane tilting while maintaining minimal intra-octahedral and out-of-plane distortions ( $< 2^\circ$ ) (Fig. 3a, b, Supplementary Fig. 3 and Supplementary Tables 2–6). Their photoluminescence spectra feature remarkably narrow emission peaks (FWHM of ~30 nm), and temperature-dependent measurements reveal minimal linewidth broadening, providing strong evidence against STE formation (Fig. 3c). To further validate our design rationale, we introduce an asymmetric A-site cation, 1,3-CMA, as a control to induce an asymmetric

polarization environment around Pb–Br octahedron. As expected,  $(1,3\text{-CMA})\text{PbBr}_4$  exhibits significant out-of-plane tilting and intra-octahedral distortion, accompanied by a broad STE emission peak (Inset of Fig. 3c).

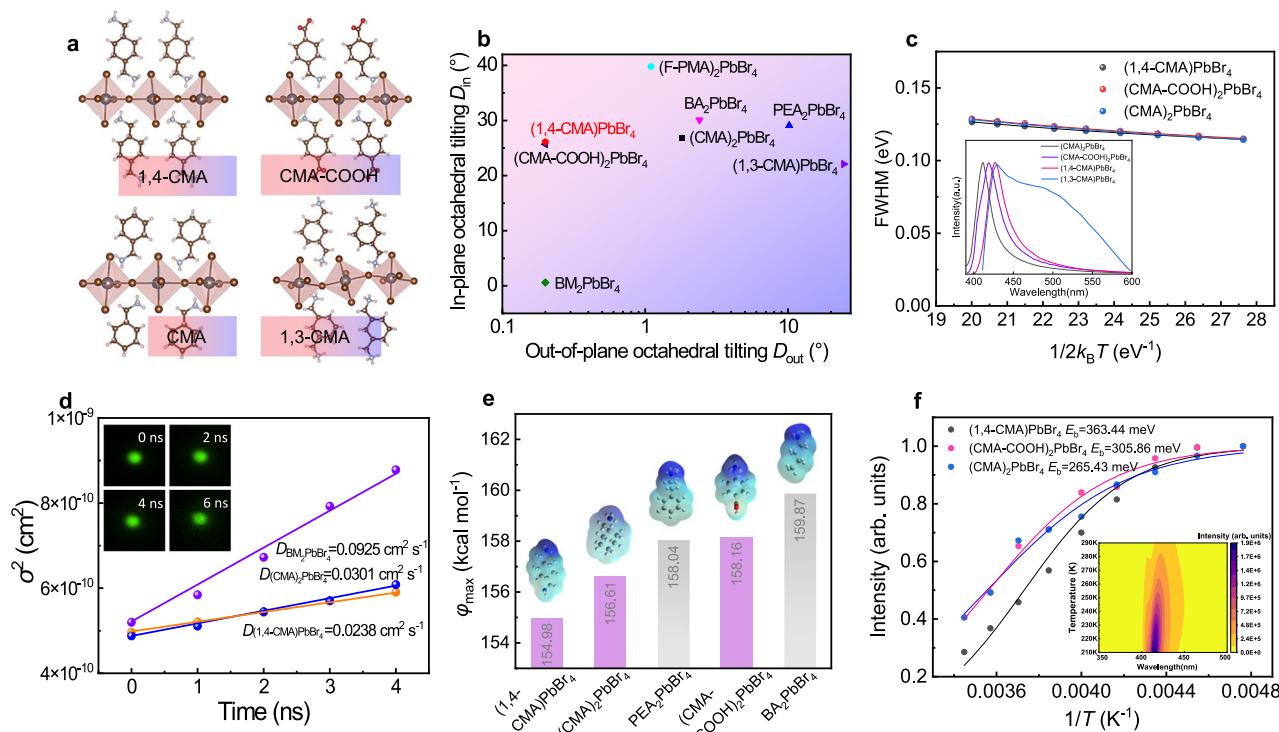
We then quantify exciton diffusion dynamics by performing transient photoluminescence microscopy (TPLM) on single crystals (Fig. 3d and Supplementary Fig. 8–9 and Supplementary Note I). The diffusion constant  $D$  is determined by  $D = (\sigma_t^2 - \sigma_0^2)/4t$ , where  $\sigma_t^2$  represents the time-dependent variance of the exciton population, and  $\sigma_0^2$  corresponds to the initial variance. The diffusion coefficient of  $(1,4\text{-CMA})\text{PbBr}_4$  is  $0.0238 \text{ cm}^2 \text{s}^{-1}$ , significantly smaller than  $\text{BM}_2\text{PbBr}_4$  ( $0.0925 \text{ cm}^2 \text{s}^{-1}$ ), indicating much stronger in-plane exciton confinement than  $\text{BM}_2\text{PbBr}_4$ . The serious in-plane exciton diffusion of  $\text{BM}_2\text{PbBr}_4$  also explains its relatively low light yield (~3,000 photons  $\text{MeV}^{-1}$ )<sup>23</sup>.

We also calculate the electrostatic potentials of the organic amine molecules (Fig. 3e and Supplementary Fig. 10). 1,4-CMA exhibits the lowest electrostatic potential, which enables stronger out-of-plane dielectric confinement<sup>37,38</sup>. Consequently, the exciton binding energy ( $E_b$ ) of  $(1,4\text{-CMA})\text{PbBr}_4$  reaches 363.4 meV, slightly higher than  $(\text{CMA-COOH})_2\text{PbBr}_4$  (305.9 meV) and  $\text{CMA}_2\text{PbBr}_4$  (265.4 meV) (Fig. 3f), and much higher than previously reported 2D perovskites, including  $\text{PEA}_2\text{PbBr}_4$  (190 meV) and  $\text{BA}_2\text{PbBr}_4$  (260 meV)<sup>23,39</sup>.

We further evaluate the photoluminescence performance through fluence-dependent photoluminescence quantum yield (PLQY) measurements, which reveal that all samples exhibit a characteristic rise-to-roll-off trend (Supplementary Fig. 11). The initial rise suggests efficient filling of non-radiative trap states at lower excitation densities. The subsequent universal roll-off is attributed to intensified exciton–exciton interactions at high exciton densities, where the decreased average distance between excitons significantly raises the probability of synergistic energy loss through multiple recombination pathways<sup>31</sup>.

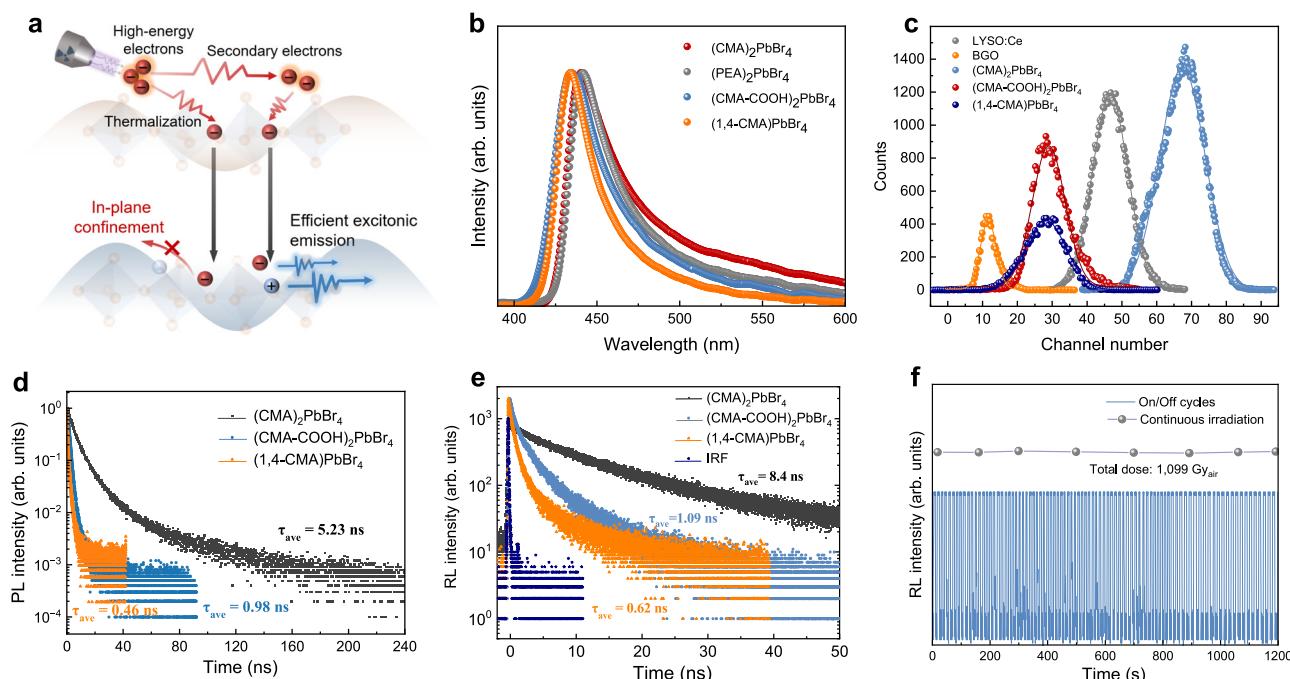
### Scintillation performance

We then characterize the radioluminescence (RL) properties. During the scintillation process under X-ray excitation, primary high-energy electrons are initially generated and rapidly thermalize (on the picosecond to femtosecond timescale) to form band-edge excitons. The response speed of scintillators is determined by the slow recombination process of these excitons near band edges (Fig. 4a). The atomic-level confined exciton in these 2D perovskites enable efficient and fast recombination. To verify their radioluminescence properties, we grow high-quality, large single crystals of  $(\text{CMA})_2\text{PbBr}_4$ ,  $(\text{CMA-COOH})_2\text{PbBr}_4$ , and  $(1,4\text{-CMA})\text{PbBr}_4$  (Fig. 4b). The RL spectra of these samples show a broad emission band centered at ~420 nm (Fig. 4c), which is consistent with the PL spectra (Fig. 3d). The RL intensity is significantly enhanced for the 2D perovskites compared to the 3D perovskites, indicating their potential as efficient scintillators (Fig. 4d). The RL decay kinetics of the 2D perovskites are also faster than the 3D perovskites, suggesting a faster recombination process (Fig. 4e). The RL efficiency of the 2D perovskites is also higher than the 3D perovskites, indicating their potential as efficient scintillators (Fig. 4f).



**Fig. 3 | Structure-property relationship between structural distortions and exciton confinement in 2D perovskites.** **a** Crystal structures of  $(1,4\text{-CMA})\text{PbBr}_4$ ,  $(\text{CMA})_2\text{PbBr}_4$ ,  $(\text{CMA-COOH})_2\text{PbBr}_4$  and  $(1,3\text{-CMA})\text{PbBr}_4$ . **b** The summary of in-plane tilting ( $D_{in}$ ) and out-of-plane tilting ( $D_{out}$ ) of representative 2D perovskites. **c** Temperature-dependent full width at half maximum (FWHM) of PL spectra for the synthesized 2D perovskites. The inset is the PL spectra at room temperature. **d** The

exciton diffusion dynamics of  $\text{BM}_2\text{PbBr}_4$ ,  $(\text{CMA})_2\text{PbBr}_4$  and  $(1,4\text{-CMA})\text{PbBr}_4$ . The inset shows the transient photoluminescence mapping of the  $(1,4\text{-CMA})\text{PbBr}_4$  single crystal.  $D$  represents carrier diffusion coefficient. **e** Electrostatic potential ( $\varphi_{max}$ ) of the representative amines in 2D perovskites. **f** Temperature-dependent intensity of PL spectra and the fitted excitonic binding energy ( $E_b$ ). The inset is the temperature-dependent PL spectra for  $(1,4\text{-CMA})\text{PbBr}_4$ .



**Fig. 4 | Scintillation performance of atomic-level confined excitonic 2D perovskites.** **a** Schematic of exciton recombination processes under X/ $\gamma$ -ray excitation. **b** Normalized radioluminescence spectra of 2D perovskites. **c** Pulse height

spectra of BGO, LYSO:Ce,  $(\text{CMA})_2\text{PbBr}_4$ ,  $(\text{CMA-COOH})_2\text{PbBr}_4$  and  $(1,4\text{-CMA})\text{PbBr}_4$  under  $^{241}\text{Am}$  (59.5 keV) excitation. **d** PL decay. **e** RL decay. **f** Radiation stability of  $(1,4\text{-CMA})\text{PbBr}_4$ .

$\text{COOH}_2\text{PbBr}_4$ , (1,4-CMA) $\text{PbBr}_4$ , and (1,3-CMA) $\text{PbBr}_4$  using the slow cooling solution method according to their respective solubility curves (Supplementary Fig. 12). The basic properties, including optical images, absorption spectra and band structures, are shown in Supplementary Figs. 13–15.

As presented in Fig. 4b, the RL spectra of these single crystals are measured using an integrating sphere, with the well-studied  $\text{PEA}_2\text{PbBr}_4$  single crystal of the same size and thickness serving as a reference. The RL spectra closely resemble their corresponding PL spectra excited by ultraviolet light (inset of Fig. 3c), indicating the same radiative recombination pathway. (CMA) $_2\text{PbBr}_4$ , (CMA-COOH) $_2\text{PbBr}_4$ , and (1,4-CMA) $\text{PbBr}_4$  all exhibit narrowband RL emission while (1,3-CMA) $\text{PbBr}_4$  displays broadband STE emission. The scintillation light yields of (CMA) $_2\text{PbBr}_4$ , (CMA-COOH) $_2\text{PbBr}_4$ , and (1,4-CMA) $\text{PbBr}_4$  are evaluated relative to a LYSO:Ce reference (32,000 photons  $\text{MeV}^{-1}$ ) using standard pulse-height analysis. In this setup, the crystals are optically coupled to a photomultiplier tube (PMT), and the signals excited by  $^{241}\text{Am}$  (59.5 keV) are recorded on an oscilloscope. The acquired spectra are presented in Fig. 4c. We calculate the yield based on the channel number of the full-energy peak centroid, incorporating a correction for the PMT's wavelength-dependent efficiency (Supplementary Fig. 16). The full energy peak centroids for the three crystals are located at the channel number 68.52, 28.99, and 27.99, which are 1.47, 0.62, and 0.60 times that of the LYSO:Ce standard, respectively. Thus, the scintillation yields of (CMA) $_2\text{PbBr}_4$ , (CMA-COOH) $_2\text{PbBr}_4$ , and (1,4-CMA) $\text{PbBr}_4$  are estimated to be 50,460, 20,460, and 19,730 photons  $\text{MeV}^{-1}$ , respectively.

Furthermore, we record the PL and RL decay dynamics of these scintillators using a pulsed 367 nm laser and a pulsed X-ray source, respectively. The average lifetime of (CMA) $_2\text{PbBr}_4$  (CMA-COOH) $_2\text{PbBr}_4$  and (1,4-CMA) $\text{PbBr}_4$  under laser excitation was 5.23 ns, 0.98 ns and 0.46 ns, respectively (Fig. 4d). However, the RL decay was slightly slower than PL decay (Fig. 4e), primarily because X-rays excites high-energy electrons, which are more likely to escape from their exciton-bound states, leading to longer lifetimes. This difference is further accentuated by enhanced non-radiative recombination at the surface under PL excitation—a contrast to the bulk-dominated carrier generation and comparatively lower defect-mediated recombination in RL. This also implies that, due to the energy being 3–5 orders of magnitude higher, achieving fast RL is much more challenging than for materials like phosphors or OLEDs, making exciton confinement even more crucial. For a more rigorous comparison, we adopt RL decay time. We find that (1,4-CMA) $\text{PbBr}_4$  achieves an average scintillation lifetime of 0.62 ns, while (CMA-COOH) $_2\text{PbBr}_4$  and (CMA) $_2\text{PbBr}_4$  exhibit average lifetimes of 1.09 ns and 8.4 ns, respectively. The fast and slow component of RL decay for (CMA) $_2\text{PbBr}_4$  is 4.37 ns and 9.61 ns, while for (CMA-COOH) $_2\text{PbBr}_4$ , they are 0.61 and 2.13 ns, respectively. Notably, the fast and slow component of RL decay for (1,4-CMA) $\text{PbBr}_4$  reach as short as 0.25 and 1.16 ns, respectively (Supplementary Table 7). In comparison,  $\text{BaF}_2$ <sup>40</sup>—one of the fastest commercial scintillators—exhibit the fast component of 0.6 ns, lagging behind (1,4-CMA) $\text{PbBr}_4$ , underscoring the advantage of our materials.

We also assess the stability of the fastest scintillator (1,4-CMA) $\text{PbBr}_4$  toward moisture and continuous radiation (Fig. 4f and Supplementary Fig. 17). No detectable weight loss is recorded for the sample following a 35-day period at 70% relative humidity. In terms of radiation stability, the light output is measured while the bare single crystal (without encapsulation) is irradiated with X-rays in an ambient environment (25 °C and 70% relative humidity), as shown in Fig. 4f. After continuous X-ray irradiation (916.09 mGy  $\text{s}^{-1}$ ) and repeated on-off X-ray cycles, accumulating a total dose of 1,099 Gy<sub>air</sub>, no deterioration in light output was observed. This dose is equivalent to 20,000 times CT scans, demonstrating exceptional radiation hardness. The structural and optical integrity after testing are confirmed by XRD and PL measurements, which show negligible degradation (Supplementary

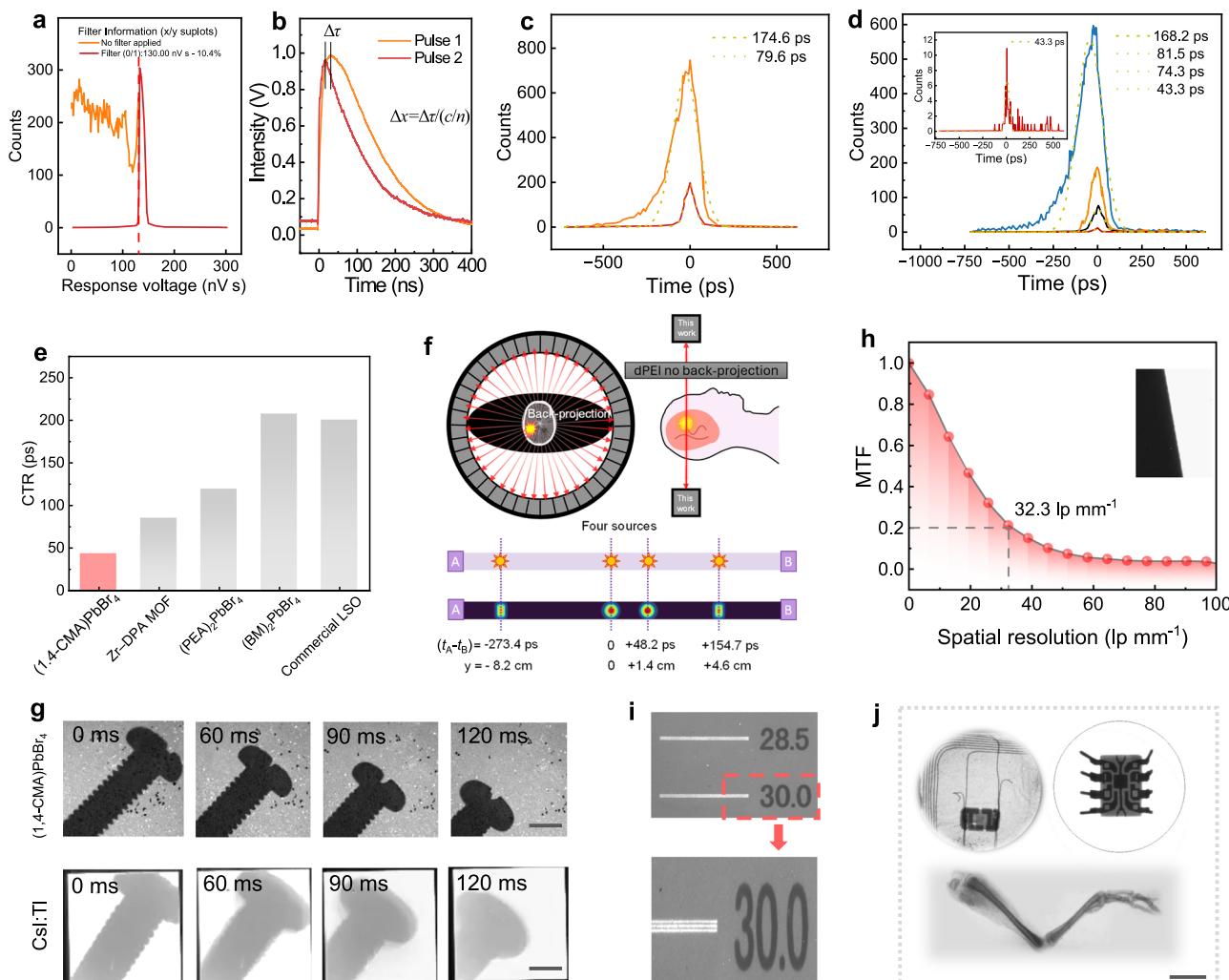
Figs. 18–19). The scintillation kinetics are largely retained post-irradiation (Supplementary Fig. 20), and the variations are consistent with the creation and evolution of point defects<sup>41,42</sup>, as evidenced by the stable crystal structure (Supplementary Fig. 18).

### Applications in fast radiation detection and imaging

Next, we investigate the advantages of the scintillator for time-of-flight PET (TOF-PET) and dynamic X-ray imaging. In TOF-PET, CTR serves as a critical performance parameter, where a lower CTR results in better spatial resolution<sup>43</sup>. LYSO:Ce,Ca crystals were selected as a benchmark reference due to their rapid timing characteristics and high stopping power (Supplementary Fig. 21). An alternating stack consisting of three naturally grown (1,4-CMA) $\text{PbBr}_4$  single crystal layers and four manually cut LYSO:Ce,Ca layers was constructed, with all surfaces polished. The overall dimensions were  $3 \times 2.8 \times 3 \text{ mm}^3$  and covered with enhanced specular reflector films, wrapped in Teflon thereafter, except for the light extraction face. This dual-layer coating can effectively reduce light escape from the crystal corners<sup>44</sup>. The reference and sample crystals are directly coupled to the optical windows of the SiPMs. A pair of detectors detects coincidence signals from two back-to-back 511 keV  $\gamma$ -photons produced by positron annihilation from a  $^{22}\text{Na}$  radioactive source. A coincidence event was recorded when two photons trigger both detectors within energy and time difference windows. The signals were selected in a coincidence window of  $\pm 1 \text{ ns}$  filtered through an energy/voltage filter to isolate those in the photopeak region of the energy spectrum, using the valley between Compton edge and photopeak as the reference point (Fig. 5a). The acquisitions were post-processed using custom Python code to detect features like different CTR voltage thresholds for each voltage bias, analyzing and sanitizing acquired events, as detailed in previous work<sup>45</sup>.

The integrated charge spectrum shows clear full energy peaks with an energy resolution of 10.4% at 511 keV (Fig. 5a). To enhance the effective count rate of the system, a laminated configuration consisting of LYSO and perovskite was implemented. By employing a filter (0/1, 130 nV s) to process the coincidence pairs of output pulses and performing overlapping integration, a CTR of  $79.6 \pm 0.8 \text{ ps}$  was achieved for the LYSO/(1,4-CMA) $\text{PbBr}_4$  laminated detector (Fig. 5b, c), smaller than the intrinsic CTR of the LYSO detector itself of  $83 \pm 3 \text{ ps}$ . To further obtain the intrinsic CTR of (1,4-CMA) $\text{PbBr}_4$  independently, a single-layer perovskite was directly coupled to SiPM surface through optical grease. Remarkably, a record CTR of 43.3 ps of (1,4-CMA) $\text{PbBr}_4$  detector was obtained (Fig. 5d). This performance significantly surpasses that of commercial scanners (~200 ps), previously reported  $\text{BM}_2\text{PbBr}_4$  (207 ps), and  $\text{PEA}_2\text{PbBr}_4$  (119 ps), and  $\text{Zr-DPA MOF}$  (85 ps), highlighting the excellent temporal resolution of the (1,4-CMA) $\text{PbBr}_4$  (Fig. 5e)<sup>22,23,46</sup>. The improved timing resolution for TOF-PET implies the ability to diagnose smaller lesions or tumors. When the timing resolution is sufficiently precise to directly localize the source, a new regime is entered, where images can be obtained directly without the need for reconstruction steps. As shown in Fig. 5f, traditional TOF-PET requires image reconstruction to precisely locate the source position. Our detector, leveraging its ultra-short resolution, eliminates the need for reconstruction and directly accomplishes the task with just a pair of detectors. Beyond medical applications, the fast response and high efficiency of these scintillators make them highly promising for astrophysical cosmic ray detection, nuclear monitoring, border control, and high-energy physics experiments, all of which require fast response to mitigate pile-up background and accurately localize event vertexes.

Benefiting from the short decay time, (1,4-CMA) $\text{PbBr}_4$  demonstrates a suppressed afterglow. Residual images of a round nut with a camera after 10 sec of X-ray exposure were captured (Supplementary Fig. 22). (1,4-CMA) $\text{PbBr}_4$  presents negligible residual images in all frames, whereas the commercial  $\text{CsI:Tl}$  film shows distinct residual images even after 2 s. As shown in Fig. 5g, a screw was mounted on a



**Fig. 5 | Application of (1,4-CMA)PbBr<sub>4</sub> scintillator in fast radiation detection and imaging.** **a** Integrated charge spectra for 511 keV photons along with the intervals filtered by different response voltages. **b** Decay curves collected by the two opposite detectors, and the location ( $\Delta x$ ) can be calculated based on the time difference ( $\Delta t$ ) via  $\Delta x = \Delta t c/n$ . **c** The speed of light in a vacuum;  $n$ , the refractive index of the medium. **c** Delay time histograms of LYSO:Ce,Ca against (1,4-CMA)PbBr<sub>4</sub> before (orange line) and after (red line) photopeak event selection. **d** Delay time histograms of (1,4-CMA)PbBr<sub>4</sub> before photopeak event selection (blue line) and after photopeak event selection with different filters (orange, black and red lines). The FWHMs are extracted using a Gaussian fit. The inset is the enlarged curve

of the red line showing the Coincidence time resolution (CTR) of 43.3 ps. **e** CTR comparison of state-of-the-art scintillators. **f** Advantages of the high timing resolution in reconstruction-free PET imaging, and the corresponding spatial precision. **g** Dynamic X-ray imaging of a moving screw by (1,4-CMA)PbBr<sub>4</sub> and commercial CsI:Tl scintillator. The scale bar represents 5 mm. **h** Modulation transfer function (MTF) curve. **i** X-ray image of the standard X-ray resolution test chart plate. **j** High-resolution X-ray images (campus card, chip, bird leg). The bird leg specimen used for X-ray imaging was obtained from a commercial food market; no live animals were used in this experiment. The scale bar represents 1 cm.

slide-traction plate, and the steel plate applied a pulling force to the screw while continuous X-ray irradiation occurred (Supplementary Fig. 23). The image was captured in an interval of 30 ms. As the screw moved, the serrated structure on the outer edge of the screw was clearly visible with no residual shadow.

Furthermore, featuring a high light yield, (1,4-CMA)PbBr<sub>4</sub> exhibits a spatial resolution of 32.3 lp mm<sup>-1</sup> at a modulation transfer function (MTF) of 0.2, measured by the slanted-edge method (Fig. 5h). This performance outperforms commercial CsI:Tl (3–5 lp mm<sup>-1</sup>) and GOS:Tb (4–8 lp mm<sup>-1</sup>), and also exceeds recently reported perovskite (22 lp mm<sup>-1</sup>) and organic scintillators (18 lp mm<sup>-1</sup>)<sup>13,20,47–49</sup>. X-ray images of a standard resolution test chart indicate that the line pairs at 30 lp mm<sup>-1</sup> remain clearly resolvable (Fig. 5i), validating this high spatial resolution. This exceptional imaging resolution enables the precise identification of ultra-fine internal structures, as demonstrated in a campus card (showing embedded components), an integrated circuit

chip (revealing trace distributions), and a bird leg (visualizing bone structure) (Fig. 5j).

Finally, we evaluate the practical performance of (1,4-CMA)PbBr<sub>4</sub> by comparing its radiation hardness and X-ray stopping power with commercial BaF<sub>2</sub> (Supplementary Fig. 24). While BaF<sub>2</sub> exhibits superior radiation resistance and higher stopping power, (1,4-CMA)PbBr<sub>4</sub> demonstrates reasonable radiation hardness (~2% degradation after 5 kGy<sub>air</sub>) despite its lower attenuation coefficient.

## Discussion

In summary, we elucidate the structure-property relationship between long-chain organic cations, various structural distortions, and the resulting exciton confinement in 2D perovskites. By carefully screening A-site cations with highly symmetric structures and small functional ammonium groups, we develop a series of fast and efficient scintillators with atomic-level exciton confinement.

Our strategy overcomes a fundamental limitation in luminescent materials: the typical correlation between strong exciton localization and STE emission. By precisely engineering the lattice distortions, we achieve in-plane exciton confinement without inducing the strong electron-phonon interaction that dictates STE formation. Importantly, the in-plane tilting in our system is a controlled structural feature within a stable lattice framework, which does not compromise the mechanical integrity or phase stability, as evidenced by our stability tests.

Our strategy provides a viable solution to the intrinsic trade-off between efficiency and sub-nanosecond response, surpassing nearly all existing scintillators. This exceptional performance enables significant advancements in medical imaging, including reconstruction-free PET imaging, low-dose CT and dynamic X-ray imaging. Moreover, the rational distortion engineering approach we apply to enhance exciton confinement can be extended to the design of other materials, such as nanocrystals, quantum dots, nanoclusters and low-dimensional perovskites. While the bulk crystals in this study are optimized for scintillation, the underlying photophysics suggests that nanostructured or thin-film versions of these materials could open avenues for efficient electroluminescence. This will further propel the development of applications in LEDs, lasers, quantum light sources, and LiDAR technologies.

## Methods

### Materials

Cyclohexylmethylammonium (97%), 4-(aminomethyl)cyclohexanecarboxylic acid (99%), 1,3-bis(aminomethyl)cyclohexane (98%) were purchased from Adamas Reagent Co. Ltd. 1,4-cyclohexane-diyl-dimethanamine (98%), hydrobromic acid (HBr) (40%), were purchased from Aladdin Chemical Co. Ltd. Lead bromide (99.999%) was purchased from Advanced Electron technology Co. Ltd. All the chemicals were used as received without any further purification.

### Preparation of single crystals and film

The (1,4-CMA)<sub>2</sub>PbBr<sub>4</sub> crystal was synthesized using the solution cooling method. PbBr<sub>2</sub> (0.15 mmol) and cyclohexane-1,4-diyl-dimethanamine (1,4-CMA) (0.15 mmol) were separately dissolved in HBr (2 mL and 4 mL, respectively). The solutions were mixed, heated to 120 °C until dissolved, then cooled at 0.5 °C h<sup>-1</sup> to promote crystal growth. The (CMA)<sub>2</sub>PbBr<sub>4</sub> crystal was synthesized similarly. PbBr<sub>2</sub> (0.5 mmol) and cyclohexylmethylammonium (CMA) (1 mmol) were separately dissolved in 1 ml and 6 ml of HBr. Then the solutions were mixed, heated to 125 °C, then cooled at 2 °C h<sup>-1</sup> to allow crystal growth. The (CMA-COOH)<sub>2</sub>PbBr<sub>4</sub> crystal was synthesized by dissolving PbBr<sub>2</sub> (0.5 mmol) and 4-(aminomethyl)cyclohexanecarboxylic acid (CMA-COOH) (1 mmol) in HBr. The mixed solution was heated to 100 °C and cooled at 2 °C h<sup>-1</sup> to promote crystal growth. The (1,3-CMA)<sub>2</sub>PbBr<sub>4</sub> crystal was synthesized by dissolving PbBr<sub>2</sub> (0.5 mmol) and 1,3-Bis(aminomethyl)cyclohexane (1,3-CMA) (0.5 mmol) in 1 mL and 3 mL of HBr, respectively. The mixed solution was heated to 125 °C and cooled at 2 °C h<sup>-1</sup> for growth. The (1,4-CMA)<sub>2</sub>PbBr<sub>4</sub> crystal was mechanically ground for 30 min to obtain micron-sized powders. Polymethyl methacrylate (PMMA) was dissolved in ortho-xylene by magnetic stirring and heating at 100 °C until fully dissolved. The prepared powders were then mixed with PMMA at a mass ratio of 2:1, followed by mechanical stirring for 24 h. Finally, the mixture was dropped onto a mold. After drying at room temperature for 5–6 h, the film was formed and could be peeled off.

### Materials characterizations

Single-crystal X-ray diffraction measurements were conducted on an XtaLAB PRO MM007HF X-ray diffractometer equipped with a liquid nitrogen cooling system and a Cu K $\alpha$  radiation source. Powder XRD measurements were performed using a Rigaku MiniFlex600 X-ray

diffractometer, utilizing a Mo K $\alpha$  X-ray tube ( $\lambda = 0.71073 \text{ \AA}$ ), operating at 40 kV and 15 mA, with a step size of approximately 0.08° s<sup>-1</sup>. Photoluminescence and photoluminescence excitation spectra were recorded using an FL3-22 time-resolved fluorescence spectrometer. Ultraviolet-visible-near infrared (UV-Vis-NIR) absorption spectra were obtained using a Lambda 750s spectrophotometer in transmission mode. Time-resolved fluorescence spectroscopy measurements were carried out with a Spirit 1040-8-SHG06900 FR Rack LCRI spectrometer, using a 340 nm pulsed laser for excitation. PLQY was measured using an XPQY-EQE photoluminescence efficiency measurement system. For RL and X-ray imaging, we used X-ray tube (Moxtek MAGPRO 70 kV 12 W) as the source. X-ray photons were generated under an accelerating voltage of 50 kV and operating currents from 20 to 140  $\mu$ A. The distance between the X-ray tube and the detector varied from 10 to 80 cm. The X-ray dose rate was calibrated using an ion chamber dosimeter (IBA Dosimetry MagicMax). The stability was recorded using an oscilloscope (RIGOL DS1202Z-E). The RL spectra were recorded with a PG2000-PRO spectrometer. X-ray images were captured using a digital camera (FL 20BW). The RL decay dynamics of the scintillators were characterized using a picoX 5084 X-ray excited scintillation lifetime/decay (TRRL) testing system developed by Shanghai UPU Optoelectronic Technology Co., Ltd. The system employs a Ni6432 photo-excited X-ray tube (Hamamatsu), which generates pulsed X-rays controlled by the illumination of a photocathode. By utilizing a picosecond laser to produce high-repetition-rate, sub-100-picosecond X-ray pulses, the system enables rapid measurements based on time-correlated single-photon counting (TCSPC).

### Structural distortion calculation and statistics

Structural distortion parameters have been calculated (using VESTA software) from the crystallographic data, including both published data and data from crystals synthesized in this work. The intra-octahedral distortion was evaluated by mean of the following parameters<sup>50-53</sup>:

$$\theta^2 = \frac{1}{11} \sum_{i=1}^{12} (\alpha_i - 90)^2 \quad (1)$$

where  $\alpha_i$  are the Br-Pb-Br angles.

The extent of structural distortion within the inorganic framework was quantified through a detailed analysis of the bridging Pb-Br-Pb angles. All calculations were performed based on the Cartesian atomic coordinates derived from the single-crystal X-ray diffraction structure refinement.

The total octahedral tilting ( $D_{\text{tilt}}$ ) for a given Pb-Br-Pb angle is defined as the angular deviation from the ideal linear geometry:  $D_{\text{tilt}} = 180^\circ - \theta_{\text{Pb-Br-Pb}}$ , where  $\theta_{\text{Pb-Br-Pb}}$  was measured from the refined crystallographic information file (CIF) using the angle measurement tool in the VESTA software.

To decouple the distinct components of the total tilt, the three-dimensional Pb-Br-Pb angle was vectorially decomposed into its out-of-plane and in-plane projections. This decomposition was carried out by defining a best-fit plane through the three Pb atoms ( $\text{Pb}_i$ ,  $\text{Pb}_j$ , and an adjacent  $\text{Pb}_k$ ) constituting the local inorganic layer environment. This approach of using atomic positions to define a local plane, rather than relying on the crystallographic (001) plane, was adopted because in certain cases the Pb atoms do not lie perfectly in the (001) plane.

The unit normal vector ( $\hat{\mathbf{n}}$ ) of this locally defined plane was computed. The vectors corresponding to the Pb-Br-Pb bonds were then projected onto the direction of  $\hat{\mathbf{n}}$  (out-of-plane component) and onto the plane itself (in-plane component). The angles between these projected vectors,  $\theta_{\text{out}}$  and  $\theta_{\text{in}}$ , were calculated using the standard arccosine dot product formula (For vectors  $\mathbf{a}$  and  $\mathbf{b}$ , the angle between them is given by  $\theta = \cos^{-1}[(\mathbf{a} \cdot \mathbf{b}) / (|\mathbf{a}| |\mathbf{b}|)]$ ).

From these projections, the respective distortion parameters were defined as: Out-of-plane tilting ( $D_{\text{out}}$ ):  $D_{\text{out}} = 180^\circ - \theta_{\text{out}}$ , In-plane tilting ( $D_{\text{in}}$ ):  $D_{\text{in}} = 180^\circ - \theta_{\text{in}}$ . Here,  $\theta_{\text{out}}$  and  $\theta_{\text{in}}$  represent the angles between the out-of-plane and in-plane projections of the Pb-Br bond vectors, respectively. The uncertainty  $\delta f$  in an arbitrary function  $f(x)$  is  $\delta f = |df/dx| \cdot \delta x$ , hence  $\delta(\cos^{-1}x) = |(1-x^2)^{-1/2}| \cdot \delta x$ . This results in higher error in  $D$  values when the Pb-Br-Pb angles are closer to  $180^\circ$ . This does not represent lower precision in a particular X-ray structure, but is a necessary result of correct error propagation.

Following the calculations, the validity of the vector decomposition was corroborated by verifying that the following relation holds approximately:  $\cos^2(\theta_{\text{tilt}}) \approx \cos^2(\theta_{\text{in}}) + \cos^2(\theta_{\text{out}}) - 1$ , which derives from the spherical law of cosines.

### TOF-PET measurement

TOF-PET measurements were conducted using a custom-made standard apparatus. This setup incorporated electronic boards providing dual outputs (energy and timing) and NUV-HD-MT SiPMs manufactured by Fondazione Bruno Kessler (FBK), tested with various coupling materials<sup>54,55</sup>. The sample crystal pixel was coupled to the SiPM surface using optical grease, and a LYSO:Ce layer was positioned above it. The device has an overall volume of  $3 \times 3 \times 0.4 \text{ mm}^3$  and an extraction face of about  $3 \times 3 \text{ mm}^2$ . For the reference detector, a bulk  $3 \times 3 \times 5 \text{ mm}^3$  LYSO:Ce,Ca crystal sourced from SIPAT (China) was employed. A  $^{22}\text{Na}$  source is placed between a pair of facing gamma-ray detectors to form the coincidence system. The detector unit comprises a scintillator for gamma-to-light conversion, followed by a SiPM that generates electrical signals from these photons. These signals are then conditioned by front-end electronics while preserving a high signal-to-noise ratio. To capture the full pulse dynamics and extract all relevant signal features, we utilized an 8 GHz bandwidth oscilloscope (Rhode & Schwarz RTP084) connected to the timing and energy channels<sup>45</sup>. The device was configured to trigger only on specific voltage thresholds, thereby performing coincidence filtering directly within the hardware.

### First-principles calculations

Density functional theory (DFT) calculations were performed using the Vienna Ab initio Simulation Package (VASP)<sup>56,57</sup>. We adopted the projector augmented-wave (PAW) formalism with a plane-wave energy cutoff set to  $500 \text{ eV}$ <sup>58,59</sup>. The valence electron configurations were defined as  $6s^26p^2$  for Pb,  $4s^24p^5$  for Br,  $2s^22p^3$  for N,  $2s^22p^2$  for C, and  $1s^1$  for H. To describe the exchange-correlation interactions, we employed the Perdew-Burke-Ernzerhof (PBE) functional within the generalized gradient approximation (GGA)<sup>60</sup>. The electronic self-consistency was achieved with a convergence threshold of  $10^{-6} \text{ eV}$ . To account for van der Waals forces in the (1,4-CMA)PbBr<sub>4</sub> system, the DFT-D3<sup>61</sup> correction was included. Structural relaxations were conducted until the residual forces on ions dropped below  $0.01 \text{ eV \AA}^{-1}$ . Spin-orbit coupling (SOC) effects were incorporated into the electronic structure calculations. The shell GGA-1/2 (shGGA-1/2) method<sup>62,63</sup> was employed to mitigate the inherent band gap inaccuracy of DFT. In the case of Br within (1,4-CMA)PbBr<sub>4</sub>, we applied optimal inner and outer cutoff radii of 0.6 Bohr and 2.9 Bohr, respectively<sup>64</sup>. The Gaussian 16 W package was utilized to compute the electrostatic potentials ( $\varphi$ ) for the organic A-site cations, applying the B3LYP functional with the 6-31 G(d) basis set. For the analysis of the NH<sup>3+</sup> terminal group, the maximum potential  $\varphi$  ( $\varphi_{\text{max}}$ ) was extracted via the Multiwfn code<sup>65</sup>.

### Calculation of scintillation light yield

We performed a cross-verification using multiple standard scintillators with different light yields and emission wavelengths to ensure the accuracy of our test setup and calibration procedure. We employed two certified reference scintillators (LYSO:Ce and BGO). LYSO:Ce (purchased from Meishan Bora New Materials Co., Ltd., China):

Certified light yield of  $\sim 32,000$  photons  $\text{MeV}^{-1}$ , emission peak at  $\sim 428 \text{ nm}$ ; BGO (purchased from Meishan Bora New Materials Co., Ltd., China): Certified light yield of  $10,000$  photons  $\text{MeV}^{-1}$ , emission peak at  $\sim 480 \text{ nm}$ .

We measured the pulse-height spectra of both LYSO:Ce and BGO under a  $^{241}\text{Am}$  (59.5 keV)  $\gamma$ -ray excitation using a photomultiplier tube (PMT, model R2059, Hamamatsu). The quantum efficiency (QE) curve of this PMT was provided by the manufacturer and is included in Supplementary Fig. 16. A high-voltage power supply (556, Ortec) was used to provide the necessary voltage to the PMT. The output signal from the PMT was read using an oscilloscope (MSO54B 5-BW-2000, Tektronix). The radioactive  $^{241}\text{Am}$ , with an activity of  $2.85 \times 10^8 \text{ Bq}$ , irradiated the crystal through a Be window from a distance of 1 cm. Both the PMT and the  $^{241}\text{Am}$  source were placed inside a completely lightproof shielding box. A total of 50,000 pulses were collected using the oscilloscope, processed into a pulse height spectrum, and fitted using a Gaussian function. The light yield of one standard was used to predict the other, following the full spectral correction methodology. The corrected light yield is calculated as:

$$\text{LY}_{\text{measured}} = \text{LY}_{\text{certified}} \cdot \frac{C_{\text{measured}}}{C_{\text{certified}}} \cdot \frac{\varphi_{\text{certified}}}{\varphi_{\text{measured}}} \quad (2)$$

where  $C$  denotes the channel number, and  $\varphi$  is the wavelength dependent detection efficiency correction constant, calculated by convolving each scintillator's emission spectrum with the PMT's QE curve:

$$\varphi_j = \frac{\int I_j(\lambda) S(\lambda) d\lambda}{\int I_j(\lambda) d\lambda} \quad (3)$$

where  $I_j(\lambda)$  represents the RL intensity of the scintillator, and  $S(\lambda)$  is the wavelength-dependent PMT detection efficiency.

Using LYSO:Ce ( $\text{LY} \sim 32,000$  photons  $\text{MeV}^{-1}$ ) to predict BGO's LY:  $\varphi_{\text{BGO}} = 0.1920$ ,  $\varphi_{\text{LYSO}} = 0.2691$ . The measured photopeak channel ratio was  $C_{\text{BGO}} / C_{\text{LYSO}} = 0.24$  (Fig. 4c). Predicted  $\text{LY}_{\text{BGO}} = 32,000 \times 0.24 \times (0.2691/0.192) = 10,764$  photons  $\text{MeV}^{-1}$ . This value is in agreement with the certified LY of BGO ( $\sim 10,000$  photons  $\text{MeV}^{-1}$ ), well within the acceptable uncertainty range for such measurements, considering factors like surface finish and coupling.

We then measured our 2D perovskite scintillator under identical conditions. The calculated  $\varphi$  values were 0.2691 for the LYSO:Ce reference, and 0.2503, 0.2611, and 0.2614 for (CMA)<sub>2</sub>PbBr<sub>4</sub>, (CMA-COOH)<sub>2</sub>PbBr<sub>4</sub>, and (1,4-CMA)PbBr<sub>4</sub>, respectively. For each sample, the light yield was calculated using the established comparative formula:

$$\text{LY}_{\text{sample}} = \text{LY}_{\text{LYSO}} \cdot \frac{C_{\text{sample}}}{C_{\text{LYSO}}} \cdot \frac{\varphi_{\text{LYSO}}}{\varphi_{\text{sample}}} \quad (4)$$

This rigorous calibration procedure, applying all necessary corrections, obtained the final light yields of 50,460, 20,460, and 19,730 photons  $\text{MeV}^{-1}$  for (CMA)<sub>2</sub>PbBr<sub>4</sub>, (CMA-COOH)<sub>2</sub>PbBr<sub>4</sub>, and (1,4-CMA)PbBr<sub>4</sub>, respectively.

### Calculation of X-ray imaging spatial resolution

The spatial resolution of the X-ray imaging system was quantitatively evaluated using the MTF based on the slanted-edge method. Sharp-edge images were acquired by imaging a copper sheet approximately 0.15 mm in thickness. The edge spread function (ESF) was obtained from the edge profile, followed by differentiation to generate the line spread function (LSF). Finally, the MTF was computed by applying a

Fourier transform to the LSF, as expressed in the equation below:

$$MTF(\nu) = F(LSF(x)) = F\left(\frac{dESF(x)}{dx}\right) \quad (5)$$

where  $\nu$  is the spatial frequency.

## Data availability

All data needed to evaluate the conclusions in the paper are present in the paper and/or the Supplementary Information or upon request from the corresponding authors. Crystallographic data for the structures reported in this Article have been deposited at the Cambridge Crystallographic Data Center, under deposition numbers CCDC 2502889, 2502891, 2502928, 2502972. Copies of the data can be obtained free of charge via <https://www.ccdc.cam.ac.uk/structures/>. Crystallographic data are also provided as Supplementary Data 1.

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## Author contributions

G.N. and M.X. conceived the idea. J.L., M.X., and G.N. designed the experiments and analyzed the data. J.L. and M.L. carried out most material design, fabrication, optimization and characterizations. P.L., R.L., and G.K. carried out the timing performance and CTR measurement and analyzed the results. J.Y. performed the theoretical simulation and analyzed the results. Z.L. and G.N. carried out light yield measurement. J.W. helped with fabrication of the single crystals. Y.X. helped with measurement of RL spectra. Q.L. helped with data analysis. M.X., G.N., and J.L. wrote the paper. All authors commented on the manuscript.

## Competing interests

The authors declare no competing interests.

## Additional information

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**Correspondence** and requests for materials should be addressed to Mengling Xia or Guangda Niu.

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