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Chemo-Divergent and Regioselective Branched Cross-Hydroalkenylation of Electron-Deficient Olefins and Vinylarenes Directed by (NHC)Ni(II) Catalysts.

Xiao Gu (谷潇)^{1-3†}, Junjie Kuang (邝俊杰)^{1-3†}, Jionghao Deng (邓炯昊)¹, Zhifeng Zhang (张志峰)¹⁻³, Junzhe Shan (单俊哲)², Man-Kin Wong (黄文健)^{4*}, Chun-Yu Ho (何振宇)^{1-3*}

¹ Guangdong Provincial Key Laboratory of Catalysis, Southern University of Science and Technology (SUSTech), Shenzhen 518055, China.

² Shenzhen Grubbs Institute, Southern University of Science and Technology (SUSTech), Shenzhen 518055, China.

³ Department of Chemistry, Southern University of Science and Technology (SUSTech), Shenzhen 518055, China.

⁴ Research Institute for Future Food, Department of Food Science and Nutrition, The Hong Kong Polytechnic University, Hong Kong Special Administration Region, China.

† Equal contributing.

* Email: mankin.wong@polyu.edu.hk; jasonhcy@sustech.edu.cn.

Abstract:

Catalytic methods for synthesizing highly substituted olefins are crucial for fine chemical and polymer production, yet alkene insertion systems featuring two highly competitive directing effects often scramble the outcome, and strategies to distinguish these effects remain underdeveloped. Here, we present a straightforward chemodivergent and regioselective cross-hydroalkenylation of acrylates and vinylarenes, enabling intermolecular access to distinct conjugated olefin isomers that were previously inaccessible from those abundant substrate classes directly. Our investigations reveal that divergence arises from two sequential, redox-neutral insertion steps, which overcome intrinsic boundaries imposed by a common olefin intermediate formation, oxidative addition, or isomerization. Unlike tactics that rely on varying prefunctionalized substrate pairs, our (NHC)Ni(II)-directed method exploits catalyst-controlled responses to polarized olefin hydrometallation and carbometallation. The green synthesis has instant utility in drug synthesis and in the preparation of electron-deficient, branched allylic centers with up to 97% e.e.. This strategy also extends to vinylBpin and vinylsiloxanes, significantly expanding the scope of divergent olefin synthesis.

Introduction

Hydrovinylation (HV)¹⁻⁴ and cross-hydroalkenylation (HA)⁵⁻¹² are fundamental transformations for the preparation of highly substituted olefins. These tools complement olefin cross-metathesis¹³, reductive cross-coupling¹⁴⁻¹⁷ and acid-catalyzed 1,4-addition of electron-deficient olefin^{18,19} by enabling atom-economical access to diverse carboskeletons and distinct olefin structures from two substrate pools. Advances employing Ru, Co, Ir, and Ni hydride catalysts with tailored ligand environments have expanded the scope²⁰⁻²⁹, delivering exceptional chemo- and regio-selectivity and high enantioselectivity (Fig. 1a–c). Besides, chain-walk processes³⁰⁻³⁷ and P ligand-mediated oxidative cross-dimerization have enabled regiodivergent olefin insertions/additions (Fig. 1d)^{38,39}. Chemodivergent cross-HA, however, remains rare^{40,41}. Systems featuring strongly competing directing effects, such as benzylic stabilization and electron-deficient substituents, frequently suffer from selectivity erosion, and effective strategies to differentiate these effects are underdeveloped (Fig. 1e)⁴². Consequently, most of the related chemodivergent syntheses are guided by a more reactive substrate, which first engages with the catalyst to form a common intermediate. This intermediate is then diverted in subsequent steps, such as oxidative cycloisomerization of 1,6-enynes, by using different ligands, olefins, or even distinct metal centers^{39,43-50} (Fig. 1f). And the current HA solutions for accessing alternate isomers typically rely on redox-active cycles involving prefunctionalized grips like alkenyl halides. This gap motivates the development of methods that can respond chemodivergently and regioselectively to hydrometallation (HM) and carbometallation (CM) preference within a given set of substrates. Here we present a [(NHC)Ni(allyl)Cl]-catalyzed (NHC = N-heterocyclic carbene) strategy that enables chemodivergent and regioselective cross-HA by reversing the donor/acceptor roles of the same substrate pairs under redox-neutral conditions (Fig. 1g, tail-to-head (t-h), tail-to-tail (t-t), and head-to-tail (h-t) cross-HA). Unlike other distinguish divergent alkene synthesis methodologies,^{40,41} our approach exploits two discrete intermolecular insertion events at the (NHC)Ni(II) center (Fig. 1g), thereby avoiding redox-active pathways and saving substrates that are vulnerable to oxidation or elimination.

Results

System development

Our investigation commenced with [(NHC)Ni(allyl)Cl] as catalyst (Fig. 2), styrene **1a** and methyl acrylate **2a** as pair (Fig. 3). We hypothesized that an optimal β -/ γ -heteroatom interactions with the (NHC)Ni(II) catalyst could override benzylic stabilization effects, thereby modulating HM chemo- and regio-selectivity (branched HM of **1a** vs. linear HM of **2a**, *b*-/*l*-HM). Next, by either aligning or opposing the **2a** polarization direction with sterically small or bulky NHCs, we anticipate a switch in substrate insertion chemoselectivity during CM step, favoring a high cross/homo-selectivity (Fig. 1g, path i-iii).

Systematic NHC screening identified two distinct catalyst classes: (i) bulky and symmetric NHCs (**L1–L5**) favored chemo- and regioselective *l*-HM of acrylate **2a** over styrene **1a**, affording homoallyl ester **3aa** with progressively higher *l*-over-*b*-HM selectivity (**L1–L5**, as reflected by **3:4** from 55:45 to >95:5); and (ii) sterically adaptive cores (**L6–L7**) or less-substituted NHCs (**L8–L9**) restored benzylic stabilization-directed *b*-HM of styrene (**L6–L11**), affording γ -branched α,β -unsaturated ester **5aa** (Fig. 3, see Fig. 2 for homo-HA products^{51,52} and NHC structures). This selectivity swap correlates primarily with changes in NHC percentage buried volume (% V_{bur}). For 5-membered ring NHCs, % V_{bur} values derived from (NHC)AuCl models^{53–55} at 2.00 Å show that **L1–L5** span 36.5–49.0%, whereas **L8–L9** fall within 34.5–27.5% range, showing that steric effects dominate (as **L1–L5/L8–L9** share similar electronic profile, A_1 CO \approx 2051 cm⁻¹ for [(NHC)Ni(CO)₃])⁵⁶.

Ring-expanded NHCs **L6–L7** exhibit even higher % V_{bur} values but reduced conformational rigidity due to enhanced internal rotation and lower symmetry^{57–59}, providing an alternative way to favor formation of **5**. A third strategy to promote high selectivity for **5** involves modifying the *N*-aryl substitution pattern. Chiral NHCs **L10–L11**, despite their steric bulk, offer open quadrants that relieve steric repulsions in the *b*-HM of **1a** and in *l*-CM of **2a**. This design enabled efficient synthesis of (*R*)-**5aa** using the C₂ symmetric chiral NHC **L11** bearing a single *ortho*-substituted *N*-aryl, analogous to **L9** (>95% yield, 97% e.e. by chiral HPLC). The absolute configuration was assigned by comparison with reported optical rotation (see SI for details). Moreover, regioselectivity of olefins on **3–6** was confirmed by comparison with authentic samples using crude ¹H NMR analysis, and no additional olefin isomers were detected. This one-step cross-HA therefore streamlines access to motifs in many pharmacologically important scaffolds, like Enterocin⁶⁰, Baccharisketone⁶¹, Curcumene, Lacvigatin, ar-Himachalene, and Erogorgiaene^{62–66}. It also replaces multistep routes based on Claisen rearrangement of branched allyl alcohol⁶⁷, allylic substitution-

olefin metathesis from acrylates⁶⁸, and convergent allylic substitution from alkynoates⁶⁶. Importantly, the neutral conditions suppress the acid-catalyzed linear-dimerization of **1a** and base-promoted MBH-like reactivity of **2a**⁶⁹ (e.g. by P and N ligand dissociation related organocatalysis). In short, this NHC-substrate cooperation shifts the paradigm away from the traditional t-t and t-h selectivities prevalent in Ni/Pd chemistry, as well as the formal homo-head-to-head (h-h) and t-h selectivities in acidic and basic conditions. It enables access to previously unexplored h-t and alternate t-h pathways and delivers mutually exclusive product sets: no **5aa/6aa** by **L1-L5**, and no **3aa/4aa** by **L6-L9** and **L11**, representing a divergent shift from one to another out of 16 isomers.

Scope exploration

Next, the two highly chemo- and regio-selective HM strategies of acrylates and styrenes were applied for expanding the h-t and t-h cross-HA scope and applications (Fig. 4a-b). First, **2a** remained as a highly competent chemo- and regio-selective HM director in the presence of other compelling effects like steric bias and hemilabile groups under the actions of the **L4-L5** catalysts, accommodating aryl- (Fig. 4a1, electronic activated, fused rings), alkyl- (Fig. 4a2, linear and branched), hetero- (Fig. 4a2, O/N-allyl and homoallyl), Bpin- (Bpin = pinacol boronate group)(Fig. 4a3, allyl and vinyl) substituted olefins, all furnishing **3** in good yield and high selectivity (**3:4** up to 94:6, no **5** detected by NMR). Similarly, other acrylates (**2b-2e**) were tolerated, producing branched γ,δ -unsaturated esters (Fig. 4a3). Hence, this h-t HA directly enables both aromatic and aliphatic terminal olefins⁷⁰ for the analogous conjugated additions of 2-alkenylmetals to α,β -unsaturated esters, free it from the risk to nucleophilic substitutions and redox-active side reactions at allylic position. On the other hand, the sterically adaptive **L11**-directed chemo- and enantio-selective *b*-HM strategy maintained at high levels in general for a broad range of aromatic acceptors (Fig. 4b, **1a-1h**, *E/Z-5* ~9:1, **5:3** >95:5, 90-97% e.e.). By using **L7/L9**, some of the racemic products preparation were achieved with 5 mol% cat. loading in a shorter reaction time. Hence, this HA also enables us to prepare an electron-deficient branched allylic carbon center from two abundant monoenes in polymer industry directly, which is valuable as it avoids complications in other routes like further 1,2-/1,4-addition in typical allylic substitutions.

Discussion

Although the above chemodivergent and regioselective cross-HA hypothesis aligns well with our observed result, a more detailed investigation was conducted since several conceivable mechanisms like oxidative cyclization^{46,71-73} and β -CH insertion⁹ are not impossible when using electron-deficient olefins as one of the substrates (Fig. 5, see SI for procedure). First, we inspected redox-active catalytic cycles involving (NHC)Ni(0) as alternate mechanism (Fig. 5a–b)⁷⁴⁻⁷⁶. Yet, experiments using the corresponding NHC/Ni(cod)₂ catalysts failed to deliver the desired products and instead resulted in nonselective consumption (Fig. 5c). This outcome is consistent with literature precedents where (NHC)Ni(0)-catalyzed conjugate additions of α -olefins to reactive α,β -unsaturated aldehydes require TESOTf as both a reactivity-directing substrate and carbonyl activator at 45 °C⁷⁷. These comparisons indicate that our HA is unlikely to proceed via (NHC)Ni(0)-mediated pathways such as nickelacyclopentane intermediates (Fig. 5a, regioselective β -H eliminations)⁷¹⁻⁷³ or ligand-directed mechanistic shifts (Fig. 5b, change from redox-neutral insertion by **L4/L5**-Ni(II) for **3** to redox-active β -CH oxidative addition by **L7/L9**-Ni(0) for **5**)^{9,24,78}. To rule out the redox-active pathways further, [L4-Ni(I)Cl]₂ was independently evaluated as a catalyst. Under otherwise identical conditions, no desired product was detected (Fig. 5d). Also, the catalytic performance remained unchanged even it was challenged by three different and highly competitive alkenes prone to radical paths (Fig. 5e, MHAT)^{14,15,17,79}.

Although we are unable to trap the hydrometallated species like the cross-HA of cyclopropenes by our catalyst⁸⁰, the use of d₈-**1a** reveals additional features of our reaction. In the catalytic synthesis of d_n-**E-5aa** by **L9**, over 33% of H (38%-40%) was incorporated at the Me position, and no deuterium was detected on carbon centers derived from **2a** by NMR (Fig. 5f). These data indicate that HM of d₈-**1a** is reversible with the **L9** catalyst and that the drop of D content at the methyl position partly supports a stepwise insertion mechanism (**1a:2a** = 1:1.5). In a separate homo-HA of **2a**, 39% conversion was observed and 28% **2a** was converted to h-h dimer within 10 mins, which may explain the D loss at the methyl position of the labeling study. In contrast, in a d_n-**3aa** preparation at a higher **L4** catalyst loading, a considerable amount of product bearing a CH₂ at the ester α -position and a gem-olefin at a low H incorporation was obtained (Fig. 5g). Hence, even though the **1a** HM is known to be facile and reversible⁸¹, and despite d₈-**1a** was used in excess (d₈-**1a:2a** = 2:1), the HM of **2a** remains competitive and limits thorough Ni-H/D scrambling. Despite the Ni-H converts to Ni-D gradually, still no d_n-**2a** and no CD₂ at the ester α -position was detected at the end (CDH:CD₂ >95:5). These observations indicate that the hydrometallated **2a** rarely undergoes the backward reaction, possibly due to heteroatom stabilization as anticipated. Collectively, these results support that our cross-HA operates via a redox-neutral, divergent

synthetic approach complementing, rather than overlapping with, existing redox-active pathways with acrylate derivatives.

Next, we inspected whether the selectivity might arise from (NHC)Ni(II) directed regioselective isomerization^{82,83}, as **3** and **5** are a pair of isomers sharing the same carboskeleton (Fig. 6, see SI for procedure). However, neither interconversion among **3aa** and *E*-**5aa** (Fig. 6a-b, by switching between the two sets of NHC catalysts used in cross-HA) nor isomerization of the independently synthesized trisubstituted olefin *i*-**5aa** (Fig. 6c) occurred under catalytic conditions, even under prolonged reaction times and elevated temperatures. Indeed, only < 8% *Z*-**5aa** was obtained from heating a mixture of *E*-**5aa** and **2a** at 80 °C for 24 h, suggesting the catalyst suppresses further insertion and limits the HM of **5aa** to just favor Ni at its α -position (Fig. 6d). These findings demonstrate that, unlike other regiodivergent strategies by isomerization, no common olefin intermediate is involved.

Further mechanistic support was obtained from experiments with electron-deficient olefins of related structural features (Fig. 7). First, olefins bearing hemilabile coordinating groups that mimic the carbonyl and polarization direction of acrylates, like vinylsiloxanes **1q-r** were examined (deduced by ¹³C NMR chemical shift of the two olefinic carbons, see SI for details). As anticipated, their corresponding cross-HA products **3** and **5** with styrene were formed likewise using **L4** and **L7** [Fig. 7a1) h-t and a2) t-h], and thus our strategy is not limited to olefins bearing C=O, and Si-OR is also usable. This reactivity is notable since the NHC catalyst using OTf as anion failed to provide the desired reactivity before.⁸⁴ Yet, a reduced **3:4** selectivity from >95:5 to 84:16 was noted (as a reflection of *l/b*-HM ratio) when replacing the Si(OiPr)₃ with a SiMe₂Ph (Fig. 7a1, **1r** vs. **1s**, see SI for the reversed polarization⁸⁴), approving the oxy-group and the olefin polarization direction assist the *l*-HM of electron-deficient olefin. Besides, the use of bulkier vinylSi(OR)₃ assists productivity (**3qa** vs. **3ra**; **5aq** vs. **5ar**), while the use of bulkier NHC was able to restore the high **3:4** selectivity in vinylSiMe₂Ph (**L4** vs **L5**) but came at the cost of reduced reactivity (**3sa**), highlighting the need for optimal steric repulsion.

Interestingly, when preparing **5** with vinylsiloxane/silane substrates using **L7** catalyst, a detailed analysis revealed dissimilar predominant undesired substrate consumption pathways (Fig. 7a2). Nonselective styrene consumption was prevalent in vinylsilane cases (**1s-t**), while vinylsiloxanes (**1q-r**) homo-HA was the main side reaction, occurring at higher levels (up to 7% vs. <5%, and approximately 17% under forcing conditions without styrene). These findings elucidated the dual role of oxy-substituents in our divergent HA: while potentially compromising styrene's HM chemoselectivity, they facilitate chemo-

and regioselective CM through a deshielded β -C olefin site, just like acrylate. Therefore, the improved cross-/homo-selectivity observed with vinylsiloxane **1r** over **1q** is attributed partly to restored styrene HM via enhanced steric repulsion. Merely employing a flexible NHC structure without concerning the competition risks generating a mixture of styrene and vinylsiloxane homo-products. Besides, the marked productivity difference between vinylSiMe₂Ph **1s** and vinylTMS **1t** (yield of **5as** >> **5at**) offers compelling evidence that the use of more deshielded β -C-position - rather than steric repulsion minimization with **L7** - as the key determinant of CM order (Fig. 7a2). Overall, the divergent HA streamlines the alkenyl and homoallyl silane synthetic routes starting from 1-silylcyclopropenes by B-H insertion⁸⁵ and from allylic silanes by anti-S_N2' allylic substitution⁸⁶, and this comparative study allowed us to underscore the key aspects for distinguishing the hemilabile coordination⁸⁷⁻⁹¹ and benzylic stabilization effectively by (NHC)Ni(II) catalysts.

Based on the above refined mechanistic insights, vinylBPin was chosen as the cross-HA partner with styrene, owing to its hemilabile oxy-coordination ability and steric bulk, despite its relatively low olefin polarization (Fig. 7b). As expected, the bulkier **L5** and the flexible **L7** catalysts delivered the target cross-HA products **3pa** and **5ap** as the major products at a slightly elevated temperature (Fig. 7b1, 69% yield, **3:4** > 95:5) and 7.5 mol% catalyst at r.t. (Fig. 7b3, 69-80% yield, **5:6** >95:5). Moreover, high e.e. of **5ap** is also achievable by using **L11** (88% e.e.)⁹². Notably, a double-branched t-t cross-HA product **6** was then observed by medium-sized NHCs (Fig. 7b2, **L3** and **L4**, up to 86% yield and 97% selectivity). These findings emphasize the dominant role of NHC steric effects in governing cross-HA product distribution, particularly with less polarized and weakly coordinating electron-deficient olefins. While chemodivergent HM of such combinations becomes more challenging, regiodivergent cross-CM can be modulated by strategic NHC ligand design - **L3/L4** favor the double-branched t-t product **6ap** over **4pa**, whereas **L7/L9** selectively yield the single-branched t-h product **5ap**.

Integrating our findings with relevant HA literature by NHC¹², we propose a (NHC)Ni(II)-directed, redox-neutral, insertion-based catalytic cross-HA cycle to account for the observations (Fig. 8). In the upper cycle (Fig. 8a), bulky NHC catalysts with optimal oxy-coordinating ability favor *l*-HM of olefin acceptor, followed by chemo- and regio-selective CM of styrene or α -olefin to deliver h-t product **3** (α -olefin shares similar polarization direction with styrene rather than acrylate). In the bottom cycle (Fig. 8b), flexible NHCs favor *b*-HM of styrene, after which chemoselective and regiodivergent CM yielding t-h product **5** or t-t product **6**. Minimizing steric repulsions among substituents on the C₂ symmetric chiral (*R,R*)-NHC and styrene directs the enantioselective *b*-HM. The open quadrant on the **L11** then allows the

chemo- and regio-selective CM of **2a** by lowering the concerned steric repulsion and matching the electronic demand in t-h isomer formation, yielding (*R*)-**5aa** (Fig. 9)^{68,93}. Notably, unlike the steric effect directed styrene- α -olefin t-t cross-HA^{51,81,94}, the donor electronic effect is a crucial factor in controlling CM step in both cycles for high cross/homo-selectivity when pairing a suitable NHC (Fig. 10). It favors **3** and **5**, despite the homo-h-t of acrylate in (a) and homo-t-h of styrene in (b) are sterically comparable with the desired paths (see the dotted lines). In addition, this model also accounts for the differences among vinyl-silane and -siloxane behaviors (c-d), since they shared similar electronic characters with styrene and acrylate, respectively. Overall, this modular cross-HA system with twofold divergence (achieving chemoselective-HM and regioselective CM) efficiently delivers up to three distinct carboskeletons, each bearing a Bpin, SiR₃, or ester group alongside an alkenyl moiety.

In summary, this study unlocks one of the most straightforward strategies for highly chemodivergent and regioselective branched cross-HA of styrenes and electron-deficient olefins (controllable from 1 to another, out of 16 possible isomers). Our NHC catalyst pairs fundamentally enable the modular insertions through their trenchant responses to the hemilabile coordination, benzylic stabilization, and polarized olefin, facilitating branch-selective cross-HA that overcomes the intrinsic boundaries associated with strategies based on acid/base-directed organocatalytic paths (or from P/N-ligand dissociation), shared intermediate formation, isomerization, alkenyl halides, or steric repulsions alone. This work also represents a prominent expansion of olefin donor scope in (NHC)Ni(II)-directed cross-HA from α -olefins to acrylate derivatives, and it demonstrates that the donor polarization direction serves as a valuable tool beyond merely minimizing steric repulsion in CM step for high cross/homo-selectivity. Besides, the highly chemoselective acrylate *l*-HM strategy extends to h-t cross-HA with α -olefins, while the styrene *b*-HM strategy enables electron-deficient chiral allylic carbon center synthesis (up to 97% e.e.) without over additions. Overall, this system establishes a sustainable, modular platform for the divergent synthesis of conjugated olefins from abundant monoenes, and it underscores the need to develop NHC catalysts capable of discriminating subtle yet critical HM and CM determinants.

Methods

General procedure of the cross-HA. In a glove box, substrates **1** and **2** (indicated amount) were added to the in situ generated [(NHC)Ni(allyl)]NaBARF catalyst solution from [(NHC)Ni(allyl)Cl]/NaBARF/1-octene (0.025 mmol, Ni:NHC:NaBARF:1-octene = 1:1:1:2). The mixture was stirred at r.t. for 1-12 h (see SI for details). After work-up, the yield, structure, and selectivity of the desired product were determined by

NMR, HPLC, HRMS, and isolation. Full procedures and characterization data are provided in the Supplementary Information.

Data availability

All data supporting the findings of this study are available within the article and Supplementary Information. Data supporting the findings of this manuscript are also available from the corresponding author upon request.

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styrene. Unfortunately, we are unable to determine the corresponding product e.e. accurately in
our hands so far.
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Author contributions

X.G. and J.-J. K performed experimental studies and analysis. J.-H. D conducted part of the racemic **5** preparation. J.-Z. S helped the substrates preparation and products purification. Z.-F. Z conducted synthesis of chiral ligands. M.-K. W. and C.-Y. H. collaborated the project and supervised the work.

Competing interests

The authors declare no competing interests.

Figure legends

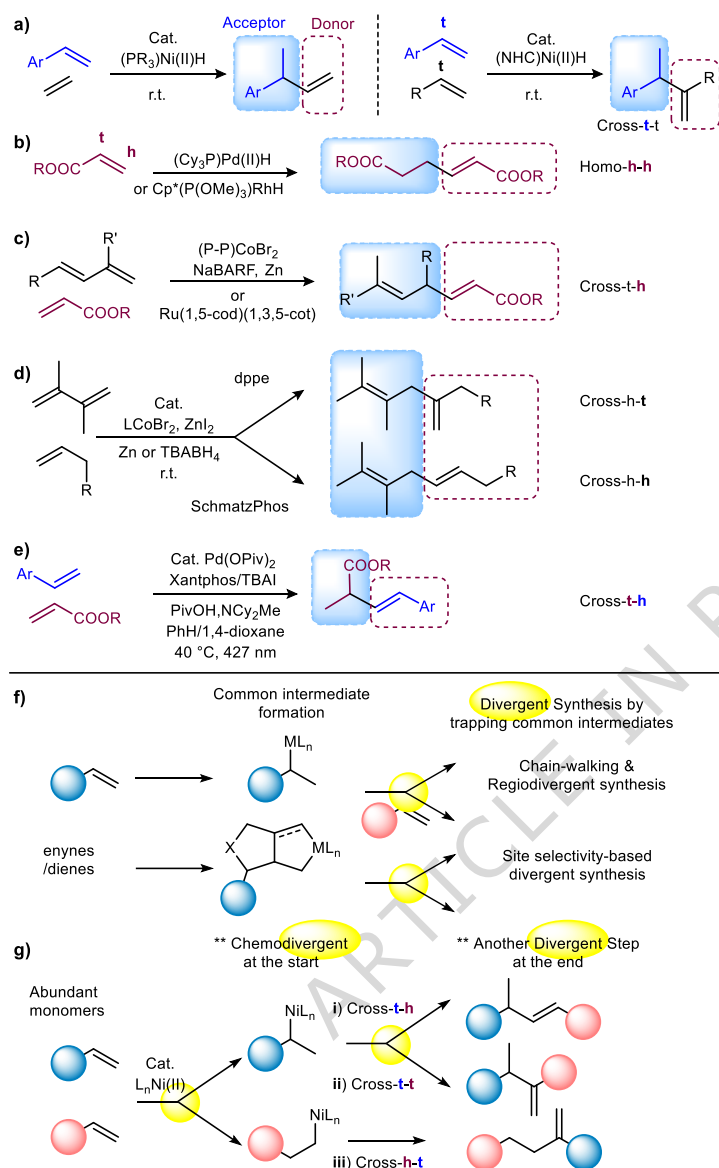


Fig. 1. Catalytic cross-hydroalkenylation: from regio- and chemo-selective to divergent. **a)** Hydrovinylation and hydroalkenylation of styrene. **b)** Homo-hydroalkenylation of methyl acrylate. **c)** Heterodimerization of methyl acrylate & 1,3-diene (oxidative cyclization). **d)** Regio-divergent cross-dimerization (trapping the same intermediate). **e)** Pd-catalyzed hydroalkenylation of acrylate and styrene. **f)** Conventional divergent alkenylation. **g)** This work: chemodivergent and regioselective cross-hydroalkenylation.

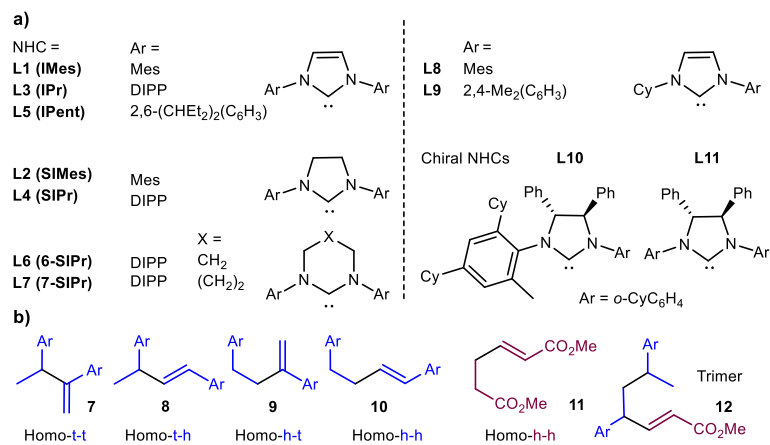


Fig. 2. Structures of NHCs, homo-hydroalkenylation and trimerization products in this work. Mes = mesityl. DIPP = diisopropylphenyl. a) Structures of NHCs. b) Structures of homo-products.

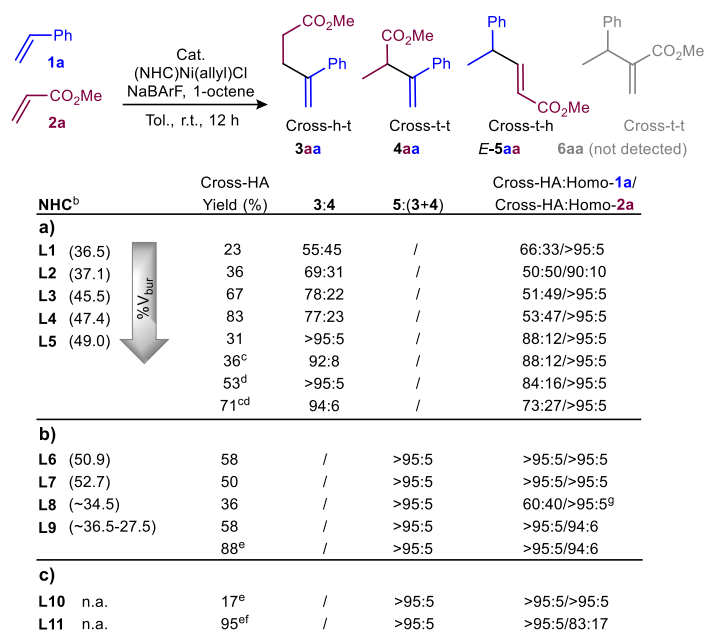


Fig. 3. Screening of NHC for chemodivergent cross-hydroalkenylation of 1a and 2a.^a n.a. = not available.

a) Bulky and symmetric NHCs. **b)** sterically adaptive cores or less-substituted NHCs. **c)** Chiral NHCs. **a** The reactions were performed by following the procedure: **1a** and **2a** (2: 1, 0.25 mmol) were added to 10 mol% [(NHC)Ni(allyl)Cl]/NaBARf (NaBARf = Sodium tetrakis[3,5-bis(trifluoromethyl)phenyl]borate) catalyst in 2 mL toluene with 1-octene (20 mol%) and stirred at r.t. for 12 h. Selectivity and yield were determined by crude ¹H NMR using CH₃NO₂ as standard. The desired cross-HA product structures, **3aa**, **4aa** and **5aa**, were confirmed by isolation, and no **6aa** was observed. The major homo-HA of **1a** and **2a** were determined as **7aa**, **8aa**, and **11aa**, and no **9aa** and **10aa** were observed (see Fig. 2 for structures). *E/Z*-**5aa** ~ 91:9. The substrate nonselective consumption was presumably caused by oligomerization. **b** Value in parenthesis is %V_{bur} for (NHC)AuCl at length of 2.00 Å. **L8** and **L9** are estimated by closely related NHCs. **L8** by a N-pinanyl substituted NHC; **L9** by the data from ICy/IMes. **c** 50 °C. **d** **1a**: **2a** = 3: 1. **e** 2.5 mol% of cat., **1a**:**2a** = 1:1.5, 1 h. **f** 97% e.e., determined by chiral HPLC ((*R*)-product, OD-H column, see SI for details). **g** < 5% of trimer **12aaa** was noted by NMR in this screening.

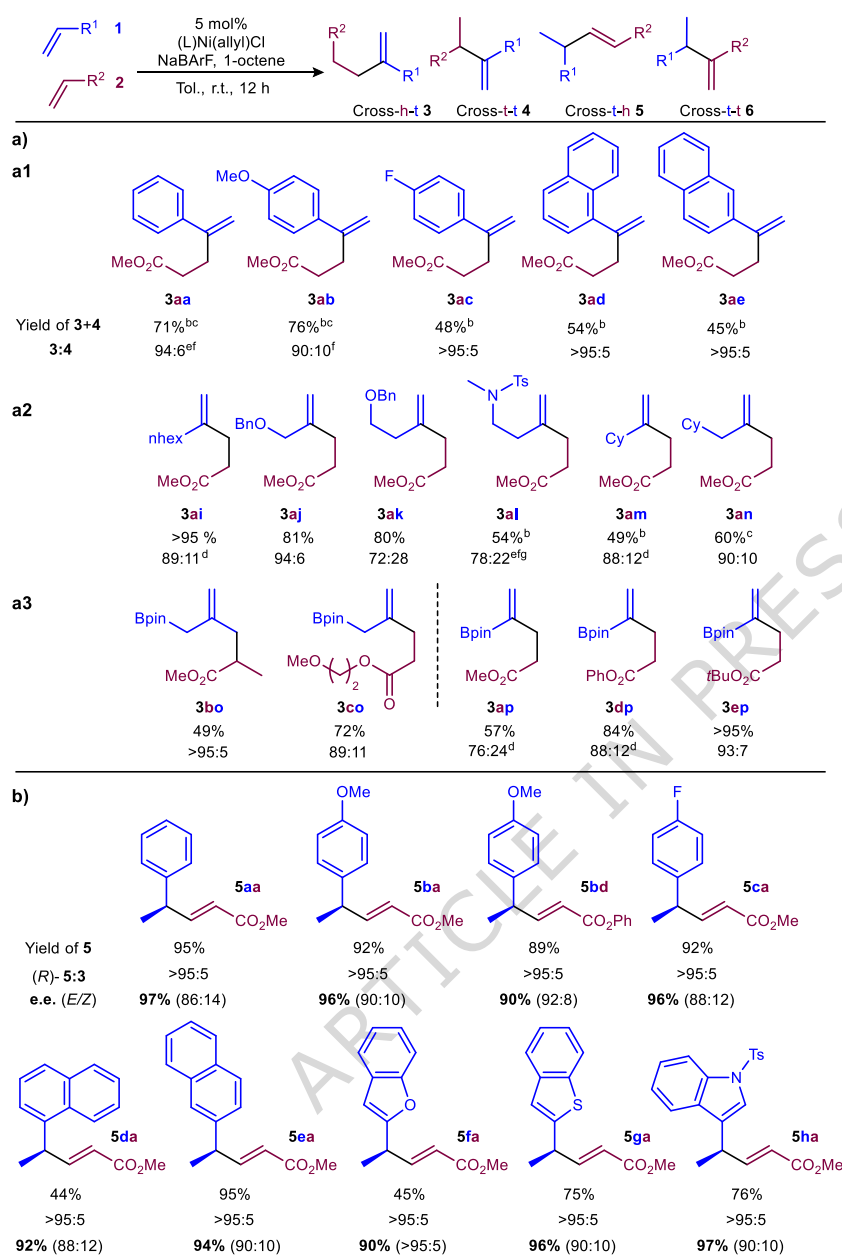


Fig. 4. Scope and applications of the chemo- and regio-selective hydrometallation.^a **a)** Acrylate as acceptor; L = **L4**; **1:2a** = 2:1; **(3+4):5** > 95:5; cross-HA:homo-**2a** > 95:5. **a1** R¹ = aryls; **2a** R² = COOMe. **a2** R¹ = alkyls; **2a** R² = COOMe. **a3** Bpin substituted **1o** & **1p**. **b)** Styrene as acceptor; L = **L11**^b; **1:2** = 1:1.5, **3** & **4** & **6** < 5%; cross-HA:homo-**1** > 95:5. **a** The reactions were performed by following the procedure: Indicated substrate ratio was added to 5 mol% [(NHC)Ni(allyl)Cl]/NaBARf in 2 mL toluene with 1-octene (20 mol%) and stirred at r.t. for 12 h. Selectivity and yield were determined by crude ¹H NMR using CH₃NO₂ as standard like Fig. 3, the desired cross-HA product structures were confirmed by isolation, the amount of minor product **4** was estimated by the characteristic ¹H NMR peaks of **4aa** and related examples reported in the literature (See SI). The e.e. of *E*-**5** was determined by chiral HPLC and (*R*)-configuration were assigned by comparison with the optical rotation of **3aa** reported by other methods. **b** 10 mol% catalyst. **c** NHC = **L5**. **d, e** 35 °C and 50 °C. **f** **1:2a** = 3:1. **g** 0.125 mmol substrate.

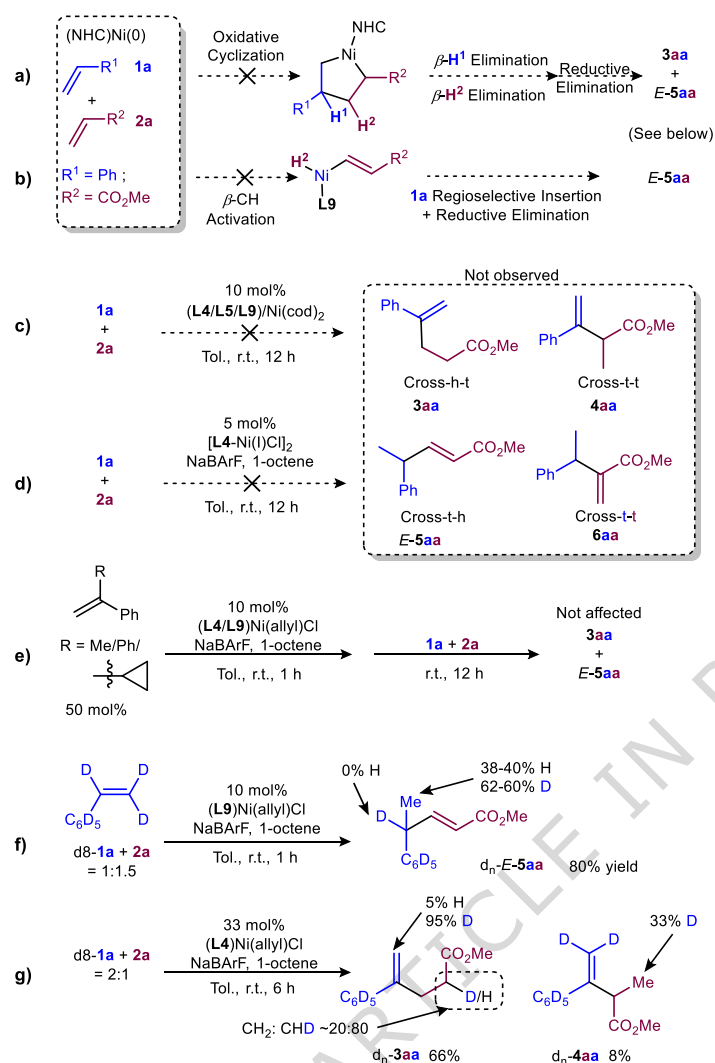


Fig. 5. Other underdeveloped pathways and mechanistic studies for the catalytic cross-HA. a) Oxidative cyclization pathway. **b)** β -CH activation pathway. **c)** Attempts of cross-HA by using Ni(0) catalyst. **d)** Attempt of cross-HA by using Ni(I) catalyst. **e)** Attempts of cross-HA with 1-substituted styrene as additives. **f)** D-labeled experiment by using L9. **g)** D-labeled experiment by using L4.

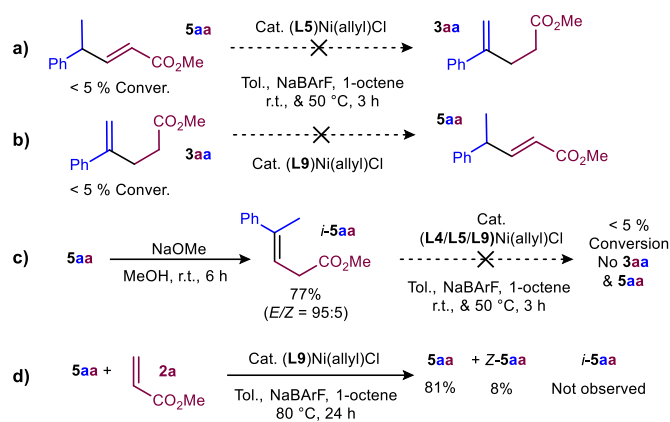


Fig. 6. Control experiments for the selective olefin formation. a) Isomerization attempt of **5aa** to **3aa** by using **L5**. **b)** Isomerization attempt of **3aa** to **5aa** by using **L9**. **c)** Preparation of *i*-**5aa** and the isomerization attempts of *i*-**5aa**. **d)** Isomerization attempt of **5aa** under prolonged reaction times and elevated temperature by using **L9**.

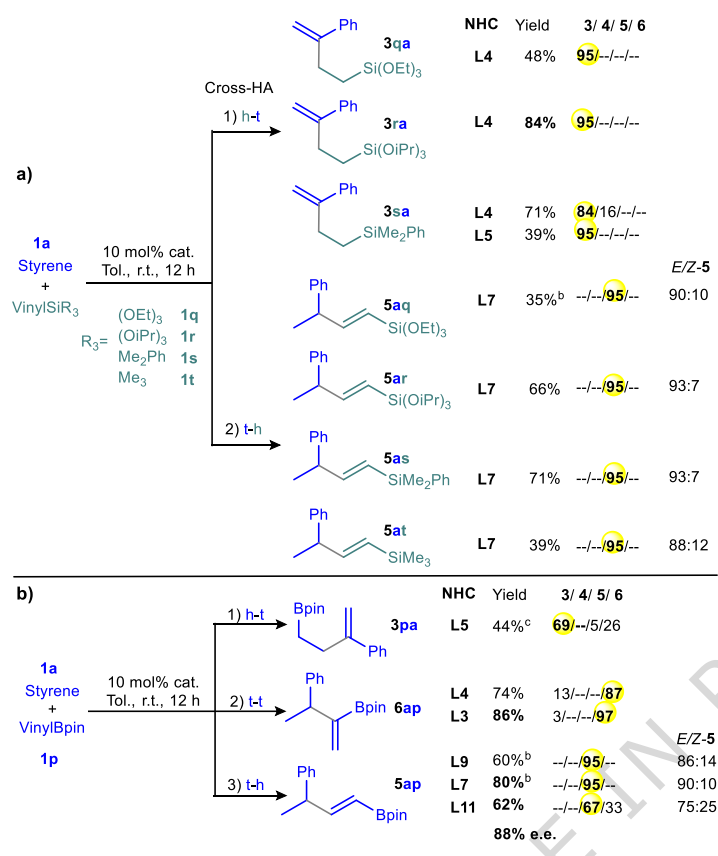


Fig. 7. Vinylsiloxanes and vinylBpin as acrylate equivalents in chemodivergent cross-HA.^a **a)** Vinylsiloxanes and vinylsilanes as acrylate equivalents. **b)** VinylBpin as acrylate equivalents. **a** The reactions were performed by following the procedure: Indicated substrate ratio was added to 10 mol% [(NHC)Ni(allyl)Cl]/NaBARF in 2 mL toluene with 1-octene (20 mol%) and stirred at r.t. for 12 h. Selectivity and yield were determined by crude ¹H NMR using CH₃NO₂ as standard like Fig. 3, the desired cross-HA product structures were confirmed by isolation, the amount of minor product **4** was estimated by the characteristic ¹H NMR peaks of **4aa** and related examples reported in the literature (See SI). The e.e. of product *E*-**5** was determined by chiral HPLC and (*R*)-configuration were assigned by comparison with the optical rotation of **3aa** reported in the literature. **b** 7.5 mol% cataly. **c** 35 °C.

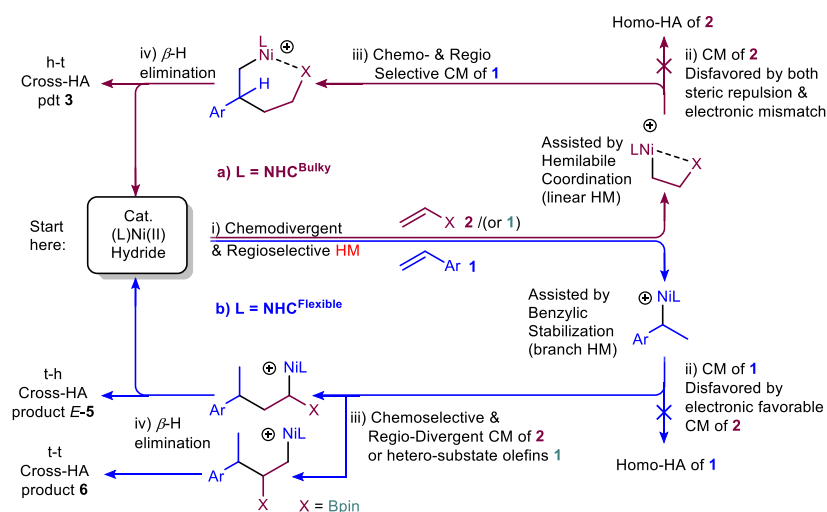


Fig. 8. Proposed mechanism for the (NHC)Ni(II) directed divergent cross-HA. HM = hydrometallation. CM = carbometallation. **a)** Bulky NHC assisted by hemilabile coordination. **b)** Flexible NHC assisted by benzylic stabilization.

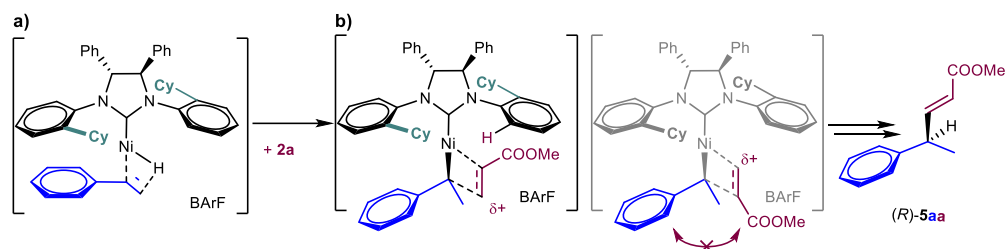


Fig. 9. Proposed model for the enantioselective preparation of (R)-5aa by L11. a) Minimizing undesired steric repulsion among styrene & (R,R)-C₂ NHC *o*-Cy. b) Open quadrant assisted regioselective CM of MA.

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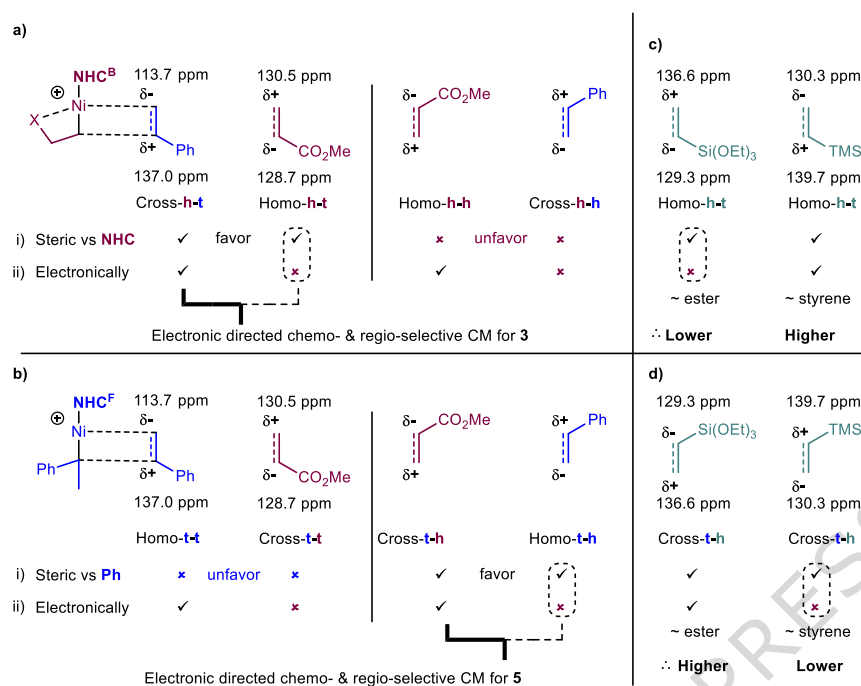


Fig. 10. Proposed models for the donor polarization effect on CM selectivity. a) CM step in Fig. 8a for **3**. **b)** CM step in Fig. 8b for **5**. **c)** rationale for different homo-h-t levels. **d)** rationale for different cross-t-h levels.

The authors present a straightforward chemodivergent and regioselective cross-hydroalkenylation of acrylates and vinylarenes, enabling intermolecular access to distinct conjugated olefin isomers that were previously inaccessible from those abundant substrate classes directly.

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