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Low contribution of oxic methane production in shallow productive lakes

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Whereas the occurrence of oxic methane (CH_4) production (OMP) in the oxygenated water column of lakes is widely accepted, its mechanisms, isotopic signature, and contribution to total CH_4 emissions remain uncertain. Evidence suggests that phytoplankton produces CH_4 , but it is unclear how this pathway contributes to ecosystem OMP rates. Shallow lakes are often productive and feature high phytoplankton biomass, which could potentially lead to high OMP rates and a substantial contribution to CH_4 emissions. Here we present results of a field mesocosm study carried out in three shallow lakes in the Pampean Plain (Argentina), designed to assess their ambient OMP dynamics. We combined this with laboratory experiments designed to estimate the potential CH_4 production by phytoplankton strains from these systems. We demonstrate that OMP occurred in all lakes, albeit at low rates; all tested phytoplankton strains produced CH_4 , yet this production contributed up to 15% to OMP rates, implying that other pathways dominate the observed OMP. The contribution of OMP to lake CH_4 diffusive emissions was low for all lakes and likely influenced by lake morphometry, suggesting that, despite their high phytoplankton abundances, other sources—such as sediment CH_4 production and/or lateral inputs—dominate CH_4 emissions in these ecosystems.

The traditional understanding of methane (CH_4) cycling in aquatic ecosystems considers that biological CH_4 is solely produced under anoxic conditions by methanogenic archaea¹. However, the frequent supersaturation of CH_4 that is observed in oxic surface waters of aquatic ecosystems cannot be explained solely by transport of CH_4 from anoxic sediments and deeper water layers^{2–5}, generating what has been termed the “methane paradox”. Over the last decade there have been numerous reports of CH_4 production in the oxic water column of aquatic ecosystems through various mechanisms, both under oxic and anoxic conditions^{6,7}. These newly identified pathways of CH_4 production are generically referred to as Oxic Methane Production (OMP), considering that they occur in oxygenated habitats, such as the oxic portion of the water column, but without necessarily implying that these specific pathways require oxygen to occur⁸. Collectively, these studies have demonstrated that there is no actual paradox but rather that the pathways of aquatic CH_4 production are more diverse and complex than previously thought^{5,9}.

There are several known metabolic pathways, in addition to archaeal methanogenesis, which produce CH_4 . There are reports of aerobic production of CH_4 as a byproduct of methyl-phosphonates (MPn) decomposition by aerobic heterotrophs in marine^{2,10,11} and freshwater environments^{12–14}. Similarly, aerobic demethylation of dimethyl sulfoniopropionate (DMSP) has been reported to produce methanethiol in marine waters, with the subsequent release of CH_4 ³. Aerobic metabolism of methylamine (MeA) has been also reported as a source of methane in lakes^{6,15} and it has even been hypothesized that all living cells can produce CH_4 by a common mechanism triggered by free iron and reactive oxygen species (ROS)¹⁶. There is also growing evidence for a coupling between OMP and phytoplankton^{17,18}. Grossart et al.^{4,19} detected methanogenic archaea in oxic waters of a lake in Germany, which were attached to phytoplankton and possibly living in micro-anoxic niches associated

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with algal cells. Moreover, several reports indicated a link between OMP and photosynthesis at an ecosystem scale^{20–23}. In this regard, it has been experimentally shown that various phytoplanktonic groups including diatoms²¹, cyanobacteria^{24,25}, chlorophytes²², cryptophytes²², haptophytes and marine microalgal species^{25–27} produce CH₄, and that the rate of production is somehow linked to temperature and light exposure^{24,28}. All of these results reflect that there appear to be multiple coexisting OMP pathways in freshwaters and these probably vary in relative importance among aquatic ecosystems, along trophic and other environmental gradients^{7,29,30}. Regardless of the mechanisms behind OMP, there is still much uncertainty as to the magnitude of the rates of OMP at the ecosystem scale and the contribution of these pathways to freshwater CH₄ emissions. There have been various attempts to address these questions, based on whole-lake^{22,29,31,32} or mesocosm²⁰ mass balances, and also based on experimental incubations of lake water^{4,5,7,29,31}. Reported ecosystem OMP rates vary from 0.01 $\mu\text{M day}^{-1}$ up to 0.52 $\mu\text{M day}^{-1}$ ^{15,20,22,29,31,32}. The studies that have quantified the contribution of OMP to total lake CH₄ production, or to total lake CH₄ emissions in the surface mixed layer of stratified lakes, have reported a wide range of values, from < 5% to up to ~80%^{29,31–33}. This is in part related to core morphometric features of freshwater ecosystems, with the contribution of OMP increasing with decreasing sediment area to volume ratio^{31,32}. Overall, as suggested by the contrasting results reported in the studies cited above, the factors that regulate OMP rates and the contribution of these pathways to total ecosystem CH₄ emissions are still not well understood.

OMP pathways also contribute to the observed isotopic CH₄ signature in the water column, and therefore to the processes that are inferred from these. The $\delta^{13}\text{C-CH}_4$ in the water column and in the sediments has been used to assess the extent of CH₄ oxidation, where the source has traditionally been assumed to be one of the two main anoxic methanogenic pathways which typically yield very depleted CH₄ (– 65‰ to – 110‰³⁴). There is increasing evidence that $\delta^{13}\text{C-CH}_4$ generated by the various OMP pathways is highly variable (– 19‰ to – 63‰) but generally more enriched than $\delta^{13}\text{C-CH}_4$ generated by archaeal methanogenesis^{25,30,32,35}. Since OMP pathways generate enriched $\delta^{13}\text{C-CH}_4$ that overlaps with the signature of oxidized methanogenic CH₄, the existence of OMP complexifies CH₄ isotopic mass balances, and it is therefore important to better assess CH₄ lake dynamics.

OMP rates and their contribution to ecosystem CH₄ emissions have been mostly explored in oligo- to mesotrophic lakes that tend to stratify, and there has been very little work done on shallow polymictic (that frequently mix) lakes²³. These lakes tend to be productive and to develop high phytoplankton biomass^{36–38}, and for this reason it could be expected that the rates of OMP might be high, yet the contribution of OMP to total CH₄ diffusive fluxes may still be modest given the importance of sediments in these shallow systems. In addition, the phytoplankton communities of shallow lakes may be dominated by very different taxa³⁹, which could potentially lead to differences in ambient OMP and in the potential values of $\delta^{13}\text{C-CH}_4$ derived from OMP as well. To test these contrasting hypotheses, we present an integrative study that combines ecosystem, mesocosm and in vitro approaches to assess the magnitude and the ecosystem-level contribution of OMP, as well as the potential contribution of phytoplankton to this process, in three shallow lakes with different abundance and composition of phytoplanktonic communities. In situ mesocosm experiments were carried out in each lake to quantify field OMP rates and to assess the potential values of $\delta^{13}\text{C-CH}_4$ derived from OMP. In addition, sampling of the lakes allowed extrapolation of the mesocosm results to determine the potential contribution of OMP to whole lake CH₄ emissions. Finally, phytoplankton strains were isolated from each one of these lakes and used to carry out in vitro experiments to assess their potential CH₄ production rates, which were subsequently used to infer the potential contribution of phytoplankton to ambient OMP in these lakes.

Methods

Study area

The Pampean Plain (35°32'–36°48'S; 57°47'–58°07'W) is a 600,000 km² lowland in central Argentina. Its low slope, geomorphology, and climate create a hydrological system with diffuse catchments, poorly developed drainage, and shallow aquifers, leading to thousands of shallow lakes⁴⁰. About 13,800 lakes exceed 10 ha, and 146,000 are between 0.05 ha and 10 ha⁴¹. These lakes are shallow, polymictic (that frequently mix), and naturally eutrophic or hypereutrophic. Most are turbid-phytoplankton, with high algal biomass, turbidity, and absence of submerged macrophytes, while others remain clear-vegetated, with abundant macrophytes, lower algal biomass, and lower turbidity. Clear and turbid lakes usually show distinct phytoplankton community structures^{36,39,42}.

Field experiments were carried out in three shallow Pampean lakes located in the province of Buenos Aires, where seasonal studies of their limnological conditions, phytoplankton structure and CO₂ and CH₄ emissions had been previously conducted^{36,43}: *La Salada* (SA), *El Burro* (BU) and *La Segunda* (SG) (Fig. S1). SA and BU are phytoplankton-turbid, whereas SG is clear-vegetated. These lakes tend to present different phytoplankton abundance and community compositions, high CH₄ emissions, and are located within an area of approximately 54 km², so they shared similar climatic conditions during the study period.

Experimental design

Field experiments were carried out in the 2021 austral summer, between 25th and 28th of January in SA; 29th of January and 2nd of February in SG; 2nd and 6th of February in BU. In each lake, three (SA) or four (SG, BU) mesocosms were deployed (Fig. 1A, Fig. S2). Mesocosms were built with the same transparent polycarbonate sheets as in Bogard et al.²⁰, which impedes the diffusion of gases (Suppl. Inf. 1). Mesocosms were 0.8 m deep, 1 m wide, with a volume of 628.3 L and a surface area of 0.8 m². They were closed at the bottom to exclude sediments CH₄ production, were equipped with a floating device and protective rim to prevent lake water entry and were anchored to the sediment for stability. The average depth of the lakes at the time of the experiments was 1.2 m, 0.9 m and 0.9 m for SA, SG and BU, respectively. The enclosures were placed between 60 and 200 m

from the shore of the lakes. Prior to the onset of the experiments, the enclosures were filled with water from 0.2 m below the surface of each lake using a submersible pump (Proactive Pump II, Waterspout 2, Proactive Environmental Products) with a velocity of 11.12 L min^{-1} . The water was run through a shower head device to equilibrate the dissolved gases with the atmosphere. The latter was done to lower the initial CH_4 baseline (while retaining saturation of O_2 and CO_2) and therefore to facilitate the detection of changes in CH_4 concentration within the mesocosms during the experimental period. In addition, water was filtered through a $55 \mu\text{m}$ pore size net to exclude large zooplankton that could graze on phytoplankton. The filling of the mesocosms did not cause sediment resuspension or alter phytoplankton morphology, as subsequently verified by the analysis of phytoplankton samples. After filling the mesocosms, high frequency oxygen (O_2) and temperature (T) sensors (miniDO₂T, Precision Measurement Engineering, Inc.) were deployed inside each mesocosm as well as in the lake, in all cases at 0.4 m depth. These devices measured T ($^{\circ}\text{C}$), O_2 (mg L^{-1}), and O_2 saturation (%) every 5 min for the entire duration of the experiment. Note that in the clear lake submerged macrophytes were not included inside the mesocosms, since our study primarily focused on exploring CH_4 production by the planktonic communities. The length of the experimental deployment varied slightly among lakes due to logistic considerations, including constraints imposed by COVID restrictions. To ensure consistency, here we present the results from the initial 100-h deployment for all experiments. After filling in the enclosures, a 24-h acclimation followed, after which the limnological sampling was carried out. The only parameters sampled immediately after filling in the enclosures were the first point of dissolved CH_4 and $^{13}\text{C-CH}_4$. The detailed sampling design is shown in Table S1.

Limnological characterization

In both mesocosms and lakes, T and O_2 high-frequency sensors were supplemented with water T and O_2 profiles measured at 10 cm intervals. An irradiance profile was carried out in the lakes, to assess the vertical attenuation coefficient for photosynthetically active radiation (K_{dpar}), and the euphotic depth was derived from this. Additionally, pH, turbidity, total suspended solids (TSS), total phosphorus (TP), total nitrogen (TN), dissolved organic carbon (DOC), dissolved inorganic carbon (DIC), chlorophyll *a* (Chl*a*), and phytoplankton abundance and composition were analyzed both in the mesocosms and the lakes. Details of all these methods can be found in Suppl. Inf. 2. Archaeal and bacterial community compositions were analyzed in the lakes and mesocosms, as described in Suppl. Inf. 3. Ecosystem metabolism was calculated based on O_2 dial variations, as specified in Suppl. Inf. 4. Atmospheric pressure, humidity and wind speed were recorded using a Kestrel 4000 Pocket Weather Tracker[®] (Nielsen-Kellerman[®]).

Greenhouse gas analysis

Dissolved gas concentration and isotopic values

Dissolved CH_4 and carbon dioxide (CO_2) concentration in the water along with $^{13}\text{C-CH}_4$ and $^{13}\text{C-CO}_2$, were obtained by means of the headspace equilibration method⁴⁴. Two 60 ml syringes were filled with 30 ml of water and 30 ml of atmospheric air, creating a 1:1 water: air ratio. Syringes were vigorously shaken for 2 min to allow equilibration of gases between water and air, and then the 30 ml of air were injected into 12 ml glass pre evacuated vials equipped with crimped rubber stoppers (Exetainer, Labco). Headspace samples were analyzed using a cavity ringdown spectrometer (CRDS) coupled with a Small Sample Isotopic Module (SSIM, Picarro G2201-i) to obtain the partial pressures (ppmv) and ^{13}C values of CH_4 and CO_2 . The original ambient partial pressure and isotopic values were obtained following Soued and Prairie⁴⁵ and partial pressure (ppmv) was converted to concentration (μM) considering alkalinity, following Koschorreck et al.⁴⁶. A more detailed description of the method can be found in Suppl. Inf. 5. Throughout the experimental course, dissolved CH_4 and CO_2 alongside $^{13}\text{C-CH}_4$ and $^{13}\text{C-CO}_2$ were measured in the lake and in each mesocosms five times in SG, six times in SA and seven times in BU (Table S2). Differences in the number of measurements respond to logistic considerations, including constraints imposed by COVID restrictions.

Diffusive fluxes

Diffusive flux of CH_4 in the air–water interface was measured using an opaque floating chamber^{36,47} (extra information in Suppl. Inf. 6 and Fig. S3). The diffusive flux rates (f_{gas}) were calculated in $\text{mmol m}^{-2} \text{d}^{-1}$, following Eq. 1. Chamber measurements were inspected for bubble events based on whether there was an abrupt increase of CH_4 or the pattern of CH_4 increase over time was not following a strong linear relationship ($R^2 < 0.85$). All chamber measurements were performed during the daytime.

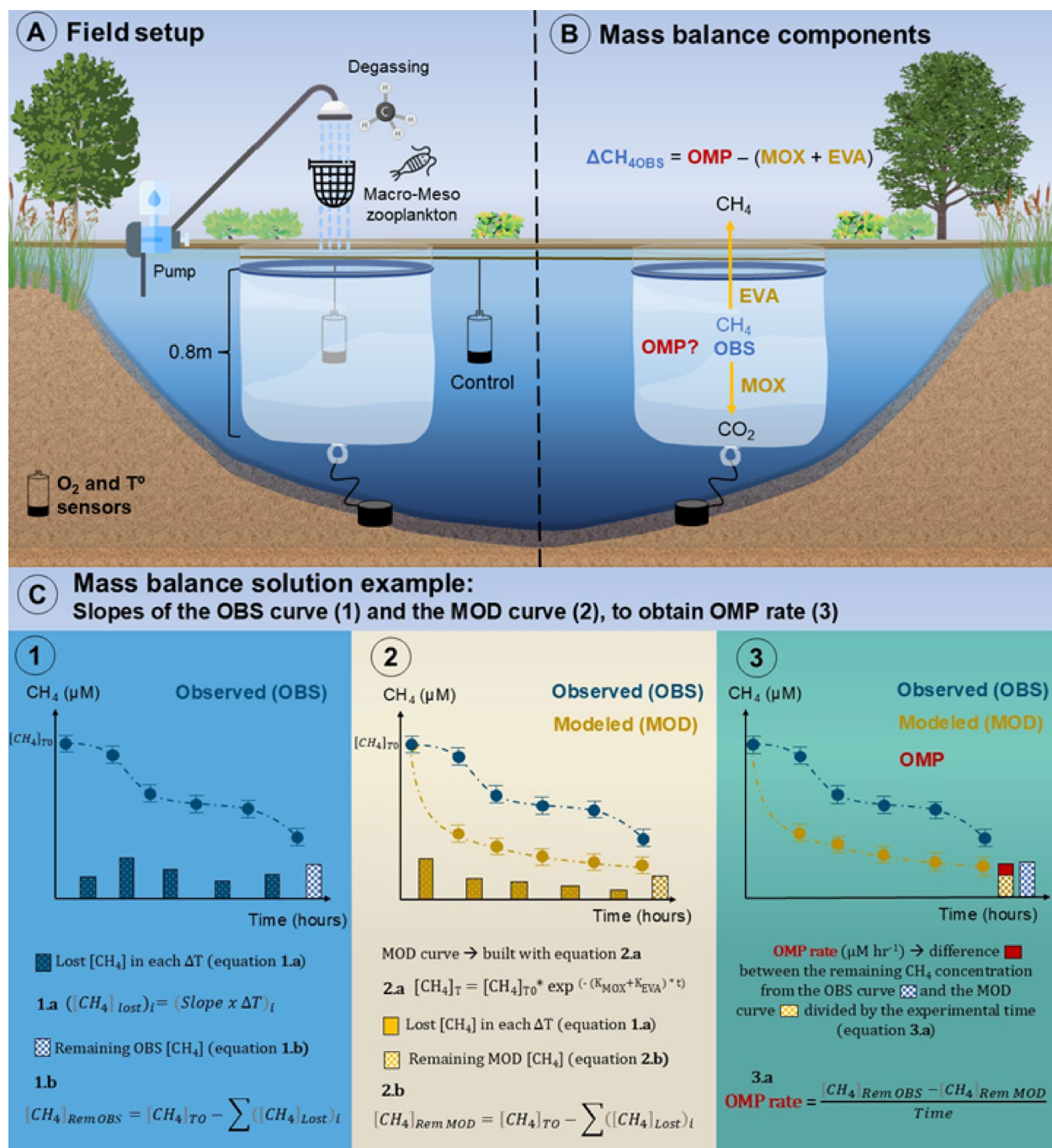
$$f_{\text{gas}} = \left(\frac{s * V}{mV * A} \right) * t \quad (1)$$

where *s* is the accumulation rate of gas in the chamber (ppm min^{-1}); *V* is the volume of the chamber (L); *A* is the chamber surface area covering the water (m^2); *mV* is the molar volume of the gas at ambient temperature and pressure (L mmol^{-1}); and *t* is a factor that converts minutes to days (1 day = 1440 min)⁴⁷.

Gas transfer velocity

Gas transfer velocities (*K*) were calculated based on floating chamber measurements of gas exchange carried out inside the mesocosms and in the lake^{43,47,48} (Eq. 2).

$$K = \frac{Flux_{\text{gas}}}{Kh * \Delta p_{\text{Gas}}} \quad (2)$$



where Flux gas is the diffusive flux for CH_4 obtained from Eq. 1 ($\text{mmol m}^{-2} \text{ d}^{-1}$), K_h is the Henry's constant correspondent corrected for atmospheric pressure and water temperature, and Δp_{Gas} is the difference between the partial pressure of the gas in the water (P_w) and the partial pressure of the gas in equilibrium with the atmosphere (P_{eq}), i.e. $\Delta p_{\text{Gas}} (\text{ppmv}) = P_w - P_{eq}$.

The obtained values of K were standardized to a Schmidt number of 600 (Eq. 3), obtaining the standardized K_{600} .

$$K_{600} = \frac{K_{CH_4}}{(SC_{CH_4}/600)^{-n}} \quad (3)$$

where Sc is the Schmidt number of a given gas at a given temperature⁷⁵, and n is a value that depends on wind speed. We used a value of $n = 2/3$ for ambient wind speeds $< 3.7 \text{ m s}^{-1}$ and of $n = 1/2$ for ambient wind speeds $> 3.7 \text{ m s}^{-1}$. Wind speed in the enclosures was measured close to the protection rim.

Given that the mesocosms were well mixed (Fig. S4 and S5), K_{600} (m h^{-1}) can be expressed as an evasion decay constant (K_{EVA} , h^{-1}) when divided by the depth of the mesocosm (0.8 m) and was used for the mesocosm CH_4 mass balance calculations (see Section 6).

◀ **Fig. 1.** Overview of mesocosm field experiments and mass balance approach to estimate OMP rates. (A) Illustration of the field setup, indicating how the mesocosms were filled and where the O_2/T sensors were located; (B) Mass balance components: the change in CH_4 dissolved inside the enclosures between two consecutive time points (ΔCH_{4OBS}) is the result of the potential CH_4 produced through OMP (OMP) minus the CH_4 oxidized (MOX) and the CH_4 lost to the atmosphere through diffusion (EVA); (C) Mass balance solution example: the exemplified modeled curve predicts the expected CH_4 concentration in the mesocosms considering loss of CH_4 through oxidation (MOX) and evasion (EVA), and compares this to the exemplified observed curve (OBS). If the observed curve is higher than the modelled one, this implies existence of OMP, because the mesocosms are isolated from the sediment. The steps for this approach are: (1) estimating the remaining CH_4 concentration at the end of the experiment by integrating CH_4 concentration over time (which is done by multiplying the slope of the CH_4 concentration vs time of each time segment, by the respective delta time) (Eq. 1.a), to calculate the difference between the initial CH_4 mass and the total change in CH_4 mass over the course of the experiment (Eq. 1.b); (2) obtaining the expected CH_4 concentration in the mesocosms as the result of loss of CH_4 by oxidation and diffusion was modeled using Eq. 2.a. The remaining modeled CH_4 concentration at the end of the experiment was obtained as described before, using Eq. 1.a to obtain the change in CH_4 in each time segment, and Eq. 2.b to obtain the final CH_4 concentration; (3) subtraction of the remaining modeled CH_4 concentration from the remaining observed CH_4 concentration, divided by the time course of the experiment (Eq. 3.a). A detailed description of the mass balance solution can be found in Section 6. *Tree and bush symbols from Dylan Taillie and Jane Hawkey, respectively, and emergent macrophyte symbols from Tracey Saxby, Integration and Application Network, University of Maryland Center for Environmental Science.*

Methane oxidation (MOX) rates

To estimate CH_4 oxidation (MOX) rates, dark incubations were carried out for each lake and for the mesocosms (scheme of the workflow and specific details in Fig. S6)^{20,49,50}. Since MOX follows first order kinetics, the instantaneous CH_4 oxidation rate (h^{-1}) for each lake or mesocosm can be obtained as the slope of the regression between \ln (dissolved CH_4) (μM) vs Time (h)^{32,49}. This estimate of oxidation decay constant (K_{MOX}) was used for the mesocosm CH_4 mass balances calculations (see Section 6 below).

Mesocosm mass balances

Given that the mesocosms were closed at the bottom, impermeable to gases and remained fully oxic during the entire length of the experiment (Fig. S5), any observed inputs of CH_4 would have to originate from the mesocosm itself, and this would correspond to OMP occurring in the water column, since the short deployment time did not allow for significant phytoplankton wall growth development. Therefore, the change in CH_4 concentration between two consecutive time points would be the result of the CH_4 produced, minus the CH_4 oxidized and diffused to the atmosphere²⁰ (Eq. 4):

$$\Delta CH_4 = OMP - (EVA + MOX) \quad (4)$$

where ΔCH_4 is the change in CH_4 concentration between two consecutive time points, OMP stands for oxic methane production rate, EVA reflects the rate of CH_4 evasion to the atmosphere through diffusion, and MOX is the rate of CH_4 oxidation. If there was no production of CH_4 inside the mesocosms ($OMP=0$), the concentration of CH_4 inside the enclosures would continuously decline to eventually equilibrate with the atmosphere, at a time frame that is dependent on the initial CH_4 concentration and the total CH_4 loss rate ($EVA + MOX$). Following this reasoning, CH_4 concentrations above what would be expected based on the total CH_4 loss would necessarily be due to OMP inputs.

The OMP component from Eq. 4 cannot be directly measured, but it can be indirectly derived from the rest of the components of the mass balance: CH_4 concentration was measured in the mesocosms at each time point; the evasion rate was measured with floating chambers (Section “Diffusive fluxes”); and the MOX rate was estimated in dark incubations (“Methane oxidation (MOX) rates”). The empirical dissolved CH_4 data obtained at each time point allows us to build an empirical curve describing the behavior of CH_4 through time. This observed curve can be further compared to the theoretical curve that predicts the expected CH_4 concentration in the mesocosm at each time point resulting from CH_4 loss due to oxidation and evasion (Eq. 5). This theoretical curve was calculated based on the K_{eva} from the diffusive flux data (Section “Gas transfer velocity”), and the K_{oxi} estimated from experimentally derived MOX data (Section “Methane oxidation (MOX) rates”). If the observed curve is higher than the theoretical curve modeled based on CH_4 loss both from oxidation and evasion (Eq. 5), this implies an excess of CH_4 relative to the expected concentration, indicating input from OMP.

$$[CH_4]_t = [CH_4]_{t_0} * exp^{(-K_{(MOX+EVA)} * t)} \quad (5)$$

where $[CH_4]_t$ corresponds to the modeled concentration of CH_4 at a given time point (t , in μM), $[CH_4]_{t_0}$ corresponds to the concentration of CH_4 at time zero of the mesocosm experiment (t_0 , μM), K_{MOX} corresponds to the decay constant of MOX (h^{-1}) obtained from the dark incubations, t corresponds to a given time (h) and K_{EVA} corresponds to the decay constant of evasion (h^{-1}) obtained from the floating chamber measurements.

To solve the mass balance proposed in Eq. 4 we used an approach based on integrating the change in the mass of CH_4 between consecutive time points for each mesocosm and for the entire length of the experiment (Fig. 1C), both for the observed concentrations (Fig. 1C panel 1), and the modeled concentrations based on Eq.

5 (Fig. 1C, panel 2), an extension of the mesocosm-based approach applied by Bogard et al.²⁰. We should point out that CH₄ concentrations declined in all mesocosms through time, so the approach described above involved reconstructing the patterns of loss in observed and predicted CH₄ concentrations, and comparing the resulting remaining masses of CH₄ to derive potential OMP rates in each mesocosm (Fig. 1C, panel 3). Positive differences between these final remaining masses represent the mass of CH₄ produced through OMP, and all the mesocosms yielded overall positive estimates.

Although the water used for the mesocosms was degassed through a shower head device during filling, the initial mesocosm CH₄ concentrations differed greatly (by orders of magnitude) between mesocosms of the different lakes, reflecting the vastly different ambient lake concentrations at the time. Given that we are modeling CH₄ losses as first order processes, which depend on initial CH₄ concentrations, we standardized the observed and modeled CH₄ concentrations in each mesocosm to their respective initial concentration to remove potential biases induced by large differences in initial ambient concentrations and thus render comparable OMP rates. Using these standardized concentrations (unitless), we derived OMP rates following the scheme presented in Fig. 1C, which yielded OMP rates in units of time⁻¹ rather than as μM time⁻¹. In Fig. S7 we present the observed concentrations as a function of time for each mesocosm, that are the basis for these calculations.

The uncertainty around the modeled curves (MOX, EVA, and MOX + EVA) was estimated using Monte Carlo simulations. These simulations incorporated the variability in the model parameters, which were the mean and standard deviation of K_{MOX} and K_{EVA} specific to each lake, and the mean and standard deviation of the standardized initial CH₄ concentration for each mesocosm. For each of the 10,000 simulations, random parameter values were sampled from normal distribution curves defined by these means and standard deviations, and the model was repeatedly evaluated over the range of time points. The resulting ensemble of model outputs was then used to calculate the mean predicted curve and its associated uncertainty.

OMP contribution to total lake CH₄ diffusive flux (OMC)

To estimate the contribution of OMP to total lake CH₄ emissions, we compared the standardized OMP rates (day⁻¹) determined in the mesocosms to the standardized CH₄ diffusive fluxes from the lakes (day⁻¹) (Eq. 6). CH₄ diffusive fluxes from the lakes were standardized to the CH₄ concentration in the lake at the moment of the diffusive flux measurement, and to the area and volume of the lake. The surface area of the lake is known from studies done previously in area³⁶ and the volume was obtained as the mean depth (m) multiplied by the surface area (m²), a good estimation for these types of shallow systems which are pan-shaped and have a relatively uniform depth⁷⁷.

$$OMC (\%) = \frac{(standardizedOMP * 100)}{standardizedlakeFlux} \quad (6)$$

where *OMC* is the contribution of OMP to lake CH₄ emissions (%), *standardized OMP* is the standardized aerobic CH₄ production measured in the mesocosms (d⁻¹) and *standardizedlakeFlux* is the standardized CH₄ diffusive flux measured in the respective lake (d⁻¹). OMC was calculated for each measured CH₄ diffusive flux in each lake.

Mesocosm isotopic mass balances

To calculate ¹³C values of CH₄ potentially associated with oxic production, (δ¹³C-CH_{4-OMP}), a two-step isotopic mass balance was carried out. First, the measured ¹³C-CH₄ in the mesocosms was corrected to remove the effect of fractionation due to evasion and oxidation. The fractionation factor of evasion (α_{eva}), a value of 1.0008, was obtained from the literature⁵¹. The fractionation factor of oxidation (α_{ox}) was calculated using data from our own dark incubations. The slope from the regression between ln [CH₄] vs ln (¹³C-CH₄ + 1000) was used to obtain α_{ox} using Eq. 7.

$$\alpha_{ox} = \frac{slope}{1 + slope} \quad (7)$$

Subsequently, ¹³C-CH₄ was corrected for evasion and oxidation using Eq. 8.

$$\delta^{13}CH_{4\ corr} = \frac{Evasion}{Evasion + MOX} * (\delta^{13}CH_{4\ ambient} - (\alpha_{eva})) + \frac{MOX}{Evasion + MOX} * (\delta^{13}CH_{4\ ambient} - (\alpha_{mox})) \quad (8)$$

where δ¹³CH_{4corr} corresponds to the ¹³C-CH₄ corrected by evasion and oxidation, *Evasion* corresponds to the expected rate of EVA (μM hr⁻¹) for each enclosure and each time point, which was obtained from the modeled curve considering only loss of CH₄ through evasion. *MOX* correspond to the expected rate of MOX (μM hr⁻¹) for each enclosure and time point, which was obtained from the modeled curve considering only loss of CH₄ through oxidation. δ¹³CH_{4ambient} corresponds to the ¹³C-CH₄ of measured CH₄ in the water column of the mesocosm, α_{eva} and α_{mox} are the fractionation factors, both in delta form (‰), obtained as ((α - 1) * 1000)³².

δ¹³CH_{4corr} was further used along with the ¹³C-CH₄ of the water used to fill the mesocosms at the onset of the experiment to derive the δ¹³C-CH_{4-OMP} following Eq. 9.

$$\delta^{13}CH_{4\ OMP} = \frac{(CH_{4\ zero} * \delta^{13}CH_{4\ zero}) + (CH_{4\ Tx} * \delta^{13}CH_{4\ corr\ Tx})}{(CH_{4\ zero} + CH_{4\ Tx})} \quad (9)$$

where $\delta^{13}\text{C}H_{4\text{OMP}}$ is the ^{13}C of CH_4 produced through OMP, $\text{CH}_{4\text{zero}}$ and $\delta\text{CH}_{4\text{zero}}$ are the concentration (μM) and the ^{13}C - CH_4 of the water used to fill the mesocosm, respectively; $\text{CH}_{4\text{Tx}}$ is the concentration of CH_4 (μM) at any given time point and $\delta^{13}\text{C}H_{4\text{corrTx}}$ is the ^{13}C - CH_4 at that given time point, which was previously corrected for fractionation due to evasion and oxidation. We estimated $\delta\text{CH}_{4\text{OMP}}$ for each time point of the experimental mesocosm time course (five time points for SG, six time points for SA and seven time points for BU), and here we report the average value for the entire experiment.

Phytoplankton cultures

To assess the potential for CH_4 production by phytoplankton present in the study lakes, phytoplankton species were isolated from each one of the three lakes. Water from SA, SG and BU was collected and filtered through a 55 μm net to exclude macro and mesozooplankton, on the 5th of May 2022. The water was transported to the laboratory, where it was inoculated in petri dishes⁷⁸ with agar mediums BG11⁵², Bold's Basal Medium⁵³ (BBM), BBM + Vitamins (cyanocobalamin, thiamine and biotin) and BBM + soil extract (3:1, v/v), in all cases using the spray technique⁵⁴. Three petri dishes per medium and lake were inoculated, obtaining a total of 48 inoculated plates. These were kept under controlled conditions of light (photoperiod 12:12 light: darkness) and temperature (25 °C). Weekly identification of growing colonies was done using a dissection microscope (Nikon SMZ 745 T, 5 \times to 50 \times). When a colony was detected, it was removed from the petri dish under sterile conditions, observed in an optical microscope (Olympus BX50, using 400 \times and 1000 \times) to identify the genera using specific bibliography^{55–57} and later inoculated in another petri dish with the same growth medium for further isolation, establishing non axenic unialgal stock cultures. Further experiments were carried out with active liquid cultures developed from the petri dish cultures, using the same culture media. Although it was not possible to isolate all the dominant genera present in these shallow lakes, further experiments were carried out including 4 genera of chlorophytes and 3 genera of cyanobacteria that were prevalent in the lakes and that are in general representative of Pampean shallow lakes^{39,58}.

Experiments to measure methane production by algal strains using membrane inlet mass spectrometry (MIMS)

Experiments were carried out to assess the potential production of CH_4 by the phytoplankton isolates using a membrane inlet mass spectrometer (MIMS, Bay-Instruments, Fig. S8)²⁴. Each culture was placed in a 3.5-ml glass chamber that was surrounded by an acrylic jacket connected to a recirculating water bath used to maintain the culture at a constant temperature of 25 °C. The culture chamber was located above a stirrer, to ensure mixing and to avoid gradients, and it was exposed to a photoperiod of 15 h light: 9 h darkness (similar to the photoperiod in summer in Argentina), at a light intensity of 120 $\mu\text{mol photons m}^{-2} \text{s}^{-1}$. The culture chamber had an inlet and an outlet, and the culture fluid was continually circulated through the MIMS exchanger by means of a small peristaltic pump. O_2 and CH_4 in the culture were measured every 12 s, and only one culture at a time could be processed. The extent of MIMS physical loss depends on CH_4 concentration within each culture: to characterize this physical CH_4 loss, autoclaved cultures were employed to establish a connection between the initial CH_4 concentration in a culture and the rate of physical CH_4 loss through the MIMS. Since these were dead cultures, they lack biological fluctuations in CH_4 concentration and solely exhibit CH_4 loss due to physical factors. Leveraging this dataset, a linear relationship between the initial CH_4 concentration and the rate of physical loss was derived. This correlation was subsequently used to estimate the physical loss for each measured culture, considering their initial CH_4 concentration (Fig. S9). Each experiment lasted between three to five days, and two to three experiments were carried out for each culture: at least one measurement of live cultures and, for most strains, one measurement of the autoclaved (dead) culture. As negative controls, ultrapure water and sterile BG11 culture media were used. Differences in CH_4 production rates between Chlorophyte and Cyanobacterial strains were analyzed using a two-way ANOVA, with *Group* (Chlorophytes vs. Cyanobacteria) as a fixed factor and *Strain* as a random factor, using package lmerTest 3.1–2⁵⁹. Assumptions of normality and homogeneity of variances were tested using package Car 3.0–8⁶⁰. Tests were performed at the 95% significance level using R version 3.6.2 in the RStudio environment version 1.2.5019.

At the beginning and end of each experiment, chlorophyll a (Chla) was measured, and ambient DNA was extracted from the culture (Suppl. Inf. 8). Chla measurements were done to standardize phytoplankton-derived methane production rates to biomass, whereas DNA extraction followed by PCR was carried out to test for the presence of methanogenic archaea and methanotrophic bacteria.

Phytoplankton methane production rates and contribution to field OMP rates

Phytoplankton methane production rates were calculated using the Stavisky-Golay function from the Signal package in R (<http://r-forge.r-project.org/projects/signal/>)²⁴. First, CH_4 concentration vs time interval curves were smoothed using the sgolay function, fitting a polynomial of second degree and no derivative. The sgolay function was then used to obtain the first derivative of this smoothed curve—which corresponds to the rate—also fitting a second-degree polynomial. The rate thus obtained was then corrected for the rate of physical loss of gas from the experimental setup (derived as described above) and was standardized to the Chla concentration of each culture, obtaining rates in units of $\mu\text{mol CH}_4 \text{ hr}^{-1} \text{ gr Chla}^{-1}$.

The potential contribution of phytoplankton CH_4 production to field OMP rates was estimated by scaling the estimated CH_4 production of each strain to the mean Chla concentration in the mesocosms of each lake, and then relativized for the mean CH_4 dissolved concentration ($\mu\text{mol L}^{-1}$) in the mesocosms at the end of the experiments, obtaining a value in day^{-1} , that was afterwards compared to the mean estimated standardized OMP rate (day^{-1}) in each lake. It was decided to do this upscaling exercise for each strain separately, assuming that the enclosures would be fully dominated by one of those strains in each case, to explore how the different

rates would affect the contribution. It was also decided to relativize the phytoplankton CH₄ production rates for the mean CH₄ dissolved concentration in the enclosures at the end of the experiment because, according to our calculations, at that point the CH₄ dissolved remaining in the enclosures is attributable to OMP, whereas at the beginning of the experiments there is CH₄ being lost by oxidation and evasion to the atmosphere, which would underestimate the contribution.

Results
Limnological characteristics

Mesocosms of SG had a higher transparency, lower total phosphorus (TP), total nitrogen (TN), total suspended solids (TSS) and phytoplankton abundance than the mesocosms of SA and BU (Table 1). Compared to SA, the mesocosms in BU had higher levels of turbidity and TSS. The mesocosms in SA had a higher Chla than the mesocosms of BU. The BU mesocosms were dominated by smaller phytoplankton species that occurred at a higher abundance, whereas the SA mesocosms had the opposite pattern, with larger phytoplankton species dominating. Concentrations of dissolved organic carbon (DOC), dissolved inorganic carbon (DIC), dissolved O₂, and O₂ saturation levels were generally high and comparable across the mesocosms of all three lakes. Based on the diel variability on O₂, the mesocosms from SG were on average net heterotrophic, whereas the mesocosms from SA and BU were on average net autotrophic. All the studied lakes were on average net autotrophic. The difference in GPP and RE between mesocosm and lake of SG is related to the presence of submerged macrophytes in the lake, but the absence of them in the mesocosm. The difference in GPP between the mesocosms and lake of SA and BU are likely related to the slight differences in the abundance of primary producers.

The water temperature, (water T), pH, O₂ concentration and saturation, and dissolved CH₄ were measured at each time point (five time points in SG, six time points in SA and seven time points in BU). Turbidity, TSS, K_{dpar}, euphotic depth, DOC, DIC, TP, TN, Chla, phytoplankton abundance and composition were assessed at the beginning and end of each experiment (two time points). T of the water, dissolved O₂ and O₂ saturation correspond to sub superficial values. NA means there is no data. Secchi depth was not registered in lake SG because the submerged macrophytes do not allow a comparable measurement. GPP (gross primary production), ER (ecosystem respiration) and NEP (net ecosystem production) were calculated based on diurnal O₂ variations obtained from the high frequency data loggers: for the lakes the informed value corresponds to the daily mean for the one miniDOT located in the lake, whereas for the mesocosms the reported value represents the daily mean of the miniDOTs deployed inside replicate mesocosms.

Phytoplankton community composition differed between the three lakes but was similar between the lake and the corresponding mesocosms (Fig. S10). In SG the dominant genera were *Chlamydomonas* sp. and *Didymocystis* sp. (Chlorophyta), *Cryptomonas* sp. (Cryptophyta) and *Coelosphaerium* sp. (Cyanobacteria). In SA, there was an almost complete dominance of *Scenedesmus linearis* (Chlorophyta) (52–68% of the total phytoplankton abundance) followed by *Oocystis* sp., *Eutetramorus* sp. and *Cosmarium* sp. (Chlorophyta). In BU, the dominant genera were *Monoraphidium* sp., *Oocystis* sp., and *Scenedesmus* sp. (Chlorophyta), and *Planktolyngbya* sp., *Geitlerinema* sp. and *Anabaenopsis* sp. (Cyanobacteria).

Parameters	SG		SA		BU	
Treatment	M	L	M	L	M	L
K _{dpar} (m ⁻¹)	1.86 ± 0.42		4.37 ± 0.08		8.53 ± 1.93	
Secchi depth (m)	NA		0.38 ± 0.05		0.19 ± 0.01	
Euphotic depth (m)	2.6 ± 0.6		1.03 ± 0.05		0.56 ± 0.11	
Water T (°C)	23.02 ± 3.17	22.86 ± 3.23	27.78 ± 2.22	27.85 ± 2.12	23.08 ± 1.70	22.71 ± 1.71
pH	9.3 ± 0.17	9.41 ± 0.07	9.46 ± 0.02	9.41 ± 0.04	9.15 ± 0.03	9.06 ± 0.06
Dissolved CH ₄ (μmol L ⁻¹)	26.02 ± 24.45	33.37 ± 39.20	0.42 ± 0.21	1.48 ± 0.28	0.04 ± 0.04	0.27 ± 0.03
Dissolved O ₂ (mg L ⁻¹)	7.73 ± 0.49	10.86 ± 5.13	10.01 ± 1.68	10.21 ± 2.72	10.99 ± 0.96	11.13 ± 1.13
O ₂ saturation (%)	90.18 ± 10.23	128.79 ± 67.43	128.37 ± 27.14	131.68 ± 41.13	127.79 ± 13.39	128.69 ± 16.72
Turbidity (NTU)	2.48 ± 0.78	2.10 ± 0.33	46.45 ± 10.27	46.15 ± 7.55	94.53 ± 8.84	107.25 ± 12.75
TSS (mg L ⁻¹)	2.31 ± 0.56	3.60 ± 0.73	24.35 ± 0.46	22.85 ± 3.35	55.50 ± 2.96	53.79 ± 0.21
DOC (mg L ⁻¹)	38.86 ± 0.51	38.01 ± 1.28	44.12 ± 0.08	43.12 ± 0.39	36.59 ± 0.55	36.14 ± 0.36
DIC (mg L ⁻¹)	98.58 ± 3.64	96.79 ± 3.44	108.65 ± 4.23	108.88 ± 0.67	81.21 ± 3.97	83.69 ± 0.08
TP (μg L ⁻¹)	82.50 ± 19.38	75.00 ± 9.00	180.00 ± 44.90	183 ± 39	259.50 ± 33.91	288 ± 0.00
TN (μg L ⁻¹)	2175 ± 417	2370 ± 330	2310 ± 475	1920 ± 0	2400 ± 983	2850 ± 570
Chla (μg L ⁻¹)	2.45 ± 0.93	6.58 ± 1.41	117.14 ± 22.72	126.12 ± 0.71	87.09 ± 17.55	127.46 ± 4.74
Phytoplankton (ind mL ⁻¹)	1364.75 ± 463.05	2572.50 ± 399.50	113,906.25 ± 18,905.78	73,287.50 ± 4104.50	148,524.43 ± 32,870.87	78,454.00 ± 15,270.00
Cyanobacteria (%)	24.56 ± 13.66	23.83 ± 5.24	1.05 ± 1.06	1.50 ± 0.70	72.14 ± 6.80	62.49 ± 4.32
Chlorophyta (%)	46.65 ± 12.21	31.79 ± 5.19	96.59 ± 1.05	95.44 ± 1.87	25.22 ± 5.33	33.67 ± 6.36
GPP (g O ₂ m ⁻² d ⁻¹)	0.47 ± 0.05	13.37 ± 0.33	6.24 ± 0.43	11.60 ± 0.68	4.91 ± 0.09	7.48 ± 0.09
RE (g O ₂ m ⁻² d ⁻¹)	0.59 ± 0.05	11.92 ± 0.41	6.98 ± 0.41	7.59 ± 0.64	4.60 ± 0.10	4.04 ± 0.10
NEP (g O ₂ m ⁻² d ⁻¹)	-0.21 ± 0.20	2.63 ± 4.50	0.99 ± 1.37	4.01 ± 0.41	1.97 ± 1.12	3.44 ± 1.15

Table 1. Mean ± standard deviation values for mesocosms (M) and the lake (L).

Methanogenic archaea were detected in water samples of all three lakes and their respective mesocosms (Fig. S11a). The class Methanomicrobia was the most widespread methanogenic group and was detected in all three lakes and their mesocosms, whereas the class Methanobacteria was only detected in the lake and mesocosms of SG. Methanotrophic bacteria were also detected in water samples of all three lakes and their respective mesocosms (Fig. S11b). Methanotrophs from the class Gammaproteobacteria were detected and most abundant in all lakes and mesocosms, whereas methanotrophs from the class Alphaproteobacteria was detected in mesocosms and lakes in BU and SA, but in SG only in the mesocosms at the end of the experiment.

CH₄ dynamics in lakes and experimental mesocosms

Patterns in dissolved CH₄ and $\delta^{13}\text{C}-\text{CH}_4$

The lakes differed greatly in ambient surface water CH₄ concentration at the time of mesocosms deployment, with average concentrations of $122.8 \pm 10.9 \mu\text{M}$, $1.5 \pm 0.2 \mu\text{M}$ and $0.3 \pm 0.1 \mu\text{M}$ for SG, SA and BU, respectively (Fig. S12 a, c and e). CH₄ concentrations in the mesocosms were consistently lower than in the surrounding lake, suggesting partial degassing during filling. The initial CH₄ concentration in the mesocosms at the onset of the experiments nevertheless differed by orders of magnitude between lakes, still reflecting ambient lake differences: $65.1 \pm 5.7 \mu\text{M}$, $0.9 \pm 0.1 \mu\text{M}$ and $0.1 \pm 0.0 \mu\text{M}$ for SG, SA and BU, respectively (Fig. S12 b, d and f). CH₄ concentrations subsequently declined in all mesocosms during the experimental time course, whereas in lakes the dynamics of surface water CH₄ followed different patterns (Fig. S12 a–f). The mean CH₄ concentration in the mesocosms was $40.32 \pm 6.30 \mu\text{M}$, $0.42 \pm 0.06 \mu\text{M}$ and $0.04 \pm 0.01 \mu\text{M}$ for SG, SA and BU, respectively, whereas for the lake was $50.6 \pm 45.5 \mu\text{M}$, $1.5 \pm 0.3 \mu\text{M}$ and $0.3 \pm 0.1 \mu\text{M}$ for SG, SA and BU, respectively. The isotopic composition of ambient CH₄ ($^{13}\text{C}-\text{CH}_4$) generally ranged between -20‰ and -40‰ in both the lake and the mesocosms (Fig. S12 g–i), except for a period of very depleted CH₄ that occurred in BU mesocosms between 45 and 75 h (up to -60‰).

CH₄ exchange velocity and diffusive fluxes

Diffusive CH₄ fluxes were higher in the lakes than in the mesocosms (Fig. S13a), which is expected given that the lakes had both higher ambient CH₄ concentrations and higher exchange velocities (Fig. S13b). The mean CH₄ diffusive fluxes from the lakes were $24.7 \pm 13.5 \text{ mmol m}^{-2} \text{ d}^{-1}$, $21.6 \pm 19.5 \text{ mmol m}^{-2} \text{ d}^{-1}$, and $0.5 \pm 0.1 \text{ mmol m}^{-2} \text{ d}^{-1}$ for SG, SA and BU, respectively, whereas the mean fluxes in the mesocosms were $0.6 \pm 0.4 \text{ mmol m}^{-2} \text{ d}^{-1}$, $0.2 \pm 0.1 \text{ mmol m}^{-2} \text{ d}^{-1}$ and $0.02 \pm 0.00 \text{ mmol m}^{-2} \text{ d}^{-1}$ for SG, SA and BU, respectively. Similarly, gas exchange velocities were consistently higher in the lakes than in the mesocosms (Fig. S13b), likely because mesocosms are sheltered from the wind due to the protective rim on the side and reduced overall turbulence. The mean $K_{600} \text{ CH}_4$ for the lake were $0.7 \pm 0.1 \text{ m d}^{-1}$, $2.0 \pm 0.2 \text{ m d}^{-1}$, and $1.6 \pm 0.5 \text{ m d}^{-1}$ for SG, SA and BU, respectively, whereas the mean $K_{600} \text{ CH}_4$ for the mesocosms were $0.1 \pm 0.0 \text{ m d}^{-1}$, $1.2 \pm 0.4 \text{ m d}^{-1}$ and $0.6 \pm \text{NA} \text{ m d}^{-1}$ for SG, SA and BU, respectively. The estimated K_{eva} were 0.01 h^{-1} , 0.06 h^{-1} and 0.03 h^{-1} for SG, SA and BU, respectively.

Methane oxidation (MOX) rates

We observed a general decrease in CH₄ concentrations and a concomitant enrichment of $\delta^{13}\text{C}-\text{CH}_4$ in the dark in vitro incubations, in some cases also coupled with an increase in CO₂ concentration, suggestive of CH₄ oxidation (Fig. S14). The estimated CH₄ oxidation decay constants (K_{MOX}) averaged 0.03 h^{-1} , 0.01 h^{-1} and 0.02 h^{-1} , for SG, SA and BU, respectively (Fig. S15).

Estimates of OMP rates and isotopic signature of CH₄ derived from oxic production ($\delta^{13}\text{C}-\text{CH}_{4-\text{OMP}}$)

OMP rates and OMC

At almost every time point in all mesocosms (except 21 h in BU) the observed CH₄ concentration exceeded the modeled CH₄ concentration based on the combination of MOX + EVA, suggesting CH₄ production in all the mesocosms throughout the experiments (Fig. 2). A plot indicating MOX and EVA curves separately can be found in Fig. S16. The estimated (standardized) OMP rates in the mesocosms of each lake, derived as described in Section 6 of methods, were $0.01 \pm 0.00 \text{ day}^{-1}$, $0.07 \pm 0.01 \text{ day}^{-1}$, and $0.07 \pm 0.01 \text{ day}^{-1}$ for SG, SA and BU, respectively (Table 2). A table with the absolute rates can also be found in Table S3.

The contribution of OMP to total lake CH₄ diffusive flux (OMC) ranged between 0.3 and 6.7% depending on the lake (Table 2).

Isotopic signature of CH₄ derived from oxic production ($\delta^{13}\text{C}-\text{CH}_{4-\text{OMP}}$)

We used an isotopic mass balance approach to derive the potential isotopic signature of CH₄ produced under oxic conditions in the mesocosms. For this mass balance, fractionation factors for CH₄ oxidation (α_{oxi}) were derived from the in vitro MOX dark incubations, and were estimated at 1.02, 1.03 and 1.21 for SA, SG and BU, respectively (Fig. S17). α_{oxi} for BU was too high and the R^2 of this regression (0.83) was weaker than that of the regression for SA (0.99) and SG (0.99). This was presumably related to the fact that in BU CH₄ concentration was very low, which made it difficult to measure $^{13}\text{C}-\text{CH}_4$ precisely. Accordingly, we assumed that the α_{oxi} of BU = SA, since both are turbid phytoplankton-dominated lakes. The estimated $^{13}\text{C}-\text{CH}_4$ OMP for the mesocosms was consistently enriched relative to the isotopic values of CH₄ produced in the surrounding sediments (-62.21 ± 0.14 to -59.81 ± 1.11 , unpublished data from these lakes). $\delta^{13}\text{C}-\text{CH}_{4-\text{OMP}}$ for SA and BU were similar, whereas SG had a more depleted value (Table 2).

The different water colours are related to the abundance of phytoplankton in each shallow lake, which increases from SG to SA and BU.

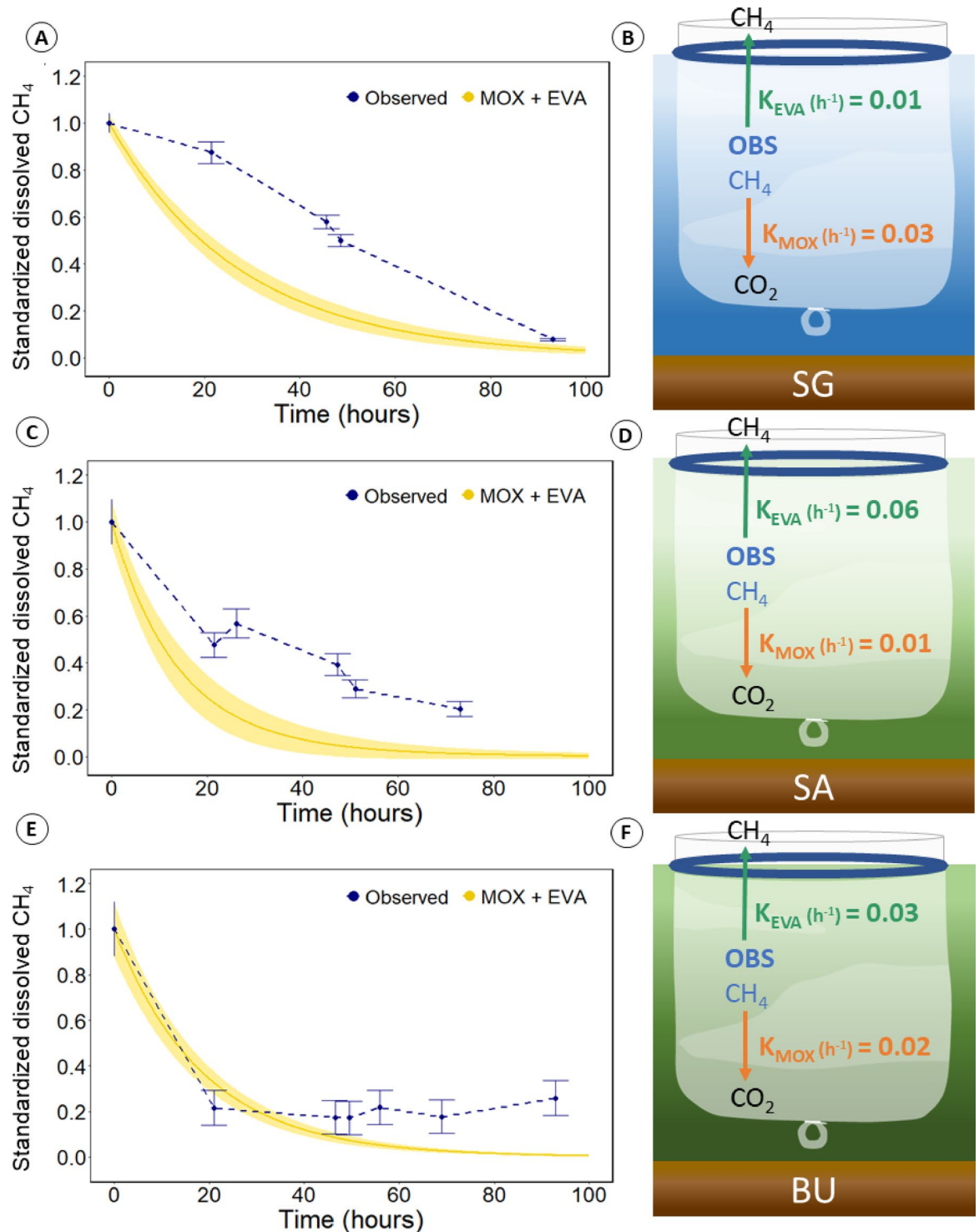


Fig. 2. Observed (blue) and theoretical curve (yellow), the latter indicating the expected CH_4 concentration in the mesocosms assuming no OMP and loss of CH_4 by oxidation (MOX) and evasion (EVA) for SG (a), SA (c) and BU (e). Decay constants of evasion (K_{EVA}) and oxidation (K_{MOX}) for SG (b), SA (d) and BU (f).

In vitro experiments to assess phytoplankton CH_4 production

Four Chlorophyte (*Scenedesmus linearis*, *Scenedesmus quadricauda*, *Monoraphidium circinale*, *Oocystis lacustris*) and three Cyanobacteria (*Phormidium* sp., *Leptolyngbya* sp., *Pseudanabaena* sp.) strains isolated from the three studied lakes were tested for potential CH_4 production with a protocol using MIMS. As controls, Milli-Q water (Fig. S18a) and BG11 medium (Fig. S18b) were used, all of them being equilibrated with sterile-filtered air before measurement in the MIMS. Milli-Q water did not show any changes in CH_4 concentration through time, as

Std OMP (day ⁻¹)	0.01 ± 0.01	0.07 ± 0.01	0.07 ± 0.01
OMC (%)	0.7 – 2.7	0.3 – 4.1	2.9 – 6.7
δ ¹³ C-CH ₄ OMP (‰)	-51.8 ± 0.5	-39.4 ± 0.6	-41.9 ± 1.3
	SG	SA	BU

Table 2. Standardized OMP rates (day⁻¹), contribution of OMP to total lake CH₄ emissions (OMC, %) and ¹³C-CH₄ OMP, with their respective standard errors, for the three lakes studied.

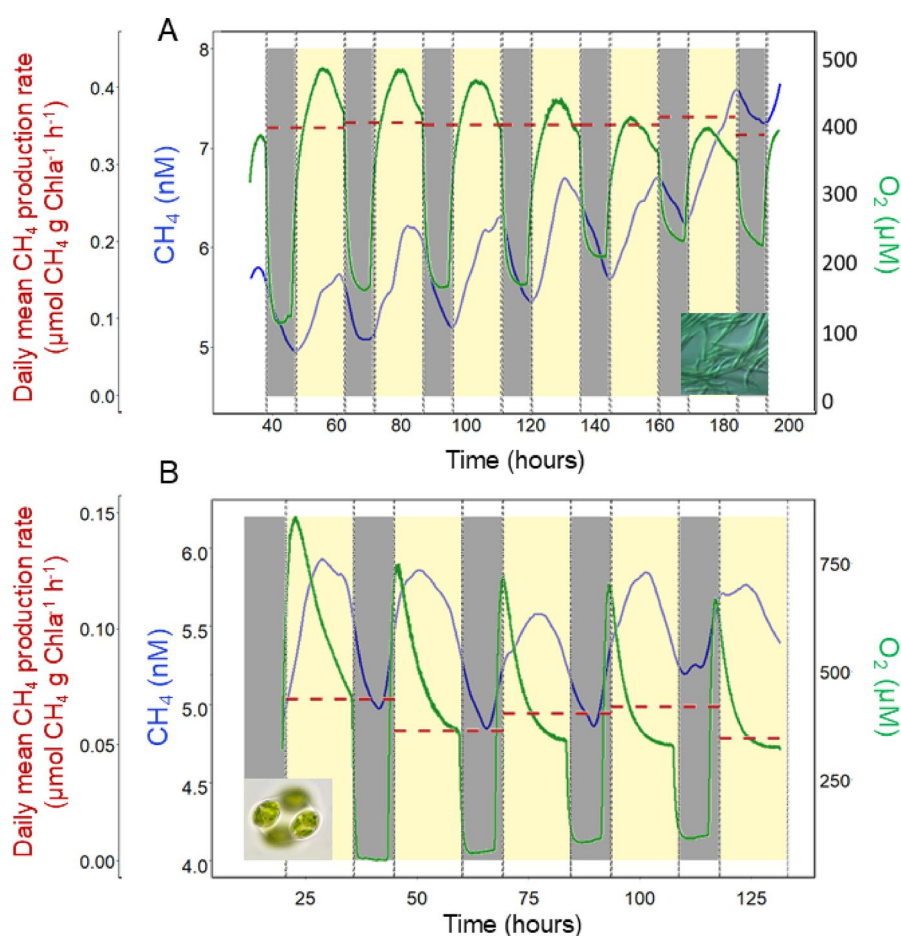


Fig. 3. Dissolved CH₄ and O₂ in the culture and derived phytoplankton CH₄ production daily mean rates, for one of the measurements of *Leptolyngbya* sp. (A) and *Oocystis* sp. (B). Yellow columns correspond to hours of light and grey columns correspond to hours of dark. Picture of *Leptolyngbya* sp. taken from Culture Collection, picture from *Oocystis* sp. taken from AlgaeBase.

expected. Likely because the BG11 medium was not sufficiently equilibrated, there was initially a slight decrease in CH₄ and O₂ concentrations. All tested cultures were alive and had a clear and recurrent diurnal pattern of photosynthesis and respiration as reflected in variations in O₂ concentrations (Fig. S19). The results from two cultures, *Leptolyngbya* sp. and *Oocystis* sp. are shown as examples (Fig. 3). All tested cultures showed increases in CH₄ concentration during light hours, followed by decreases during the dark, and there was an overall coherence

between the diurnal patterns in O₂ and those of CH₄ (Fig. S19). No methanogenic archaea nor methanotrophic bacteria were detected in any of the phytoplankton cultures (Fig. S20). This implies that the observed increases of CH₄ in light conditions can only be attributed to phytoplankton and related to photosynthesis, since there are no other methanogenic organisms present in the culture. On the other hand, the decrease in CH₄ during dark hours must be related to the physical CH₄ loss from the system that offset the decrease in CH₄ production in the dark, since there was no apparent biological CH₄ consumption in the cultures. From the diurnal variations in CH₄ concentrations we were able to derive CH₄ production rates for each of the cultures over several diurnal cycles, and Table 3 shows the mean CH₄ production rate for each culture for the ensemble of incubations that were carried out for each culture. These rates represent the mean CH₄ production per g of Chla and per hour of a 24-h cycle. CH₄ production rates ranged between 0.02 to 0.20 μmol CH₄ g Chla⁻¹ h⁻¹, and no significant differences between Chlorophyta and Cyanobacteria were detected (F (1,4)=0.7, p=0.5), although within each group there were some variations in production rates.

Discussion

We detected OMP in all mesocosms, albeit at very different rates. Mean standardized OMP rates (Table 2, SG 0.01 day⁻¹, SA 0.07 day⁻¹ and BU 0.07 day⁻¹) were obtained using standardized CH₄ concentrations, as explained in the methods section, in units of day⁻¹. The isotopic mass balances revealed an isotopic signature for the CH₄ produced through OMP that was much more enriched (~ - 38‰ to - 52‰) than the CH₄ produced in the sediment, and more similar to the one of oxidized CH₄. There are not many studies that have explored the potential isotopic signature of the CH₄ produced through OMP, but the few studies that have done so have also reported enriched signatures for the OMP-CH₄. Using isotopic whole-lake mass balances, Thottathil et al.³² reported δ¹³C-CH₄ OMP values for four Canadian lakes (- 38.0 ± 1.4‰ to - 63.6 ± 2.2‰) that were also greatly enriched relative to anoxic sediment sources. In line with this, Klitzsch et al.²⁵ explored the isotopic values of CH₄ produced directly by different cultures of marine phytoplankton species, which ranged from - 19.3 ± 0.9 ‰ to - 54.5 ± 1.6 ‰, implying a uniquely enriched signature for phytoplankton-derived CH₄. Similarly, Hartmann et al.²² reported enriched values for cultures of several freshwater phytoplankton species (~ -42‰ to - 50‰). Taenzer et al.³⁵ carried out marine water incubations and reported a MPn-derived δ¹³C-CH₄ of - 40 ± 5‰, indicating also an enriched δ¹³C-CH₄ for MPn derived CH₄. Ours and the above cited results imply that the observed isotopic signature of CH₄ in the water column of freshwater ecosystems it is not just the results of the pathway by which CH₄ was produced in the sediments (acetoclastic or hydrogenotrophic pathways) and the extent of oxidation in the water column, but it also includes the signature from diverse OMP sources, that add CH₄ in the water column with a signature similar to that of oxidized CH₄. This makes MOX mass balances derived from isotopes more complex than previously thought, because the observed isotopic signature of dissolved CH₄ in the water column also includes the confounding influence of OMP.

Estimating OMP rates at an ecosystem scale is extremely challenging, because it involves the quantification of several different processes with high spatio-temporal dynamics that cannot be directly measured and therefore must be derived from other measurable processes, usually through a mass balance. The mesocosm approach greatly simplifies this mass balance approach by excluding sediment CH₄ production, CH₄ bubble dissolution, and lateral transport, allowing us to focus on two components that can be readily measured, CH₄ oxidation (MOX) and CH₄ emission to the atmosphere (EVA), and to derive OMP by difference. At the same time, mesocosms may generate physical and limnological conditions that differ from those of the surrounding lake, yet the factors that are key to OMP, such as nutrients, Chla and DOC remained roughly comparable between mesocosms and lakes throughout the experiments (Table 1). Nevertheless, quantifying CH₄ oxidation and CH₄ diffusive flux to the atmosphere posed a challenge. We estimated MOX using dark incubations, as was done by Bogard et al.²⁰ and Thottathil et al.³². We are aware, however, that MOX rates are affected by CH₄ concentration, O₂ concentration and potentially by light irradiance, where the latter seems to result in MOX inhibition^{61–63} (but

	Phytoplankton strain (genera or specie)	Mean CH ₄ production rate (μmol CH ₄ g Chla ⁻¹ h ⁻¹)	Standard deviation CH ₄ production rate (μmol CH ₄ g Chla ⁻¹ h ⁻¹)
Chlorophyta	<i>Scenedesmus linearis</i>	0.07	0.04
	<i>Scenedesmus quadricauda</i>	0.09	0.01
	<i>Monoraphidium circinale</i>	0.04	0.00
	<i>Oocystis lacustris</i>	0.06	0.04
Cyanobacteria	<i>Leptolyngbya</i> sp.	0.20	0.15
	<i>Phormidium</i> sp.	0.02	0.01
	<i>Pseudoanabaena</i> sp.	0.09	0.02

Table 3. Mean methanogenesis rates (μmol CH₄ g Chla⁻¹ h⁻¹) for the phytoplankton isolates analyzed in the MIMS.

also see^{64,65}). CH₄ and O₂ concentrations were roughly similar between the incubations and the mesocosms, but light irradiance was higher in the latter. Therefore, dark incubations could have led to an overestimation of MOX rates, which translates into an overestimation of OMP rates from the mass balance. Conversely, potential OMP from methylated substrates was not excluded from MOX incubations, which would result in an underestimation of MOX and therefore an underestimation of OMP rates from the mass balance. We are confident, however, that our oxidation data are sound overall (Figures S14 and S15) and that MOX rates are coherent with values reported for other lakes^{29,31,32}. Regarding CH₄ diffusive fluxes, repeated measurements were taken in all mesocosms and lakes. We acknowledge, however, that diffusive fluxes were measured only in the daytime, which may introduce bias. Some studies report higher daytime CH₄ fluxes than at night^{66,67}, others report lower values^{68–70}, yet others find no significant diel differences⁷¹. Consequently, our daytime measurements could have either over- or underestimated true daily fluxes, and therefore OMP rates. Because CH₄ diffusive emissions can vary significantly with weather conditions, we minimized this variability by carrying out mass balances using average gas exchange velocities and wind speeds.

When compared to other standardized OMP rates reported in the literature, which for the most part had much lower chlorophyll concentrations, our lakes were on the lower end (Fig. S21). Despite being eutrophic to hypertrophic, these shallow lakes had OMP rates that were either within the range, or lower than what has been reported for lakes with much lower chlorophyll concentrations (Fig. S21). Previous studies had shown a relationship between chlorophyll concentration and OMP rates across a relatively narrow range of oligotrophic to mesotrophic temperate lakes^{20,32}, but these shallow, highly productive Pampean lakes do not fit this pattern at all. This suggests that chlorophyll is not a universal scaling variable for OMP across lakes, and that factors other than the phytoplankton biomass may drive OMP in lakes of different types^{7,23,29,30,72}.

In this regard, our own experimental results confirmed production of CH₄ by all the tested phytoplankton strains. In all cases, CH₄ production appeared to be linked to photosynthesis based on the coherence in the diurnal patterns of O₂ and CH₄, as had been described before²⁴. It can be noted that, even though there was a general trend of increased CH₄ after an increase in O₂, the specific alignment or lagging between these curves had different daily patterns depending on the strain. Further exploration on these daily patterns exceeds the scope of the study and require further analysis into the specific mechanisms behind CH₄ production by phytoplankton. We observed CH₄ production from both Cyanobacteria and Chlorophyta genera, with CH₄ production rates ranging between 0.02 and 0.2 μmol CH₄ g Chl a⁻¹ h⁻¹. Our results add to the increasing body of evidence of widespread CH₄ production across major marine and freshwater phytoplankton groups^{21,24–27}. Our measured phytoplankton production rates were higher than those reported by Gunthel et al.²¹ for a range of freshwater diatom strains (~0.004 μmol CH₄ g Chl a⁻¹ h⁻¹), but more similar to those reported by Bižić et al.²⁴ for cyanobacterial strains (~0.03–0.004 μmol CH₄ g Chl a⁻¹ h⁻¹) (assuming that approximately half of the dry weight is carbon, and that the Chl a to carbon ratio ranges from 1:20 to 1:60^{80,81}). We observed one order of magnitude range in CH₄ production among the strains tested but this range was not linked to light or nutrient availability since experimental conditions were similar for all strains, and there was not a clear difference in CH₄ production rates between major phytoplankton groups. There are probably intrinsic differences in metabolic pathways and growth responses between strains that shape these patterns of phytoplankton CH₄ production that require further exploration. Regardless of the underlying mechanisms, these experimentally derived rates can be extrapolated to the mesocosm field conditions to derive a first order estimate of the potential contribution of phytoplankton to ambient OMP. Our results suggest that the production of CH₄ by phytoplankton is likely to have a small contribution (maximum potential scenarios reached up to 15.9 ± 8.1%) of the estimated ambient OMP in all the mesocosms, despite the high algal biomass and chlorophyll concentration that characterized mostly SA and BU. Studies have reported a positive influence of light exposure and intensity on phytoplankton CH₄ production under controlled conditions^{21,24,28}. In this regard, the growth media and light conditions used in the CH₄ production essays do not mimic the ambient conditions that these phytoplankton strains experience in situ. Similarly, every isolation technique has its own biases and may generate potentially different growth conditions. However, there is no reason to think that the above-mentioned differences would lead to strains expressing CH₄ production rates that would be orders of magnitude higher in situ than in culture. All the evidence points to the fact that whereas the major phytoplankton groups in these shallow lakes do produce CH₄, these phytoplankton-linked CH₄ production rates account for only a small proportion of the observed OMP. This may explain the observed uncoupling between the estimated OMP and the ambient chlorophyll in these systems (Fig. S21).

CH₄ production as a by-product of MPn degradation in the process of phosphorus acquisition by bacteria is a widely known source of oxic CH₄ production in P-stressed waters^{2,10}. In the presence of phosphate, however, MPn degradation activity of bacteria can be repressed¹². Shallow Pampean lakes have high concentrations of phosphorus and, therefore, degradation of MPn is not expected to be a substantial CH₄ source, although this pathway cannot be discarded. Grossart et al.⁴ also reported that methanogenic archaea could attach to phytoplankton, possibly living in micro-anoxic niches, and this implies that they could potentially produce CH₄ through anaerobic methanogenesis but in the water column. Analysis of DNA from the water revealed the presence of 16S rRNA gene sequences of methanogens in all the mesocosms from the three shallow lakes. While this is no measure of methanogenic activity, we cannot exclude that archaea may have contributed to methane production in the mesocosms. In this regard, studies have further suggested a link between OMP and ambient primary production^{20,31}, assumed to reflect direct photosynthesis-related algal CH₄ production, but which may reflect the enhancement of other OMP pathways, including algal-associated archaeal methanogenesis. If such a connection exists, our results suggest that it is not scalable across systems, since our mesocosms had comparable OMP to those reported in oligotrophic and mesotrophic sites yet primary production rates were several folds higher than in those oligotrophic systems. In addition, methane production through photooxidation of organic matter^{73,74} has also been reported as a source of OMP, although the predominant product seems to be CO₂ rather than CH₄⁷⁹. These shallow lakes had high light irradiances during the experiments, implying that this pathway

could contribute to OMP but, if it was the case, this would probably occur in a slight proportion. Another potential source reported as explanation for OMP is bacterial degradation of dissolved organic matter (DOC)¹⁰. The three shallow lakes from this study exhibited high concentrations of DOC suggesting that this pathway could potentially also contribute to the observed OMP rates.

Our results imply that OMP is not the dominant pathway fueling overall CH₄ diffusive emissions measured in these lakes, despite being eutrophic and highly productive. Previous studies have suggested that lake morphometry plays a role in determining the contribution of OMP to total CH₄ production or emission, in particular, the ratio of sediment area (A_{sed}) to mixed layer volume (V)^{31,32}. The results from these shallow lakes are in good agreement with the patterns found in lakes elsewhere, and extend the reported patterns to a much wider range of values of A_{sed}/V (Fig. 4). This pattern suggests that CH₄ dynamics in these shallow lakes are dominated by other processes, such as sediment CH₄ production and/or lateral transport from the catchment, regardless of phytoplankton biomass and ecosystem metabolism. Deep lakes fall into the other extreme, where the water column is largely uncoupled from sediments, and where OMP plays a major role in determining CH₄ emissions, even when OMP rates may be low.

In summary, through field mesocosm experiments we were able to estimate ambient OMP rates and the potential contribution of this pathway to total CH₄ fluxes in three shallow lakes that differed in algal biomass and productivity. Furthermore, by means of controlled experiments we were also able to infer the potential contribution of phytoplankton to estimated OMP rates (Fig. 5). We have shown that OMP rates in these eutrophic lakes were comparable to those reported in oligotrophic and mesotrophic lakes despite large differences in phytoplankton biomass and primary production. The contribution of OMP to CH₄ diffusive emissions (OMC) was modest (< 15%), suggesting that in these shallow lakes, other sources dominate CH₄ emissions. Overall, the potential contribution of phytoplankton to the estimated OMP was low, despite the large algal biomass found in some of the lakes (Fig. 5). The main pathways of OMP therefore remain unclear, and the contribution of different pathways may vary among lake types, which may explain the diversity of OMP rates and potential drivers that have been reported in the literature. Our study extends the range of ecosystems where OMP has been detected, demonstrating that these shallow lakes fit previously hypothesized morphometric patterns of OMP contribution despite their high phytoplankton abundance, and establishes that phytoplankton does not appear to play a major direct role in shaping these ambient OMP rates.

Insights into OMP have shown that CH₄ sources in aquatic ecosystems are more variable and complex than previously recognized, thereby advancing our knowledge of CH₄ cycling. Although these findings do not alter current estimates of total CH₄ emissions, they refine our understanding of how these emissions are partitioned among different sources. Future research should focus on quantifying the contribution of OMP in other shallow

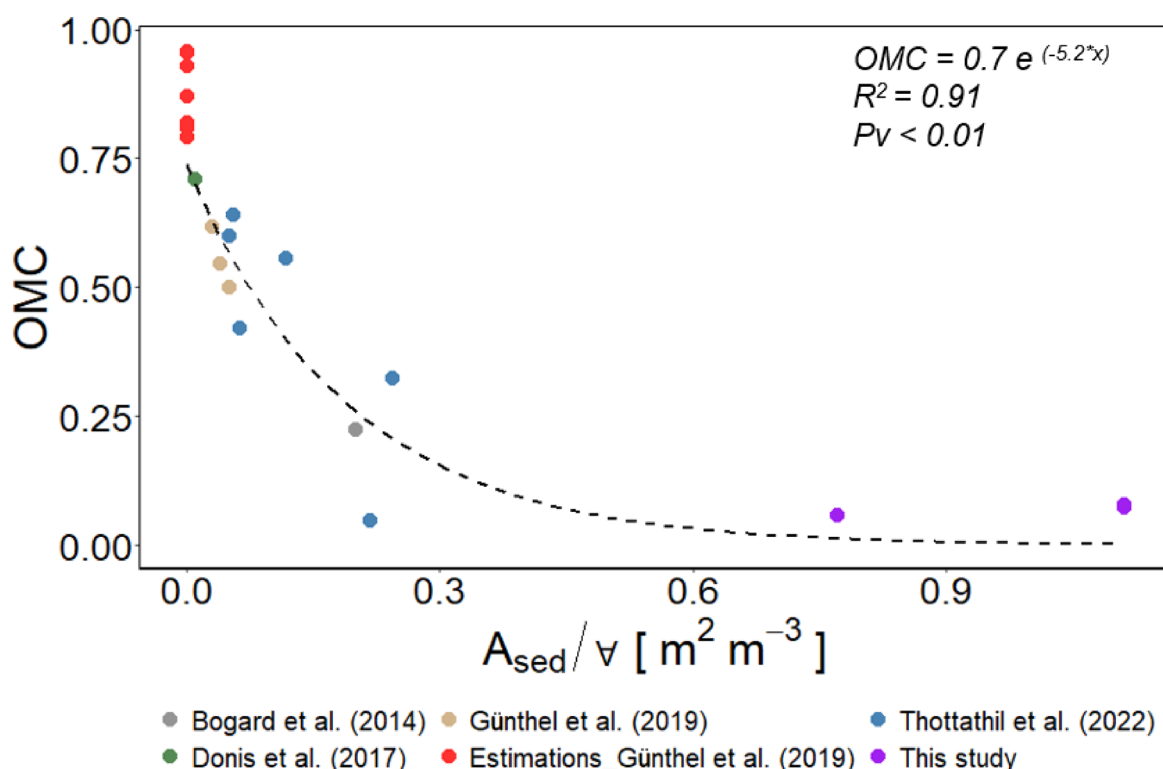


Fig. 4. Relationship between oxic methane contribution (OMC) and lake morphometry, specifically, the ratio of sediment area (A_{sed}) to mixed layer water column volume (V). For these shallow and polymictic lakes, the entire lake volume is considered as V . The colours represent different studies, and the data in purple dots correspond to this study. “Estimations_Günthel et al.²¹” refers to the estimations reported in that study for lakes other than those specifically studied, which are included as “Günthel et al.²¹”.

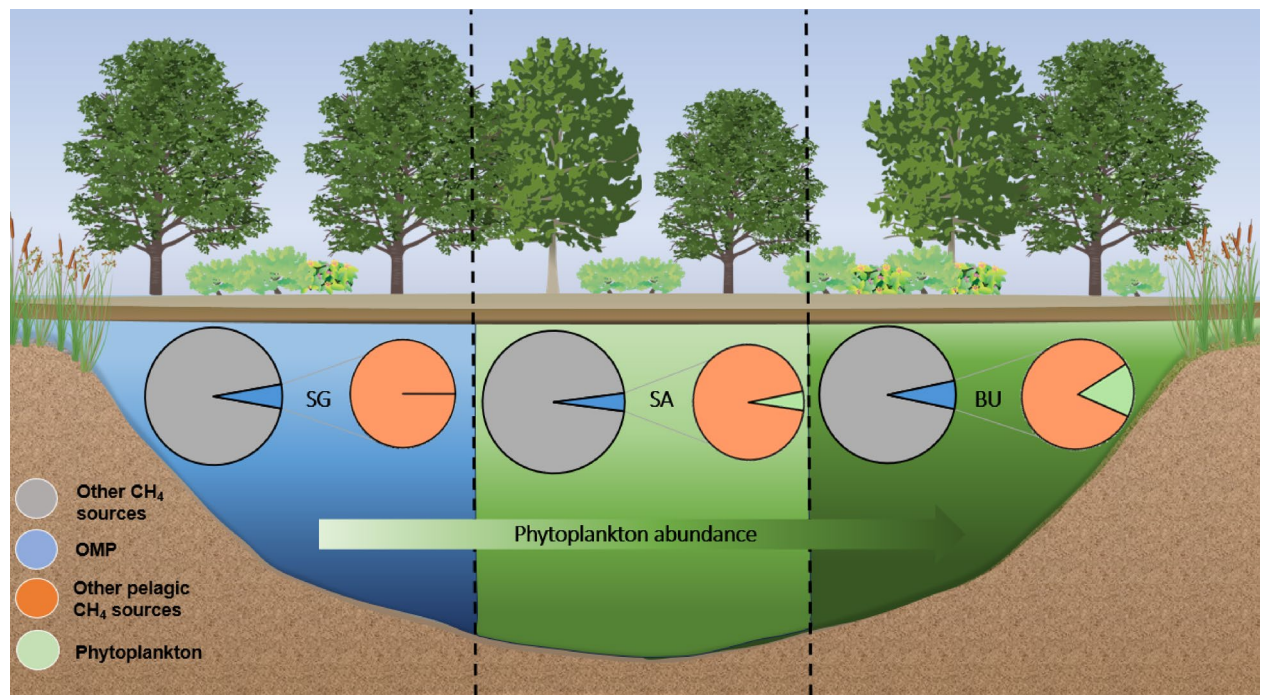


Fig. 5. Conceptual figure depicting the potential contribution of OMP (blue) to total lake CH_4 diffusive flux (grey) and the potential contribution of phytoplankton CH_4 production (green) to OMP ecosystem rates (orange), assuming the maximum potential scenario of contribution in all cases. The contribution of OMP (blue) to lake CH_4 diffusive fluxes was obtained as mentioned in Section 7, the contribution of phytoplankton to OMP rates was obtained as explained in Section 11. Tree and bush symbols from Dylan Taillie and Jane Hawkey, respectively, and emergent macrophyte symbols from Tracey Saxby, Integration and Application Network, University of Maryland Center for Environmental Science.

and diverse ecosystems, as well as elucidating the mechanisms underlying phytoplankton-mediated CH_4 production, and the metabolic and environmental factors regulating this process.

Data availability

The data that support the findings of this study are available from the corresponding author upon reasonable request.

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Author contributions

S.B. contributed substantially to the designing of the research, field work and data acquisition, phytoplankton strain isolation and phytoplankton experiments, molecular analysis, statistical analyses and writing of the manuscript. M.L.S. contributed substantially to the designing of the research, field work and data acquisition. M.B. D. I and H. P. G contributed substantially to the phytoplankton experiments carried out in Germany. S.T. contribute substantially to the isotopic mass balances. M. C. B contributed substantially to field work. A. J contributed substantially to the isolation of phytoplankton strains. P.A.G. contributed substantially to the designing of the research, analyses of the results and writing of the manuscript. All authors reviewed the manuscript.

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Declarations

Competing interests

The authors declare no competing interests.

Additional information

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